

A Comparative Evaluation of Additive Manufactured Stainless Steel Test Pieces

Kevin Thomas Scholtes, Isabella J Van
Rooyen

August 2019



The INL is a U.S. Department of Energy National Laboratory
operated by Battelle Energy Alliance

A Comparative Evaluation of Additive Manufactured Stainless Steel Test Pieces

Kevin Thomas Scholtes, Isabella J Van Rooyen

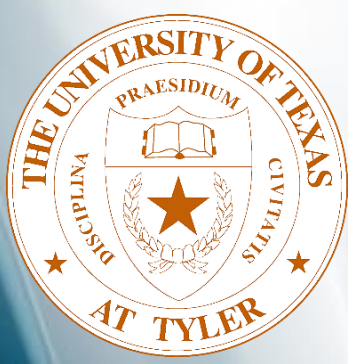
August 2019

**Idaho National Laboratory
Idaho Falls, Idaho 83415**

<http://www.inl.gov>

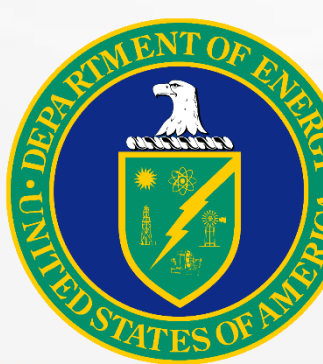
**Prepared for the
U.S. Department of Energy
Office of Energy Efficiency and Renewable Energy
Under DOE Idaho Operations Office
Contract DE-AC07-05ID14517**

A Comparative Evaluation of Additive Manufactured Stainless Steel Test Pieces



ORAU

Kevin Scholtes^{1,2}, Isabella van Rooyen¹
¹Idaho National Laboratory, Idaho Falls, ID, ²The University of Texas at Tyler, Tyler, TX



INL Idaho National Laboratory

Background

- Stainless steel is used for variety of applications with different grades, alloys, filaments and powders.
- Additive manufacturing (AM) enables unique design features and refined microstructures.
- Performance testing on AM parts are needed as input to modeling and accelerated qualification processes.
- The objective of this work is to determine the effect of surface roughness on corrosion rate for from different manufacturers.

Methods

Surface Roughness Testing:

- Initial surface roughness measurements were performed on each of the 72 samples.
- The surface roughness measurements were performed employing a Mitutoyo SJ-210 portable surface roughness tester.
- The calibration was executed employing a precision reference specimen (Ra: 2.94 µm).

Corrosion Rate:

Reference: ASTM G-1,G-31,G-42:

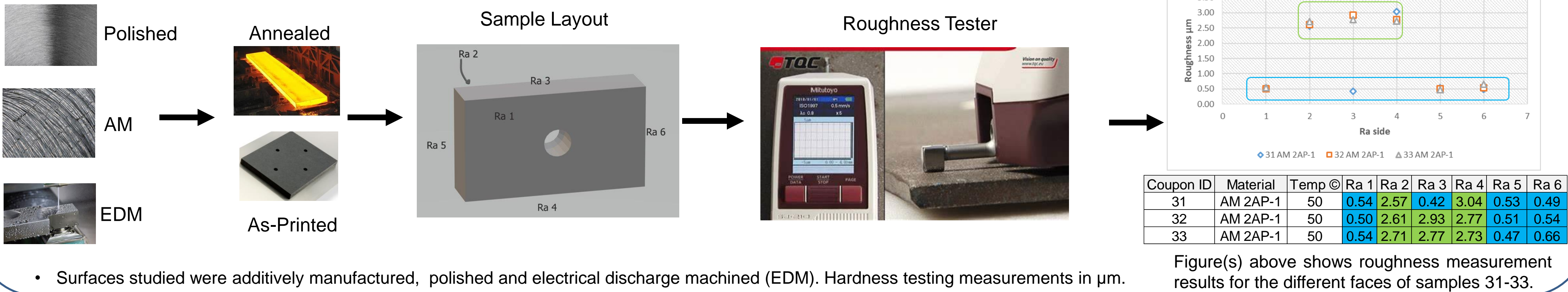
$$\text{Corrosion rate} = (K \times W) / (A \times T \times D)$$

K = constant (mils per year- mpy)
 T = time of exposure (0.01 hrs)
 A = area in cm² to nearest 0.01 cm²
 W = mass loss in g, to nearest 1mg
 D = density in g/cm³

Sample Test Groups:

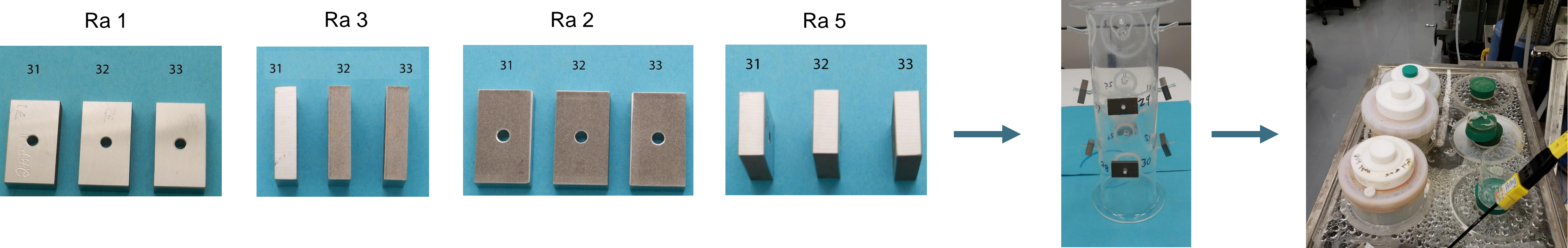
- 500, 1000 and 1500 hours.
- Each group was separated into smaller sections to investigate different surface structure and the effect of annealing.
- Sample groups are also reviewed to investigate vendor preference.

Initial Setup



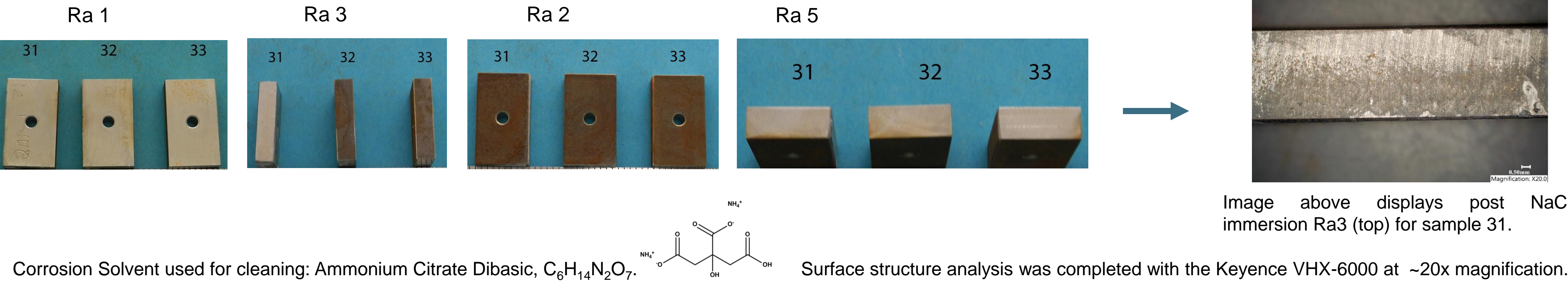
Surfaces studied were additively manufactured, polished and electrical discharge machined (EDM). Hardness testing measurements in µm.

Pre 5% NaCl Immersion



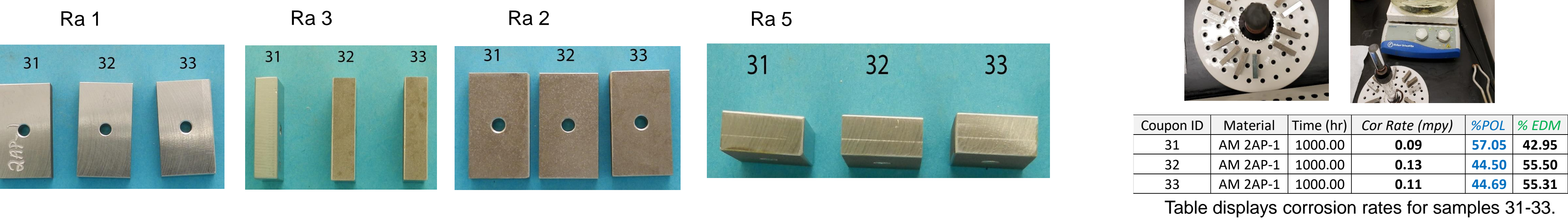
Samples were placed into static solution for predetermined time. Samples 31-33 above taken from the 1000-hour data set.

Post 5% NaCl Immersion



Corrosion Solvent used for cleaning: Ammonium Citrate Dibasic, C₆H₁₄N₂O₇.

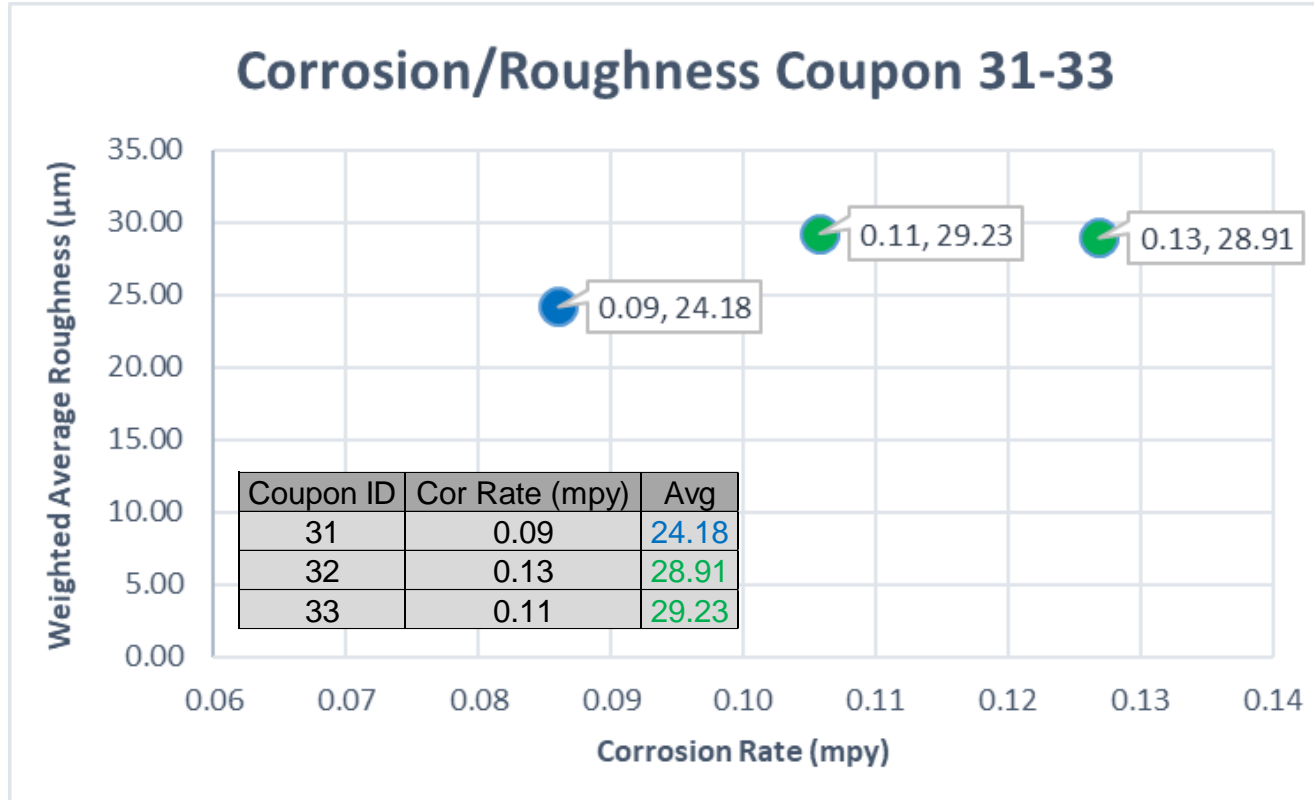
Post Corrosion Cleaning



The corrosion weight loss stabilized four cleaning cycles, each cycle was 8 minutes with predetermined temperature and flow requirements based on ASTM-G-31.

Discussion and Preliminary conclusions

- Although data analysis are in progress, as expected, the corrosion rate are lower on the polished surfaces compared with the surface finished as manufactured.



In the graph above the weighted average roughness was calculated by the percent surface area of the side (Ra #) divided by the total surface area then multiplied by the side surface roughness.

- Preliminary evaluation shows that the 500-hour data set was scattered, and trend analysis are complex, while the 1000h and 1500h test results evaluation show more distinct trends.

Future Work

Further analysis of the collected data aim to provide:

- A corrosion rate comparison between two manufacturers.
- Effect of increased corrosion holding time on the corrosion rates.
- The effect of annealing has on AM surfaces.

Acknowledgements

A special thanks to Tim Yoder for facilitating testing and analysis. Financial support from Oak Ridge Associated Universities and ORISE for arranging the internship research position. Facilities and management from Battelle Energy Alliance. Leadership, training, project development and support staff from Idaho National Laboratory.