



MRWFD Center for Radiation Chemistry Research Activities

October 2021

Changing the World's Energy Future

Gregory P Horne



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Dr. Gregory P. Horne

Center for Radiation
Chemistry Research

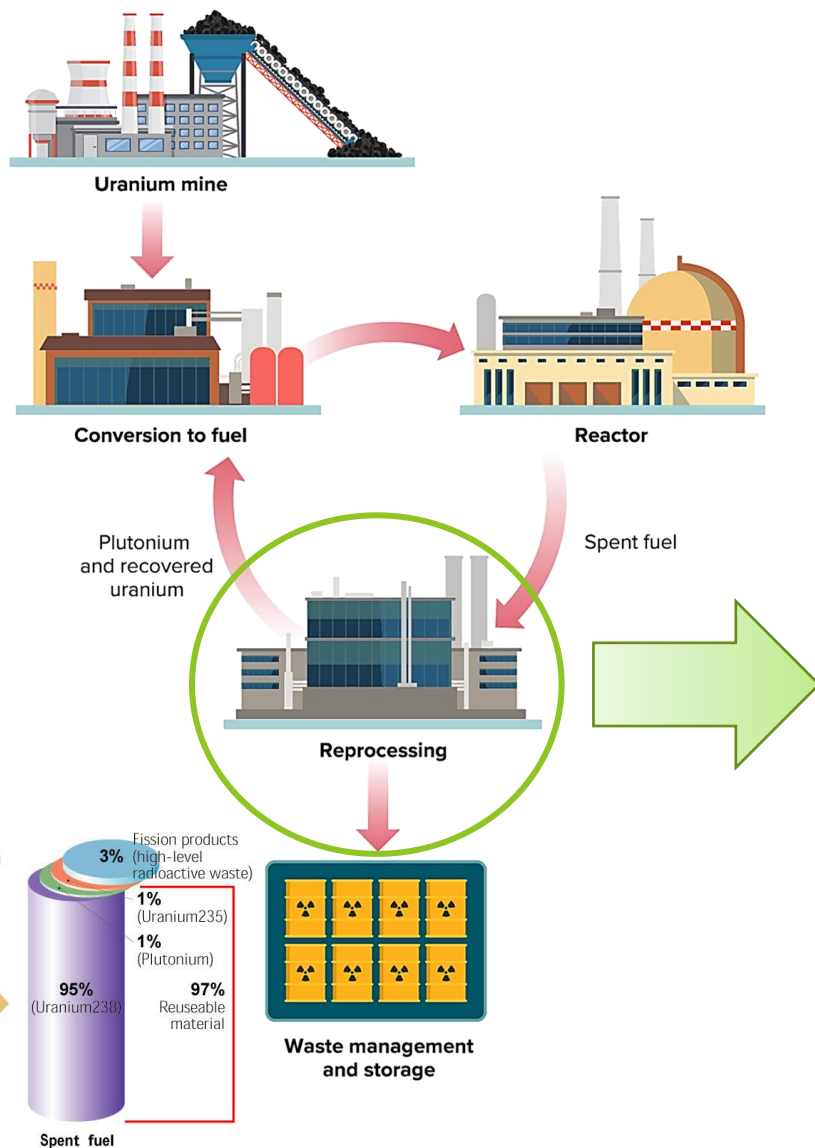
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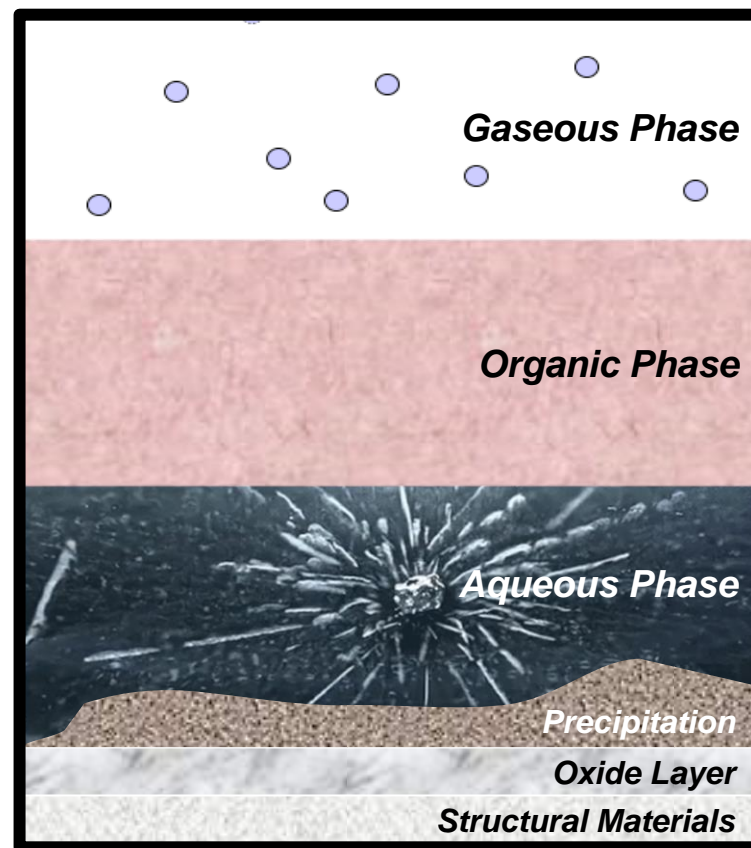
**DOE-NE Aqueous Separation
2021 Pls' Meeting**



Separations R&D and Radiation Chemistry



Solvent Extraction Reprocessing
Ligands/organic diluent: $\text{HNO}_3/\text{H}_2\text{O}$
(\pm additives)

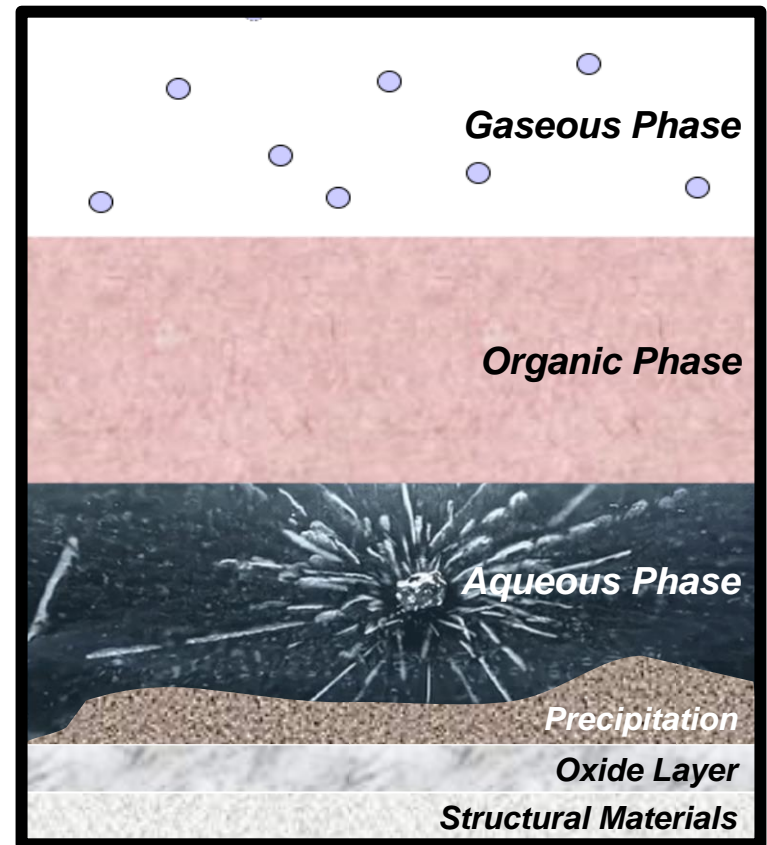


Separations R&D and Radiation Chemistry

Aim

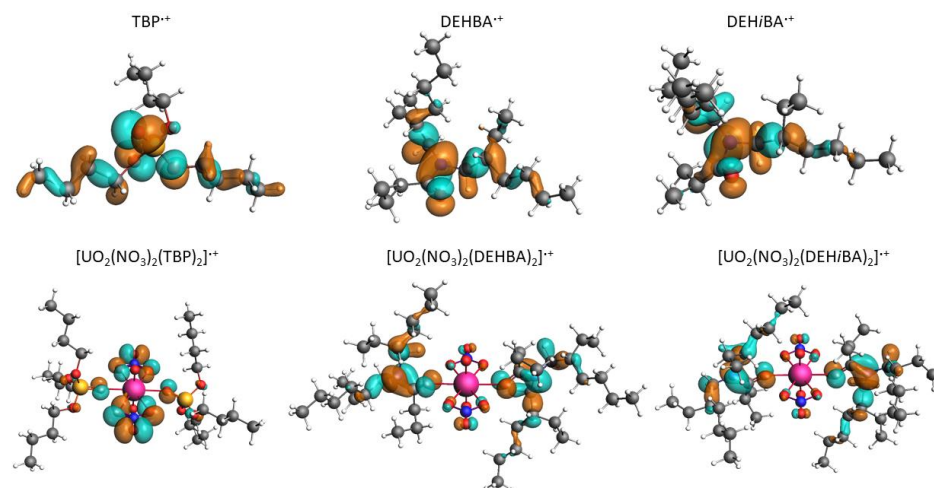
Provide quantitative, fundamental experimental data and insight into the effects of multi-component radiation fields on aqueous separation technologies to develop and evaluate complementary predictive multi-scale modelling capabilities.

Solvent Extraction Reprocessing
Ligands/**organic diluent**: $\text{HNO}_3/\text{H}_2\text{O}$
(\pm **additives**)



Effect of Metal Ion Complexation on the Radiolysis of TBP and Monoamide Extractants

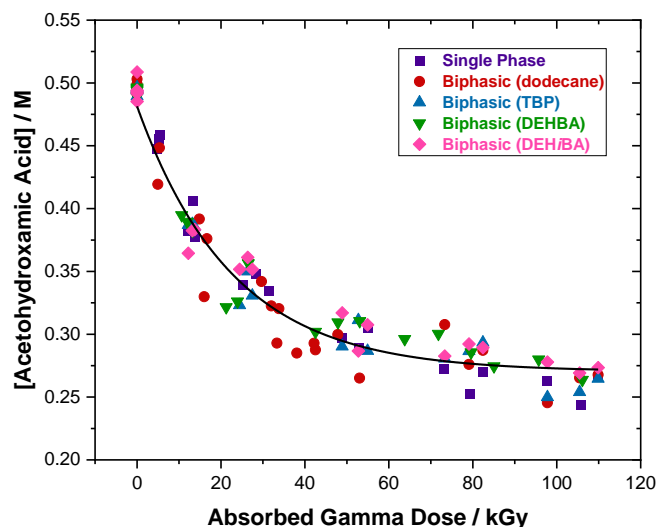
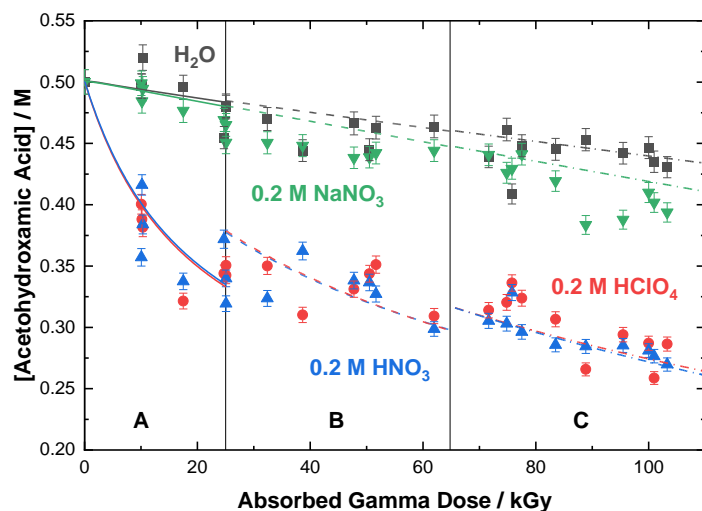
Sample	RH ^{•+} Rate Coefficient (10 ¹⁰ M ⁻¹ s ⁻¹)
TBP	1.36 ± 0.07
[UO ₂ (NO ₃) ₂ (TBP) ₂]	-
DEHBA	0.93 ± 0.02
[UO ₂ (NO ₃) ₂ (DEHBA) ₂]	2.49 ± 0.06
DEH/BA	1.14 ± 0.04
[UO ₂ (NO ₃) ₂ (DEH/BA) ₂]	1.59 ± 0.08



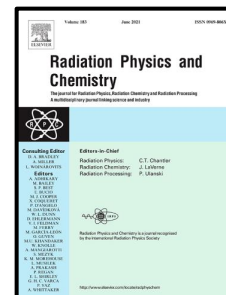
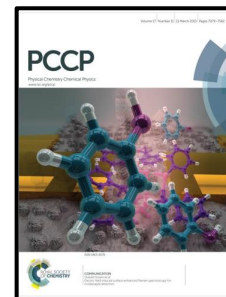
Canonical Kohn-Sham molecular orbitals of the electron holes in the geometry-optimized radical cation species for TBP, DEHBA, and DEH/BA.

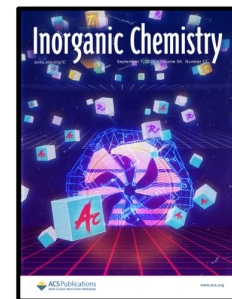
C.C. Barros, C.D. Pilgrim, A.R. Cook, S.P. Mezyk, T.S. Grimes, and G.P. Horne, Influence of Uranyl Complexation on the Reaction Kinetics of the Dodecane Radical Cation with Used Nuclear Fuel Extraction Ligands (TBP, DEHBA, and DEH/BA), *PCCP*, 2021, *in review*.

Radiolytic Evaluation of AHA and CDTA Additives by Multi-Scale Modeling

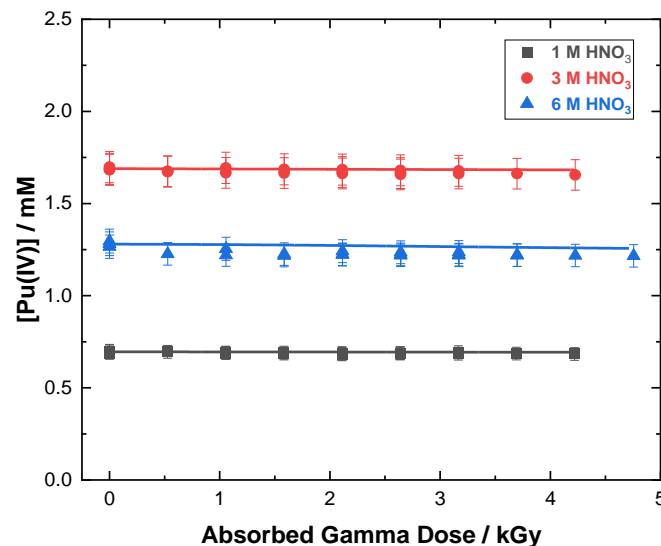
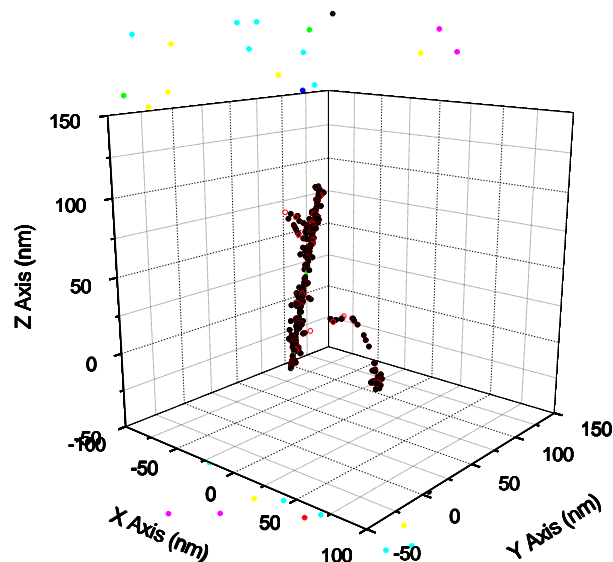


Radiolytic Species	CDTA Rate Coefficient (k , $M^{-1} s^{-1}$)
e_{aq}^-	$(5.31 \pm 0.14) \times 10^7$
H^\bullet	$(2.75 \pm 0.15) \times 10^8$
$\cdot OH$	$(6.41 \pm 0.11) \times 10^9$
NO_3^\bullet	$(4.06 \pm 0.10) \times 10^8$

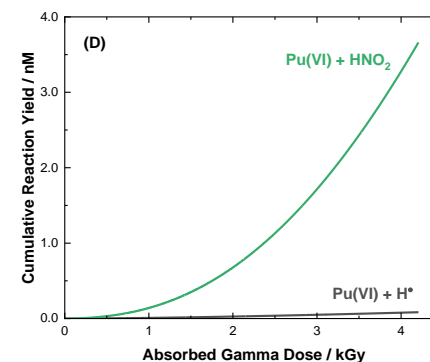
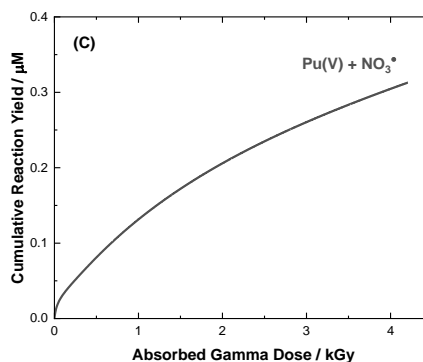
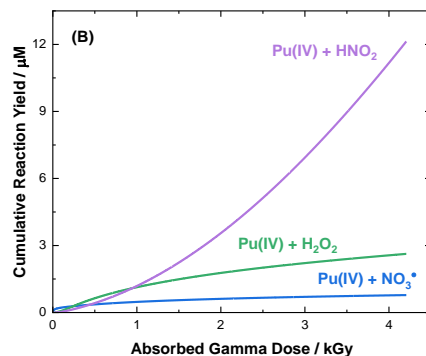
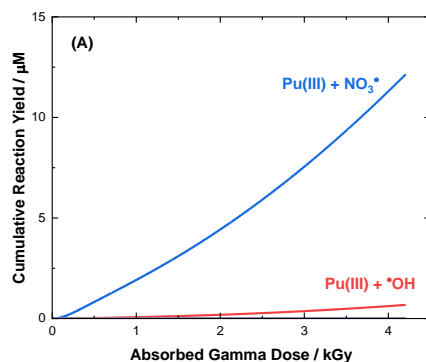




Elucidating the Radiation-Induced Redox Chemistry of Plutonium by Multi-Scale Modeling



Supplying Electrons



- Clifford, P., Green, N.J.B., Oldfield, M.J., Pilling, M.J., Pimblott, S.M., *J. Chem. Soc., Faraday Trans.*, **1986**, 82, 2673.
- Pimblott, S.M., LaVerne, J.A., Mozumder, A., *J. Phys. Chem.*, **1996**, 100, 8595.
- Horne, G.P., Donocliot, T.A., Sims, H.E., Orr, R.M., Pimblott, S.M., *J. Phys. Chem. B.*, **2016**, 120 (45), 11781.

Vision for the Future (~10 Years)

Aim

- Provide quantitative, fundamental experimental data and insight into the effects of multi-component radiation fields on aqueous separation technologies to develop and evaluate complementary predictive multi-scale modelling capabilities.

Research Needs

- Integrate multi-scale modelling capabilities with process scale codes and phenomena.
- Develop multi-scale modelling capabilities for (i) organic and (ii) biphasic solution radiation chemistry.
- Evaluate the effect of real system formulations on previously established fundamental chemistry in 'pristine' systems.
- Accelerate molecular design through the development of advanced screening approaches

Acknowledgements



U.S. DEPARTMENT OF
ENERGY



FLORIDA STATE
UNIVERSITY



The University of Manchester



IDAHO NATIONAL LABORATORY

Summary of FY21 Deliverables

1. **M3FT-21IN030101026:** “*Effect of Metal Ion Complexation on the Radiolysis of TBP and Monoamide Extractants*” submitted to **PCCP**, in peer-review.
2. **M3FT-21IN030101025:** “*Radiolytic Evaluation of CDTA and AHA Additives*” manuscripts in preparation for submission to **JACS**, **Chem. Sci.**, **PCCP**, and **Rad. Phys Chem.**
3. **M3FT-21IN030101027:** “*Modeling Radiation-Induced Plutonium Redox Chemistry*” model has been developed and milestone manuscript in preparation with CEA collaborators.

