

Hydrogen transport in yttrium hydride under asymmetric heat

May 2024

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Background

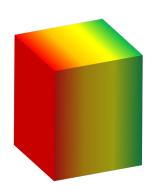
- Hydrogen diffusion in metals is driven by:
 - A hydrogen concentration gradient:

$$-D\nabla c$$

– A temperature gradient:

$$-D\frac{Q^*c}{RT^2}\nabla T$$

- Thermal diffusion (Soret effect) is often neglected because:
 - Assumed uniform temperature.
 - The Soret coefficient is unknown (experiments are difficult).
- The need to consider the Soret effect:
 - Any hydrogen situation where there is a temperature gradient.
 - Any pin or tube e.g. hydrate moderator
 - Fusion plasma facing surfaces.



Diffusion model

$$J_{dif} = -D\nabla c - D\frac{Q^*c}{RT^2}\nabla T$$

 $D[m^2s^{-1}] = \text{diffusivity}$ $c[m^{-3}] = \text{dissolved Q atom concentration in metal}$ $Q^*[J \ mol(Q_2)^{-1}, or \ J \ m^3] = \text{Soret coefficient}$ $R = 8.3145 \times 10^{-3} \ kJ \ mol^{-1}K^{-1}$ T[K] = temperature

Experiments are difficult

- Imposing a temperature gradient causes the hydrogen to redistribute.
- Releasing the gradient will cause the hydrogen to redistribute, again.
- Possible solutions.
 - Assess the hydrogen distribution while a temperature gradient is imposed.



Quench the sample.

HEATD Hydrogen Experimental Apparatus for

Thermal Diffusion

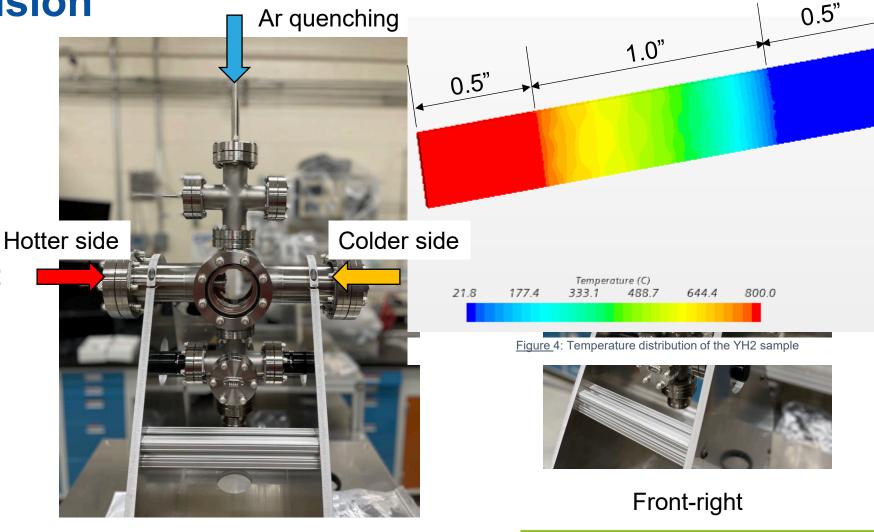
Hotter side temperature:

 Heater is able to reach closer to 1000°C.

- Planned: ~900°C

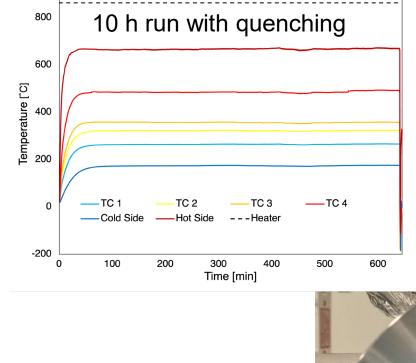
Colder side temperature:

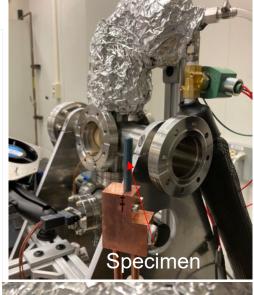
- Heater is able to reach closer to 1000°C.
- Can range from RT to close to hot side temperature

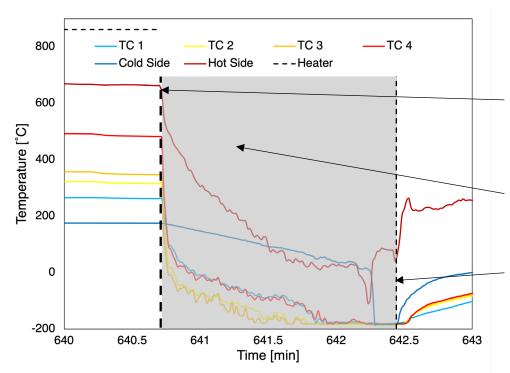


Current status and Experiments of HEATD

- Successful set up of system
- Successful heating for 1h, 10h, 100h and subsequentially liquid Ar quenching







Liquid Ar quenching starts, Heater off

Quenching and specimen disassembly

Specimen is out quenching is stopped

