



# Molten Salt Reactors: Materials Challenges

September 2022

*Changing the World's Energy Future*

Guy L Fredrickson



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# **Molten Salt Reactors: Materials Challenges**

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# Molten Salt Reactors

## Materials Challenges

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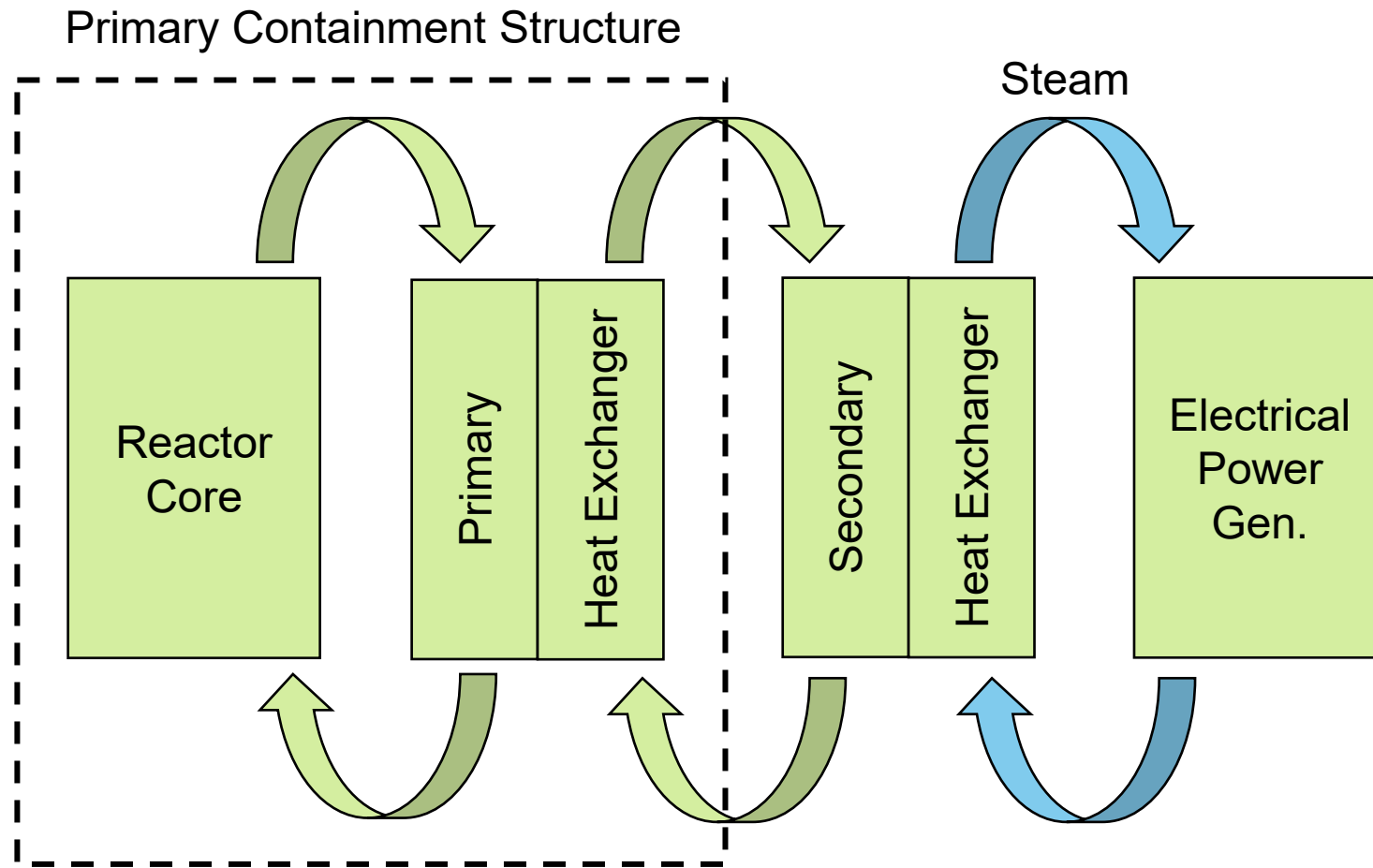


Idaho National Laboratory

# Molten Salt Reactor History

- Aircraft Reactor Experiment, November 2-12, 1954
- Molten Salt Reactor Experiment, June 1, 1965, to December 12, 1969
- Terrestrial Energy
- TerraPower
- Southern Company
- Seaborg
- Flibe Energy
- Moltex Energy
- ThorCon

# Nuclear Reactors



- Primary Coolants
  - Water
  - Heavy Water
  - Gas
    - Air
    - He
    - CO<sub>2</sub>
  - Liquid Metal
    - Pb & PbBi
    - Na & NaK
  - Molten Salt

## Comparison: Salt, Sodium, Water

	ORNL Single-Fluid MSBR	Orano Super-Phenix	Westinghouse AP1000
Reactor Type	MSBR	SFBR	PWR
Status	Conceptual	Decommissioned	Operating
Thermal Power, MWt	2,250	3,000	3,415
Electrical Power, MWe	1,000	1,200	1,117
Power Efficiency, %	44.4	40.0	32.7
Liquid Coolant	Molten Fluoride Salt	Sodium	Pressurized Water
Liquid Coolant Inventory, MT	160	3,500	178
Fuel Type	Molten Fluoride Salt	Mixed Oxide	Uranium Oxide
Breeder Fuel Cycle	$^{232}\text{Th}/^{233}\text{U}$	$^{238}\text{U}/^{239}\text{Pu}$	-
Neutron Spectrum	Thermal	Fast	Thermal

### References:

- Conceptual Design Study of a Single-Fluid Molten-Salt Breeder Reactor, ORNL-4541, Jun 1971
- Liquid Metal Cooled Reactors: Experience in Design and Operation, IAEA-TECDOC-1569, Dec 2007
- The Westinghouse AP1000 Advanced Nuclear Plant: Plant Description, Westinghouse, 2003

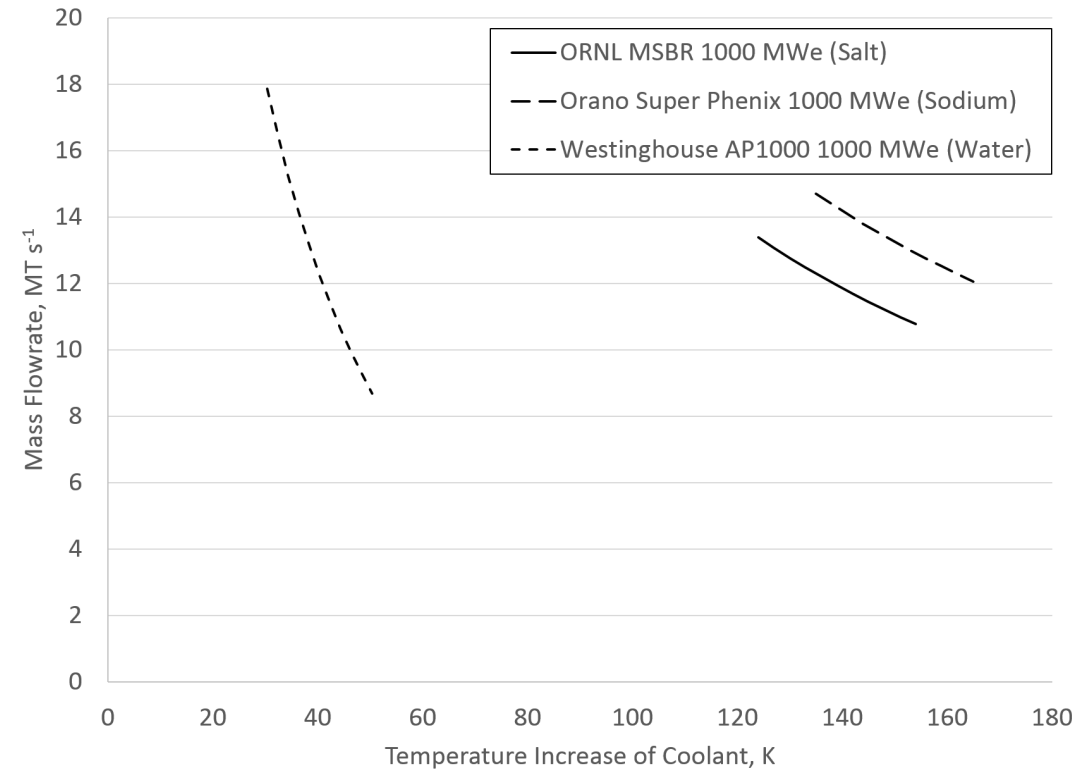
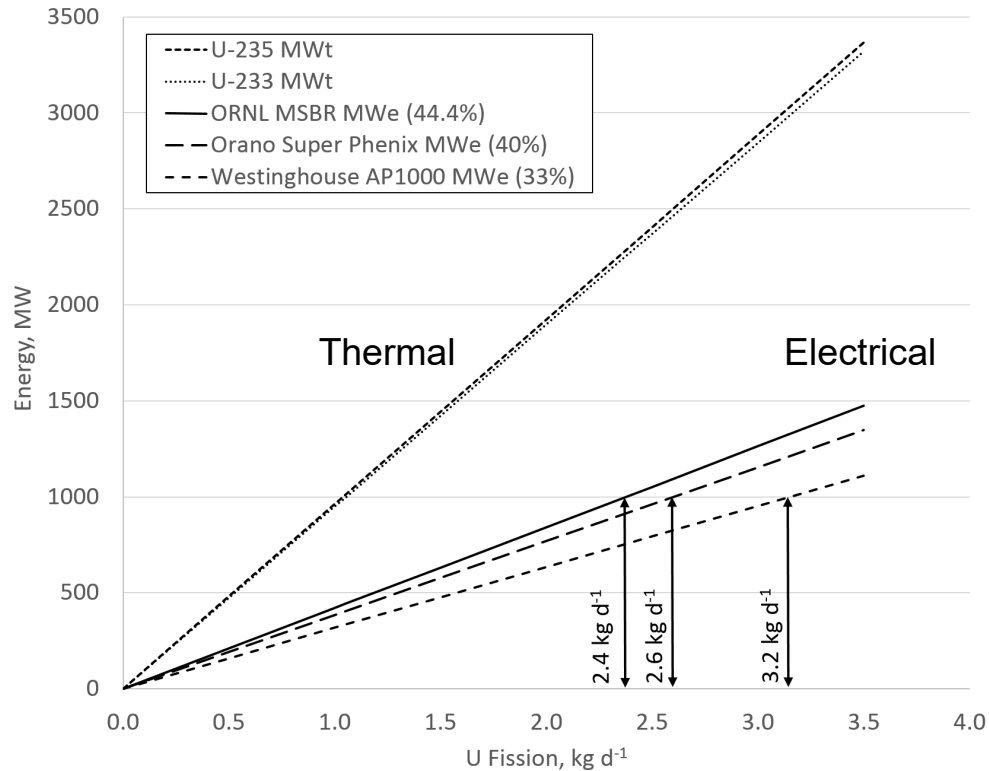
## Comparison: Salt, Sodium, Water

	ORNL Single-Fluid MSBR	Orano Super-Phenix	Westinghouse AP1000
Liquid Coolant	Molten Fluoride Salt	Sodium	Pressurized Water
Liquid Coolant Inventory, MT	160	3,500	178
Inlet Temperature, °C	566	395	281
Outlet Temperature, °C	705	545	321
Increase in Temperature, °C	139	150	40
Heat Capacity, J g <sup>-1</sup> K <sup>-1</sup>	1.357	1.258	6.137
Density, kg L <sup>-1</sup>	3.281	0.811	0.917

Takeaways: The MSBR operates at the highest temperature.  
Water has the highest heat capacity.  
Salt has the highest density.

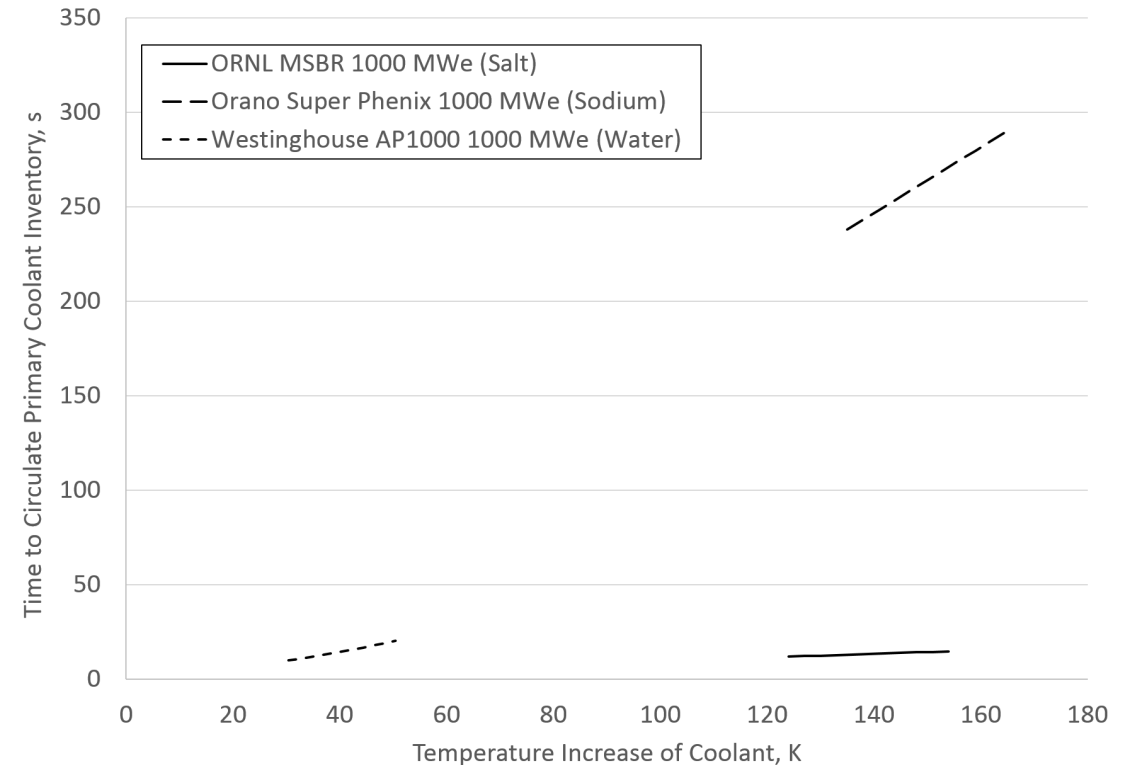
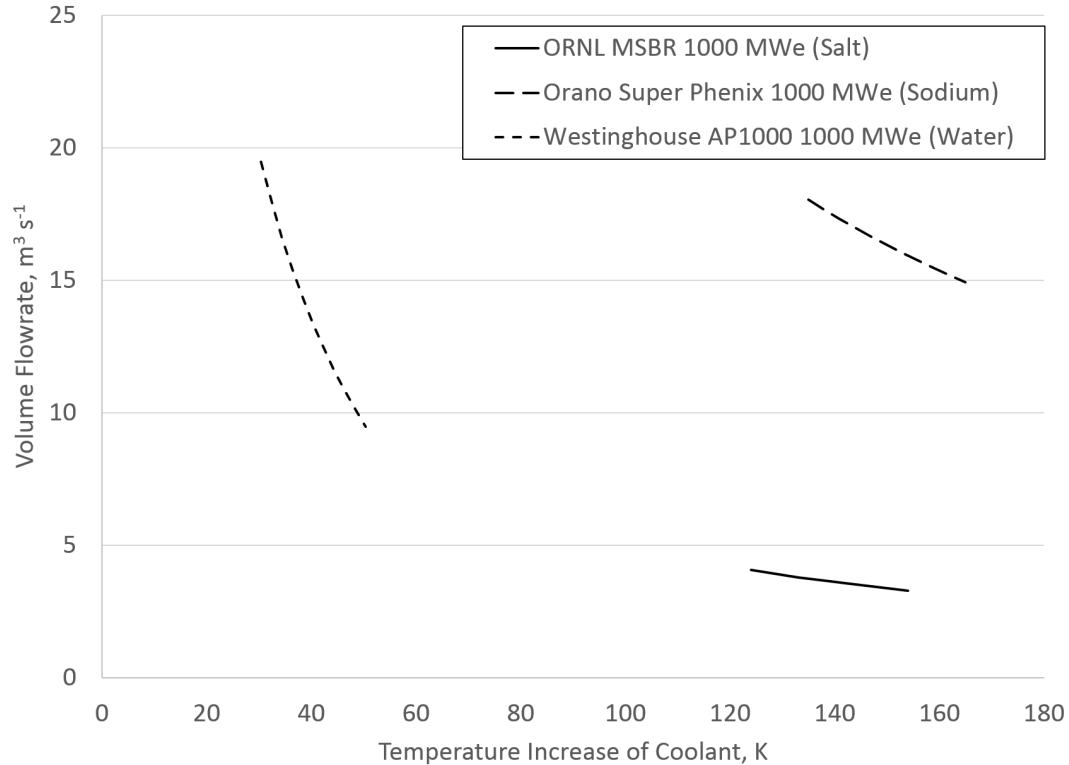


# Comparison: Salt, Sodium, Water



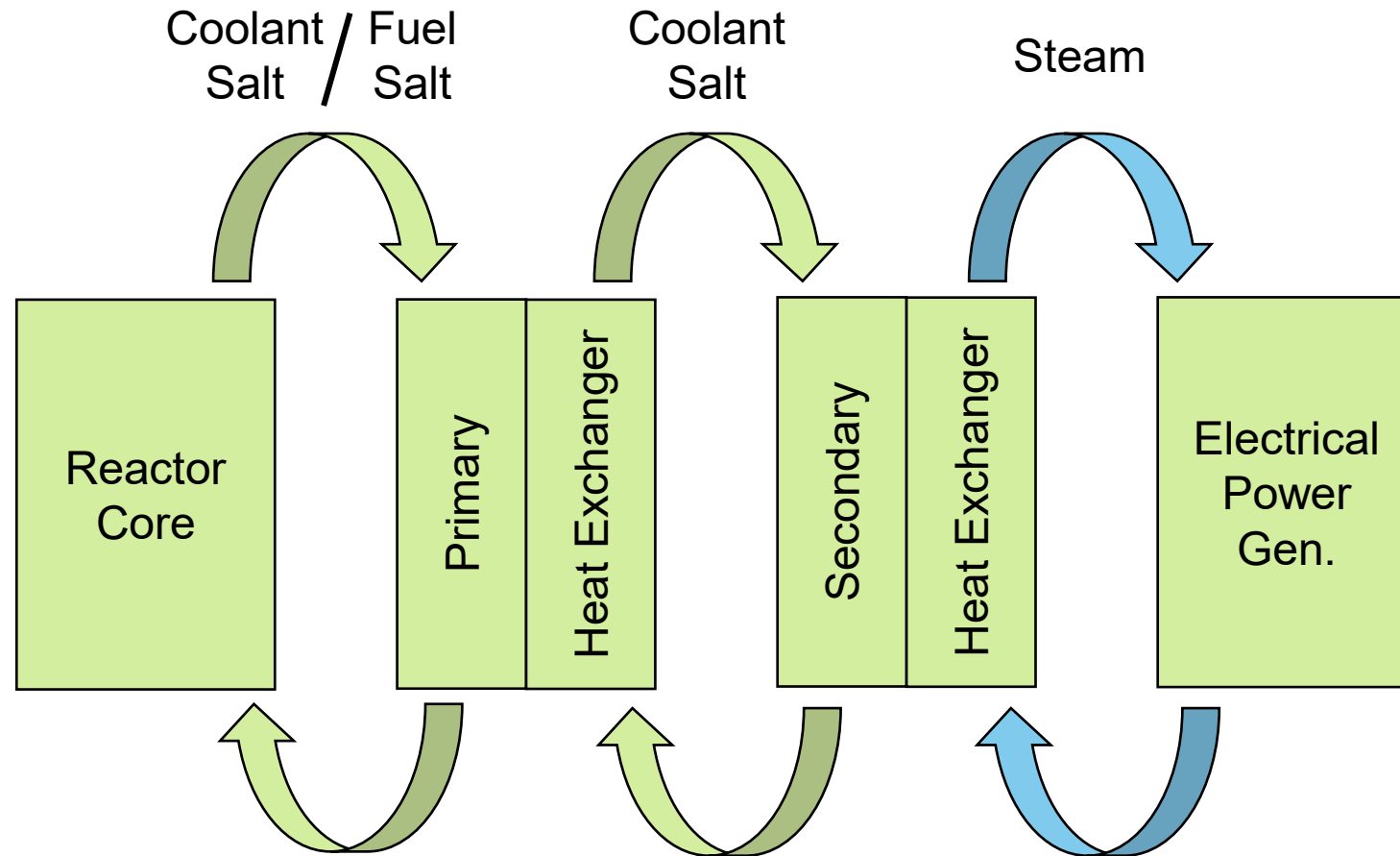
Takeaways: 1 kg d<sup>-1</sup> fission is approximately 1000 MW thermal.  
The mass flowrates of salt, sodium, and water are similar.

# Comparison: Salt, Sodium, Water



Takeaways: The volume flowrate of salt is the lowest because its density is the highest.  
The cycle times of salt and water are similar.  
The SFBR has a large inventory of coolant sodium (pool-type reactor).

# Molten Salt Reactors



Takeaway: There are three categories of salt.  
Primary coolant with fuel, primary coolant without fuel, and secondary coolant.

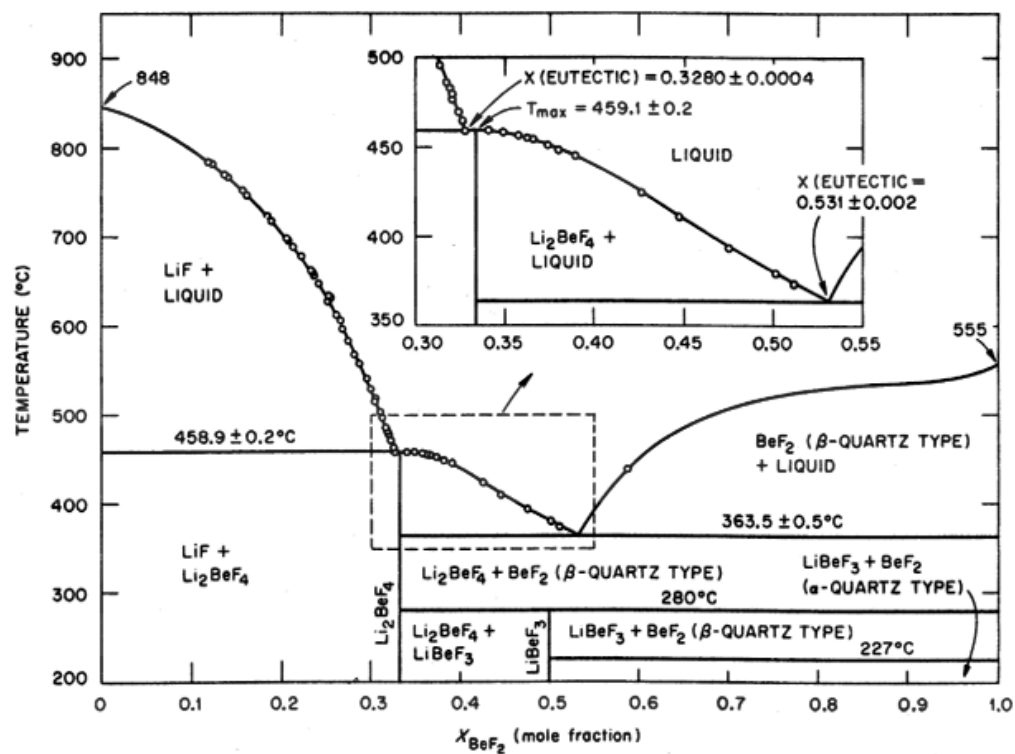
# Properties of Molten Salts: Part 1

- Selection of the Molten Salt System
  - Halide Salt Systems: Fluorides and Chlorides
  - Neutronics
  - Phase Equilibrium
  - Actinide and Fission Product Solubility
  - Chemical Compatibility with Materials of Construction
- Salt Synthesis and Purification
  - Fluorination
  - Chlorination
  - Nuclear Reactor Grade = High Purity

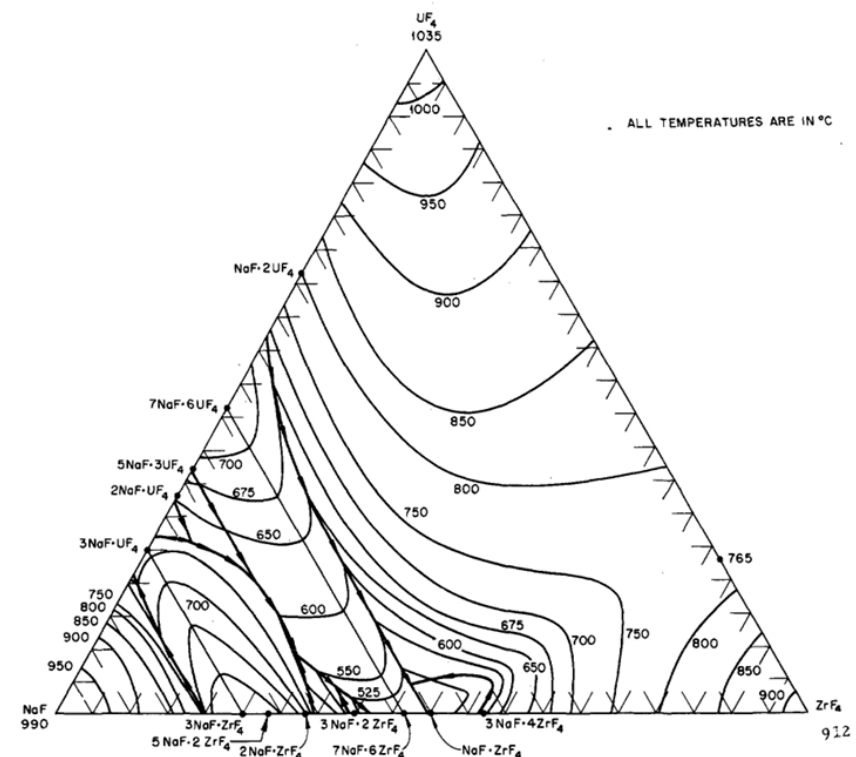
## Examples: Isotopics

- Some isotopes interfere with neutronics and produce undesirable byproducts.
- Natural Li is 7.5 wt%  $^6\text{Li}$  with balance  $^7\text{Li}$ .
- $^6\text{Li}$  can fission into  $^4\text{He}$  and  $^3\text{H}$ .
- Li-containing salts are  $^7\text{Li}$  enriched.
- Natural Cl is 75.8 wt%  $^{35}\text{Cl}$  with balance  $^{37}\text{Cl}$ .
- $^{35}\text{Cl}$  can transmute to  $^{36}\text{Cl}$ .
- $^{36}\text{Cl}$  is a long-lived radionuclide.
- Cl-containing salts are  $^{37}\text{Cl}$  enriched.

# Examples: Phase Equilibrium



$\text{LiF}-\text{BeF}_2$



$\text{NaF}-\text{ZrF}_4-\text{UF}_4$

# Examples: Salt Synthesis

- NaF, LiF
- $\text{BeF}_2$ ,  $\text{ZrF}_4$
- $\text{ThF}_4$
- $\text{UF}_4$
- $\text{PuF}_4$
  
- NaCl, LiCl
- $\text{ThCl}_4$
- $\text{UCl}_3$ ,  $\text{UCl}_4$
- $\text{PuCl}_3$

$\text{PuCl}_3/\text{NaCl}$  Salt Synthesis at INL

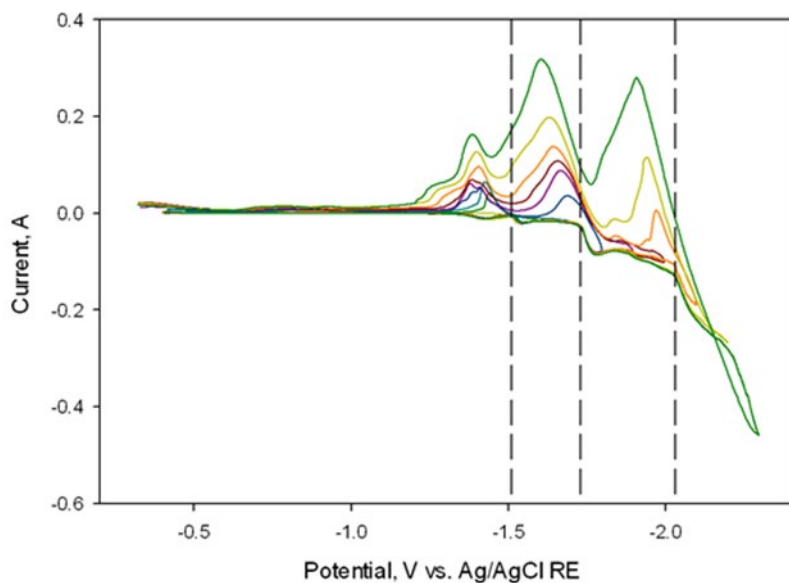


# Properties of Molten Salts: Part 2

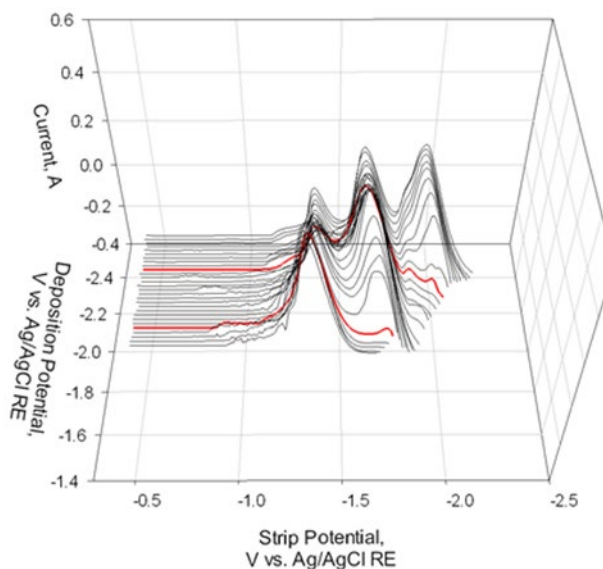
- Reduction/Oxidation (RedOx) Potential
  - Fluorine and Chlorine Chemical Potentials
  - Conceptually  $F_2(g)$  and  $Cl_2(g)$  Partial Pressures
  - Chemical Stabilities
  - Corrosion Control



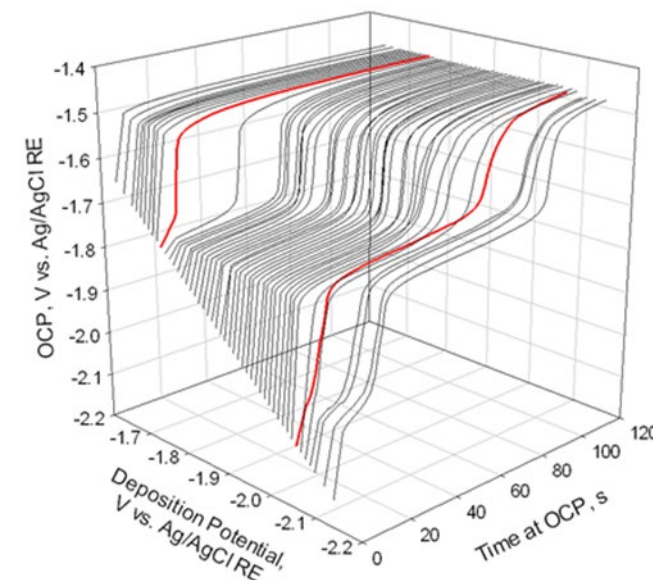
# Examples: Electrochemical Measurements



Cyclic Voltammetry



Cathode Stripping

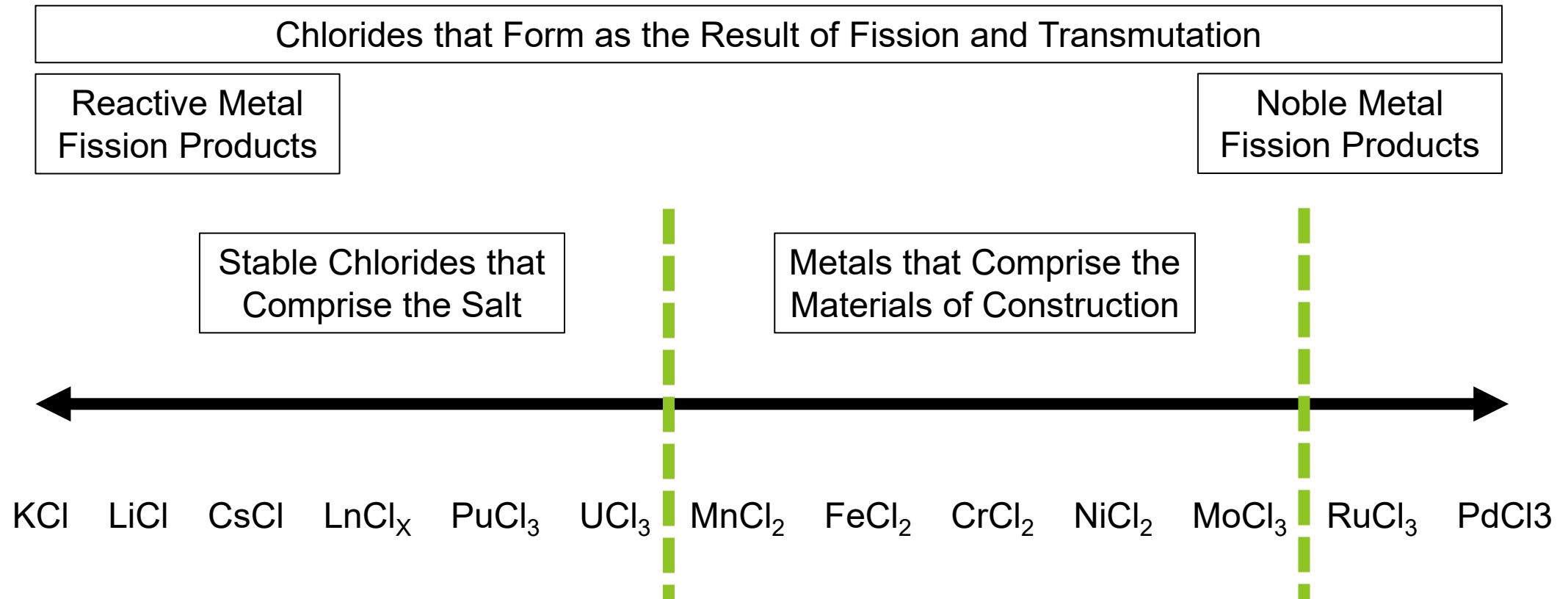


Open Circuit Potential

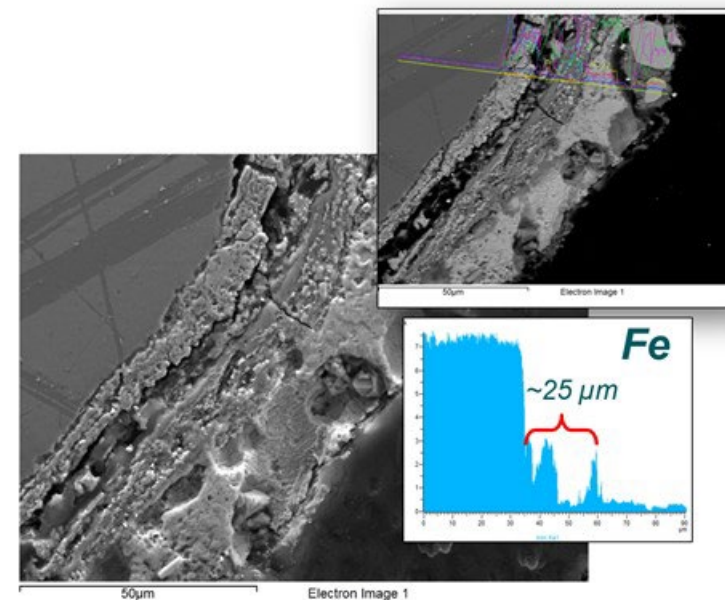
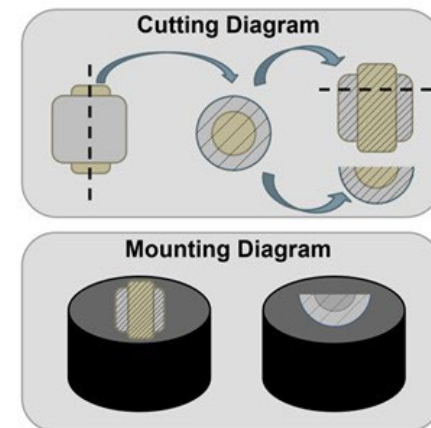
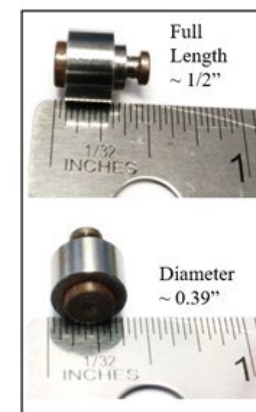
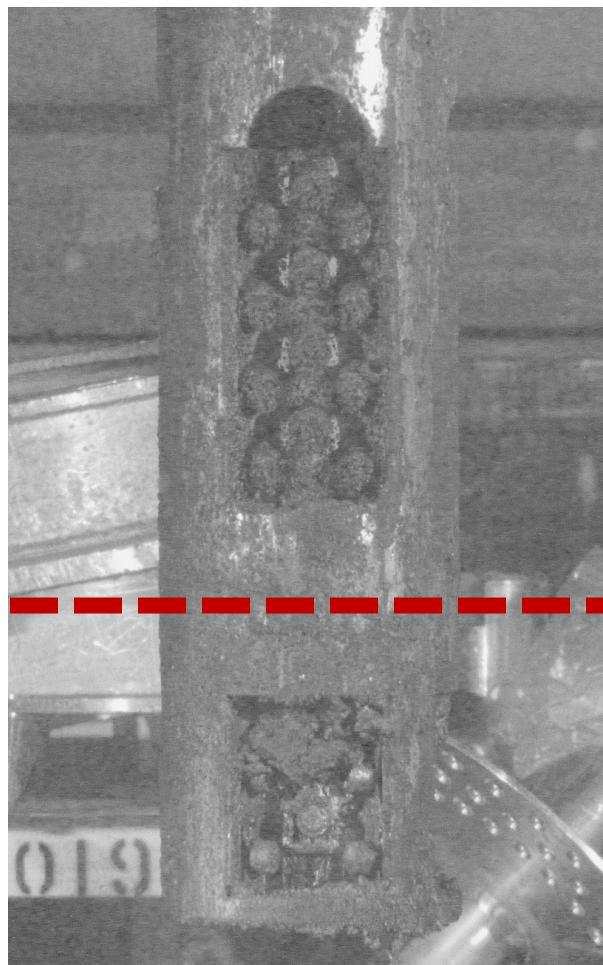
# Examples: Corrosion Mechanisms

- Dissimilar Metals (Galvanic)
  - Macro-Scale
  - Micro-Scale
- Temperature Gradient
  - Molten Salt Flow Loop
- Fission Product Generation (Fuel Salts)
  - Better Part of the Periodic Table

# Examples: Chloride Stability Scale



# Examples: Corrosion



# Properties of Molten Salts: Part 3

- Pycnometer
- Densitometer
- Thermomechanical Analysis
- Transient Hot-Wire
- Rheometer
- Thermogravimetric Analysis (TGA)
- Differential Scanning Calorimetry (DSC)
- Mass Spectroscopy (MS)
- X-Ray Diffraction (XRD)
- Electron Microscopy

Density of Solid

Density of Liquid

Thermal Expansion

Thermal Conductivity

Viscosity

Mass Change

Phase Transitions

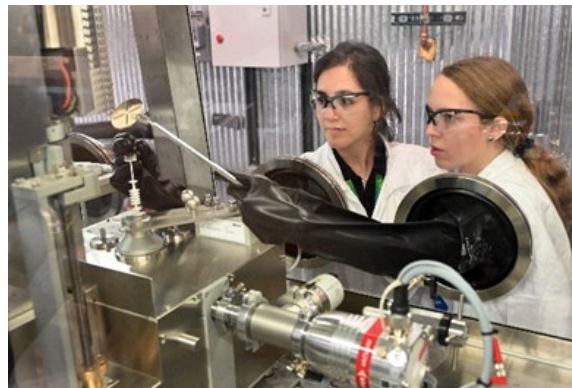
Gas Phase Analyses

Atomic Structure

Phase Analysis

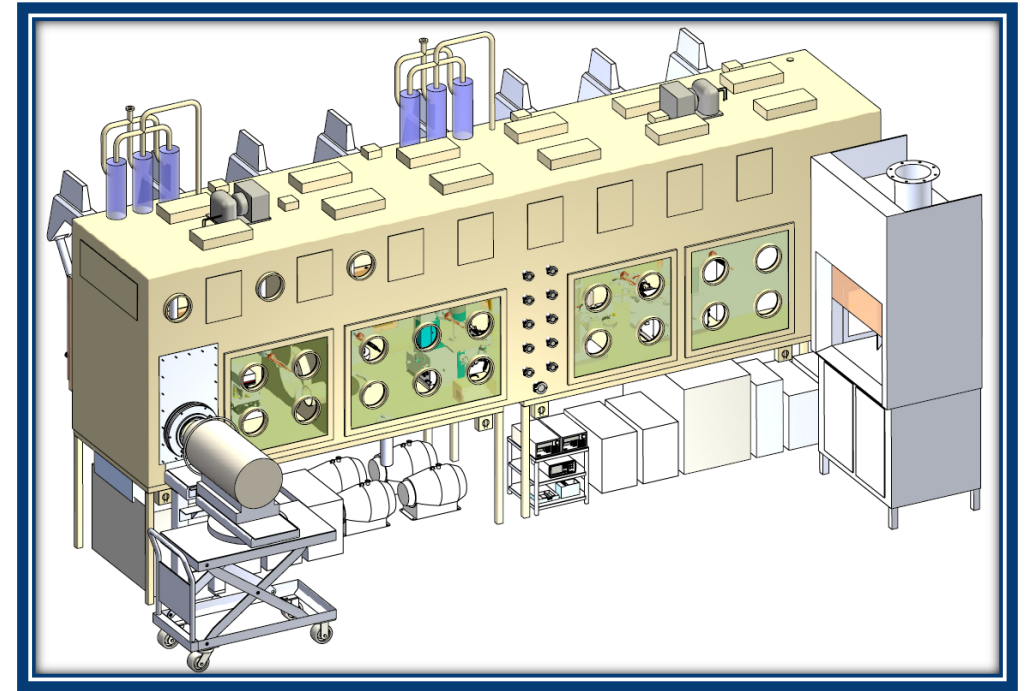
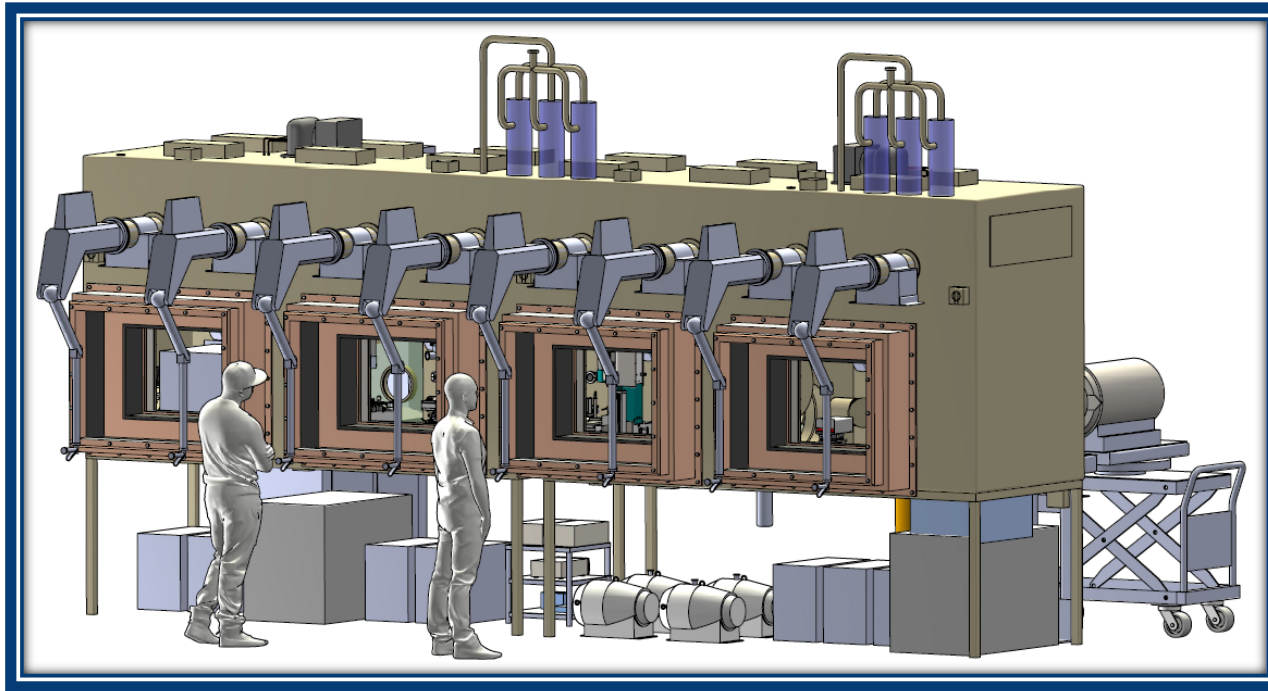
# Examples: Laboratory Facilities

- Characterization equipment located inside Ar-atmosphere gloveboxes.
- Different facilities for four classes of work:
  - Non-Radiological Lab
  - Radiological Lab (work with uranium)
  - Nuclear Lab (work with transuranics)
  - Shielded Enclosures (work with irradiated materials)

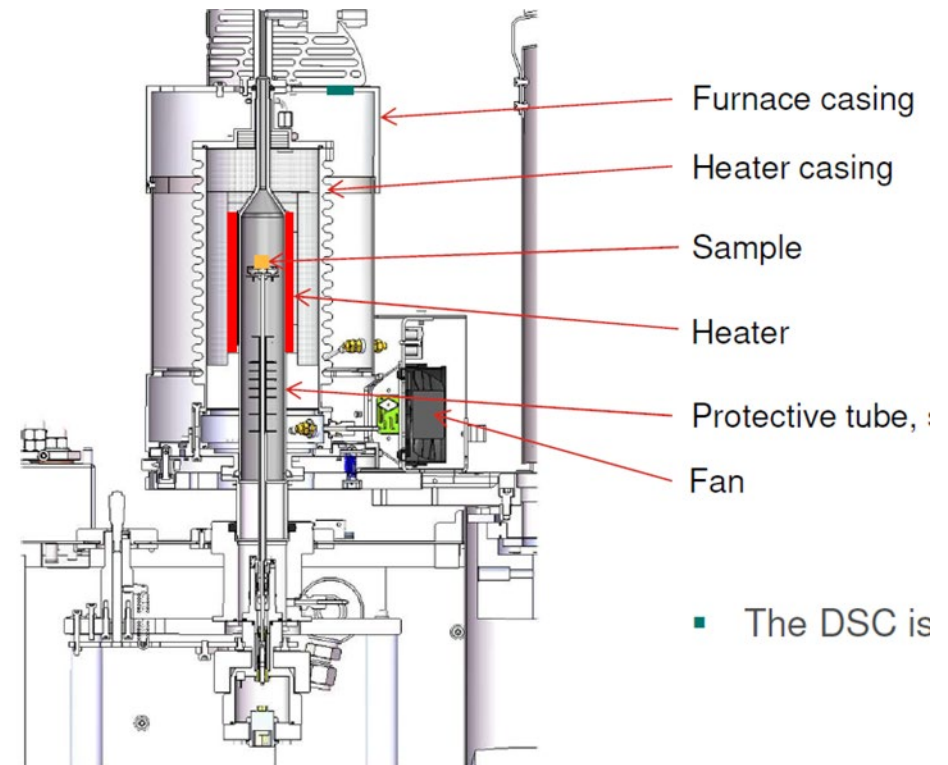
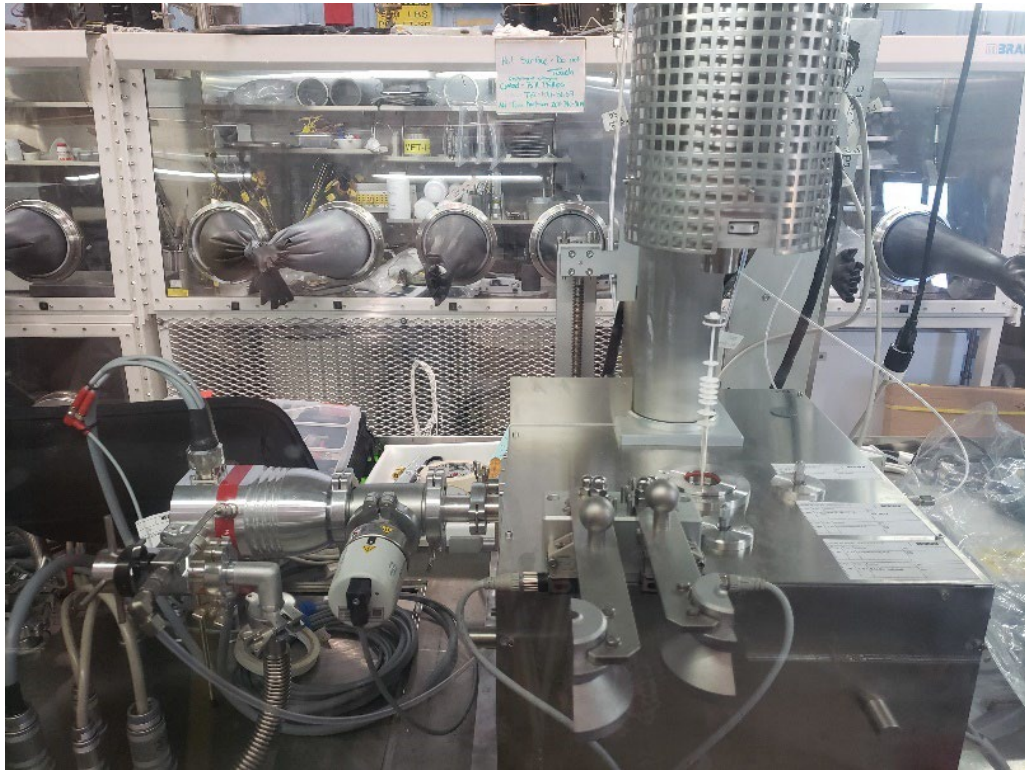




# Examples: Laboratory Facilities

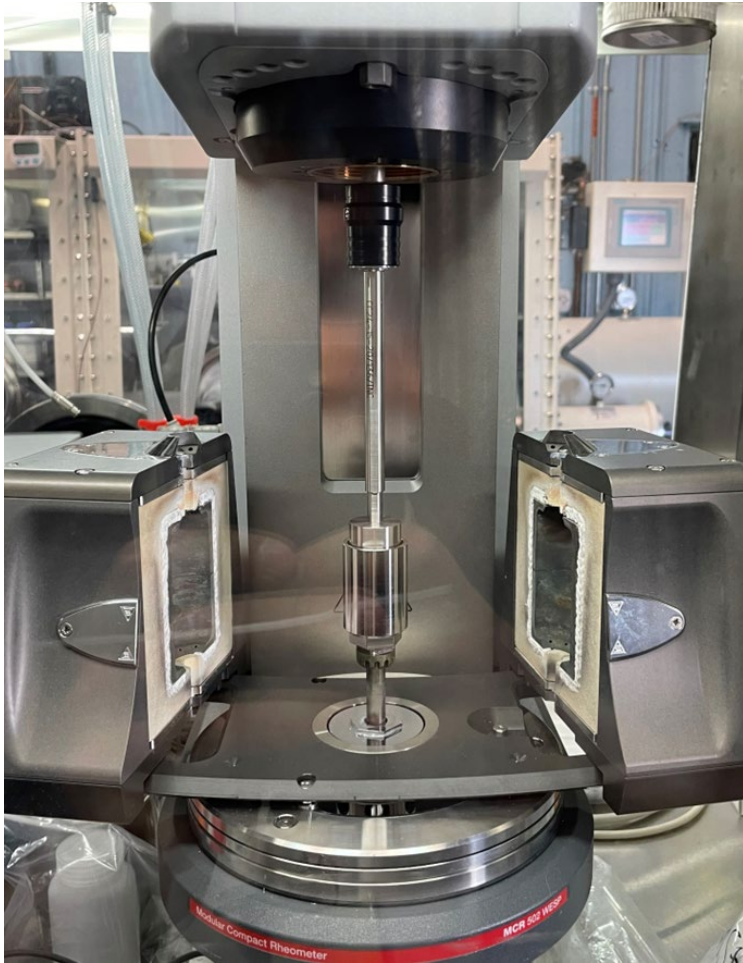


## Examples: TGA / DSC





## Examples: Rheometer





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