Modeling Electrolysisbased H2 Production for Grid Service Assessment

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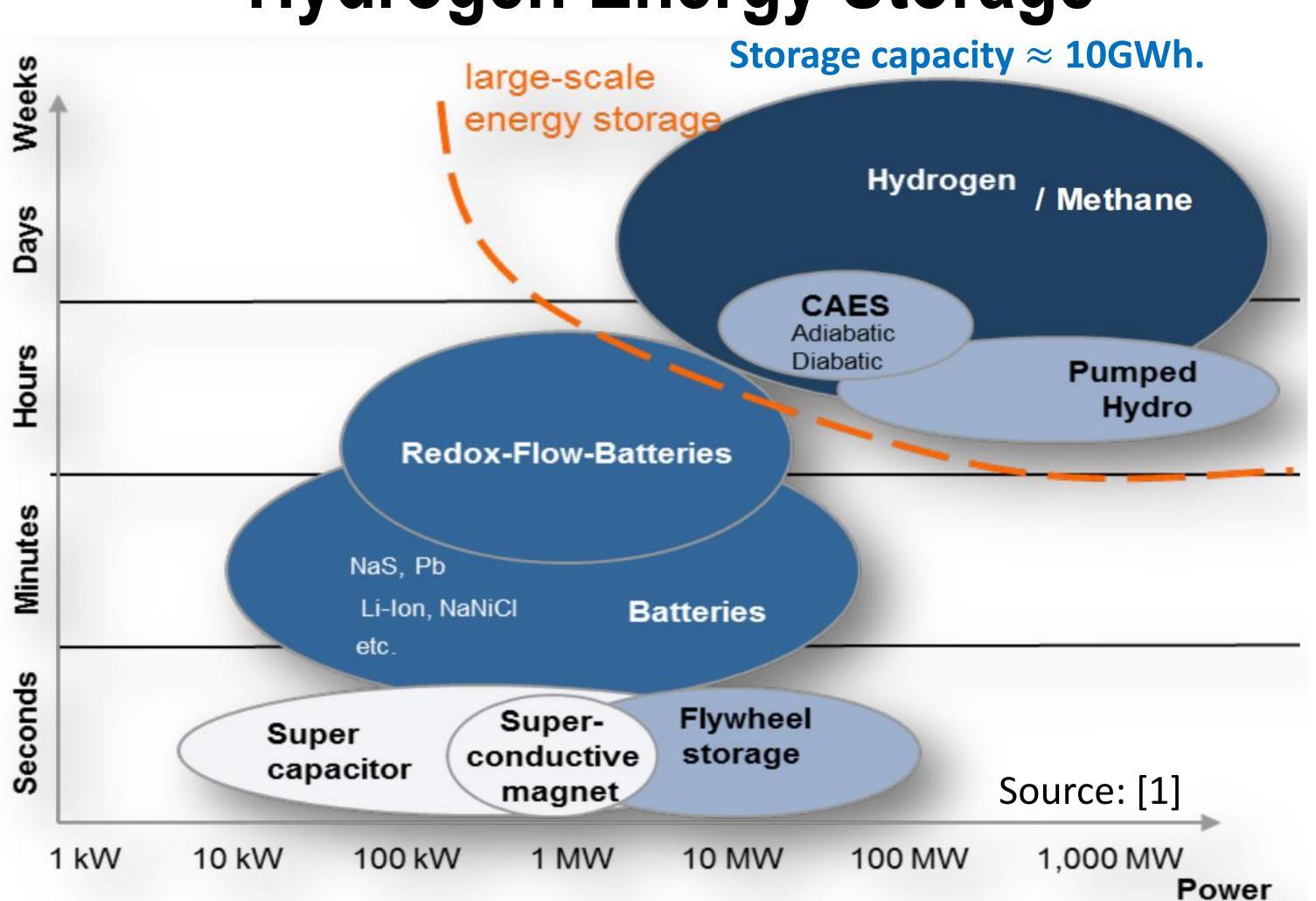
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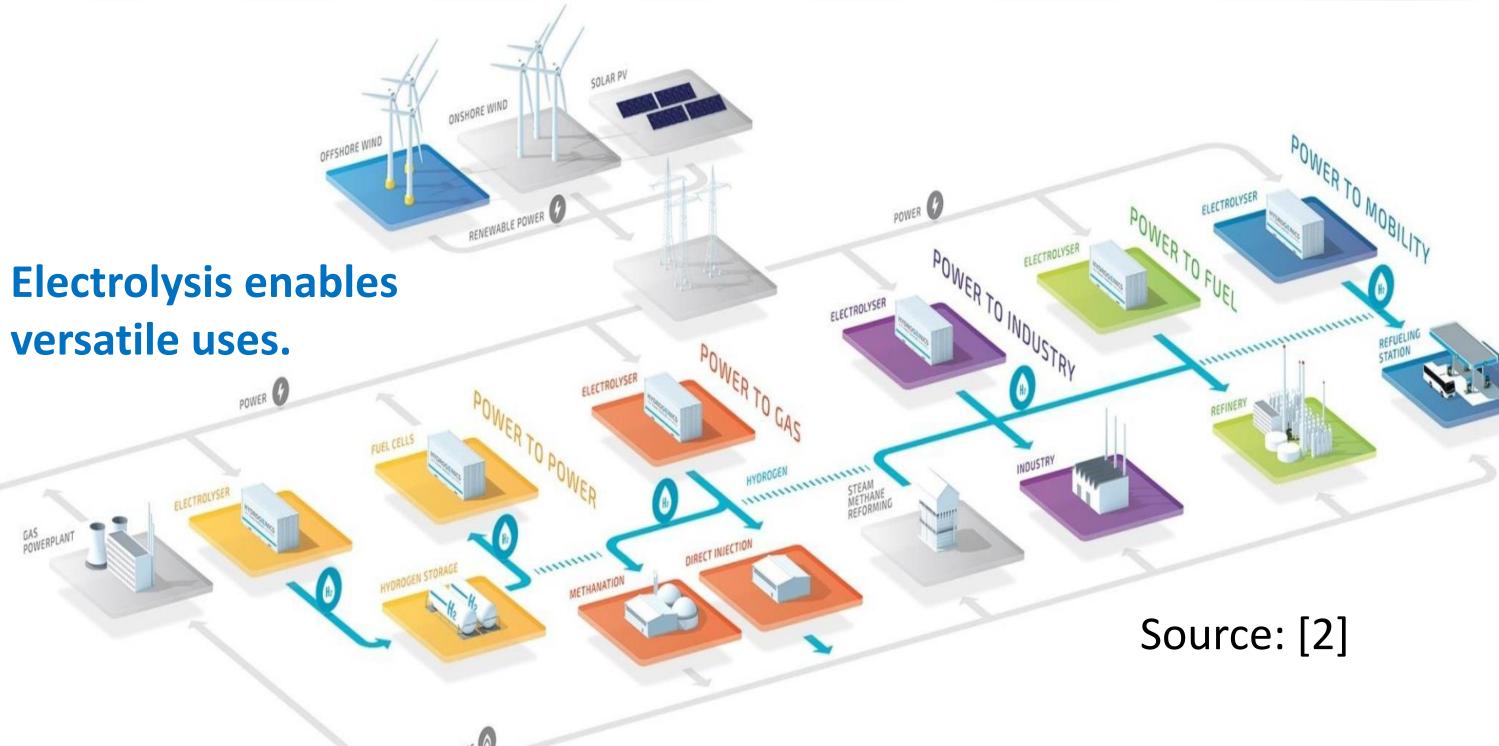
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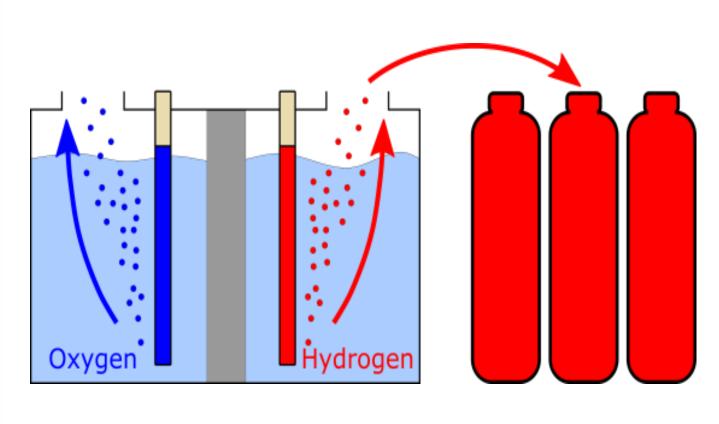
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Hydrogen Energy Storage





Electrolyzer Model



i: current density. I: current *i*₀: *exchange current density* n: number of electrons transferred α_a, α_c : transfer coeff. anode, cathode F: Faraday's constant $\eta = (V - V_{eq})$: overpotential. V: voltage *R*: *gas constant*.*T*: *temperature* N_c : number of electrolyzer cells in series η_F : Faraday efficiency V_tn: thermoneutral voltage P_{elec} : electric power. P_{comp} : compression power E_{H_2} : Low heating value of hydrogen \dot{m}_{H_2} : mass flow of produced hydrogen

> Reaction for hydrogen production in the electrolyzers Oxidation and reduction reactions at the anode and cathode: $2H_2O(l) \rightarrow O_2(g) + 4H^+ + 4e^-, 2H^+ + 2e^- \rightarrow H_2(g)$

 $\succ i - V$ curve. Butler-Volmer Equation:

$$i = i_0 \left\{ exp\left(\frac{\alpha_a nF\eta}{RT}\right) - exp\left(\frac{\alpha_c nF\eta}{RT}\right) \right\}$$

> Hydrogen production rate, voltage and charging efficiencies:

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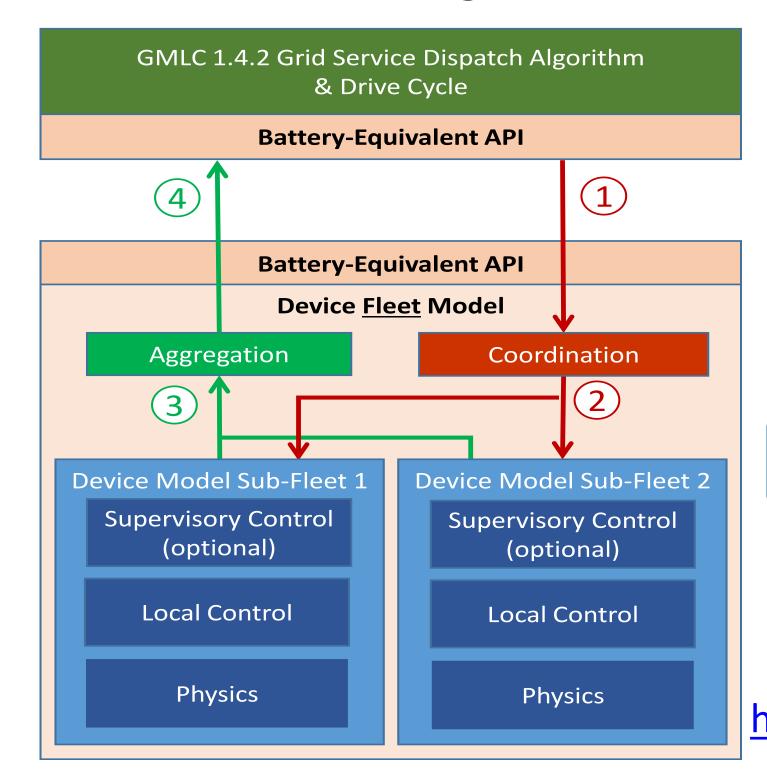
$$H_{2g} = \frac{N_c I}{2F} \eta_F$$
 $\eta_V = \frac{V_{tn}}{V}$ $\eta_{ch} = \frac{E_{H_2} \dot{m}_{H_2}}{P_{elec} + P_{comp}}$
Acknowledgement

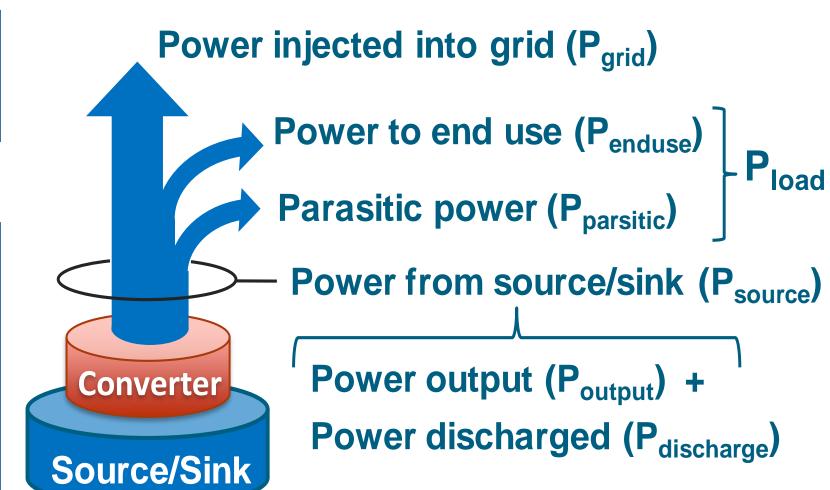
[1] GMLC 1.4.01: "Standards and Test Procedures for Interconnection and Interoperability"

(Link: https://gmlc.doe.gov/projects/1.4.01) [2] GMLC 1.4.02: "Definitions Standards and Test Procedures for Grid Services" (Link: https://gridmod.labworks.org/projects/1.4.02)



Battery Equivalent Model (BEM)





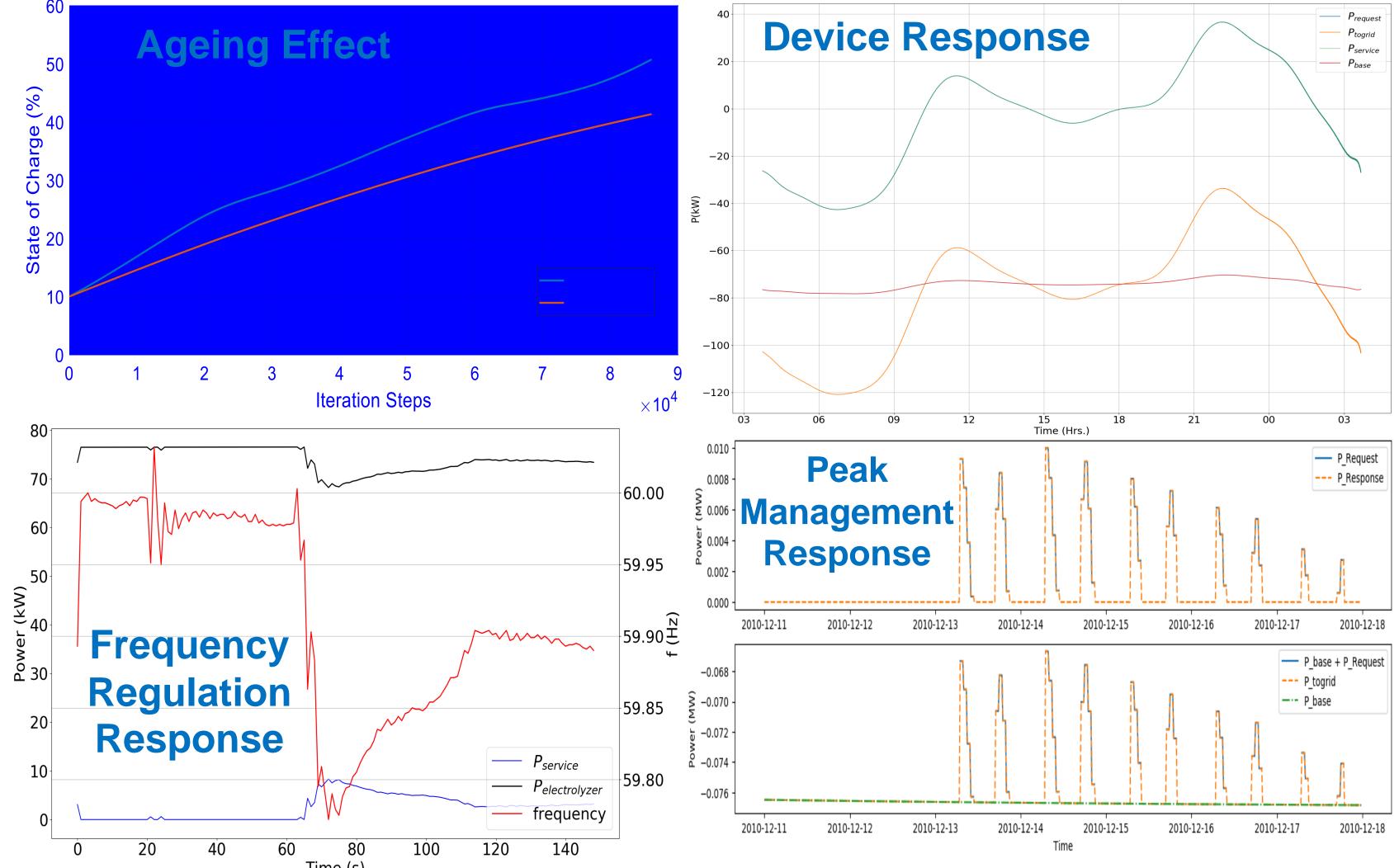
GitHub Repository

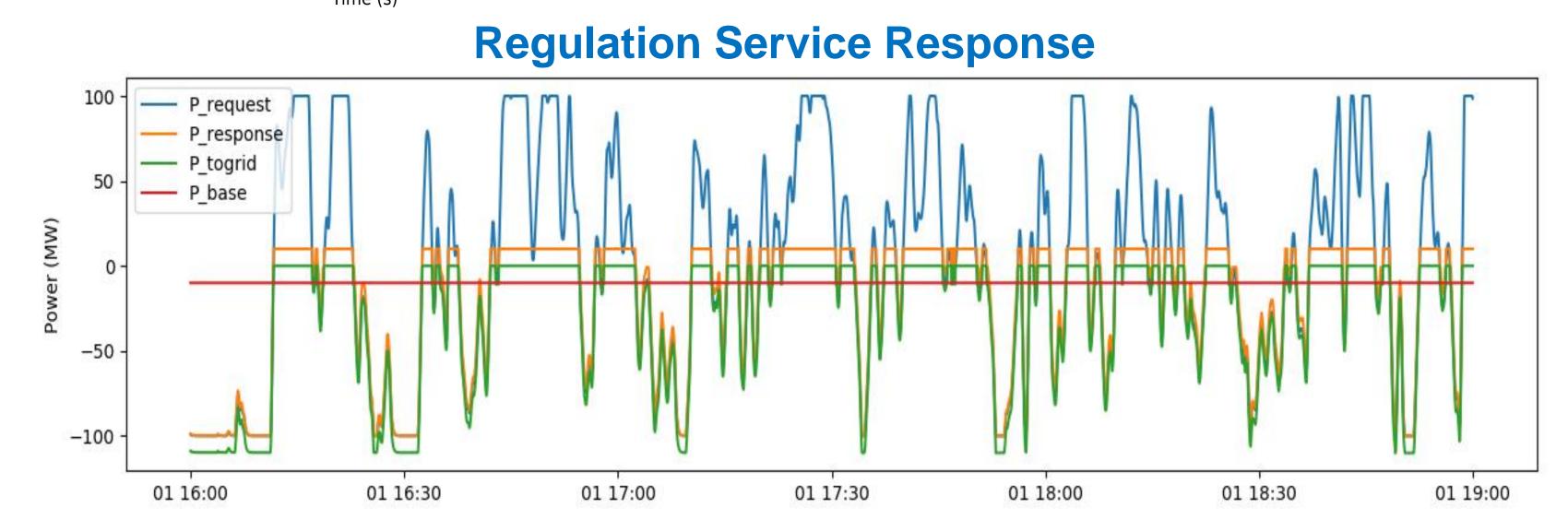
https://github.com/GMLC-1-4-2/battery interface

| Parameter | Electrolyzer Model | Battery-Equivalent Model |
|------------------|--|--|
| Power | Operating voltage (V) Operating current (I) | Input power P_{in} |
| Storage capacity | Tank(s) capacity (p_{max}) | Maximum state of charge SoC_{max} |
| Energy stored | Pressure in the storage tank(s) | Current state of charge SoC_{actual} |
| Operation time | continuous | Only if grid service is requested |
| Efficiency | Electrolyzer efficiency (η_e) | $\eta = \frac{Stored\ energy\ (H_2)}{Power\ supply\ *\ t}$ |

Results and Analysis

Electrolyzer BEM response to grid service requests





References

[1] Manfred Waidhas (Siemens), "Electrolyzer technology – the Siemens view", 2016. [2] Alan Kneisz (Hydrogenics), "Hydrogen – Most Versatile Energy Storage; Australia Energy Storage - Adelaide".

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