



Recent Advances in Radiation-Induced Actinide Redox Chemistry

September 2022

Changing the World's Energy Future

Gregory P Horne



INL is a U.S. Department of Energy National Laboratory operated by Battelle Energy Alliance, LLC

DISCLAIMER

This information was prepared as an account of work sponsored by an agency of the U.S. Government. Neither the U.S. Government nor any agency thereof, nor any of their employees, makes any warranty, expressed or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness, of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. References herein to any specific commercial product, process, or service by trade name, trade mark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the U.S. Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the U.S. Government or any agency thereof.

Recent Advances in Radiation-Induced Actinide Redox Chemistry

Gregory P Horne

September 2022

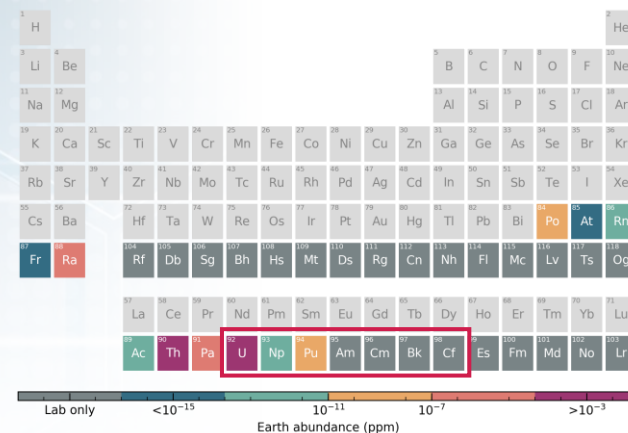
**Idaho National Laboratory
Idaho Falls, Idaho 83415**

<http://www.inl.gov>

**Prepared for the
U.S. Department of Energy
Under DOE Idaho Operations Office
Contract DE-AC07-05ID14517**

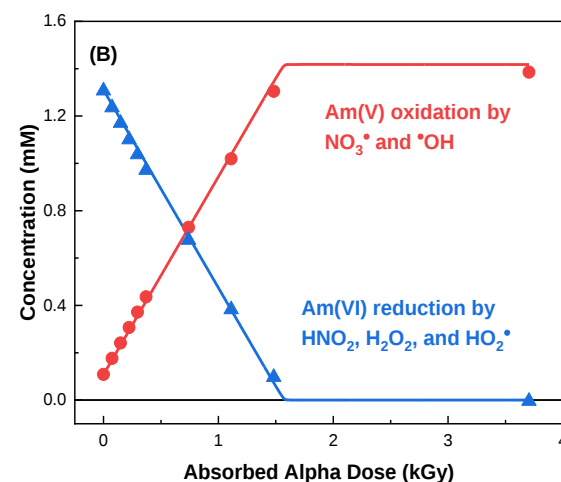
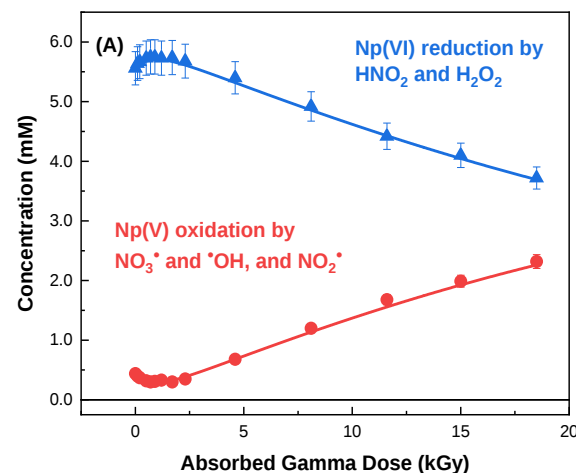
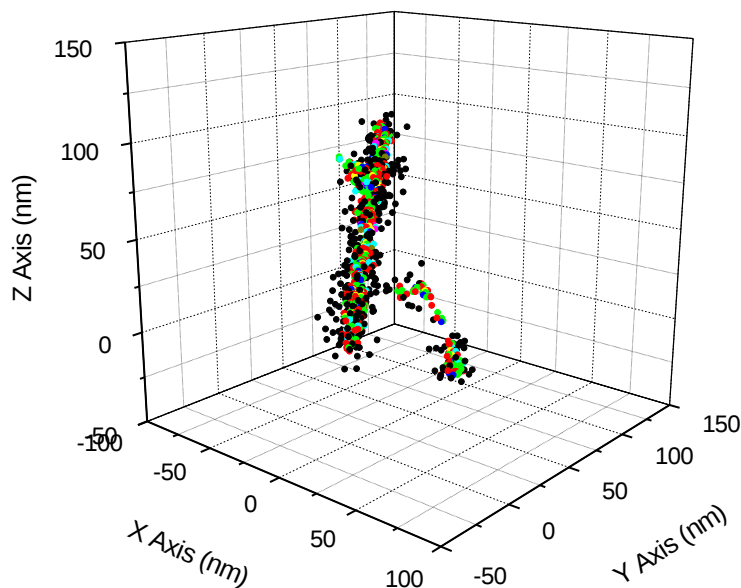
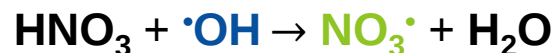
Gregory P. Horne
Center for Radiation
Chemistry Research

Recent Advances in Radiation-Induced Actinide Redox Chemistry



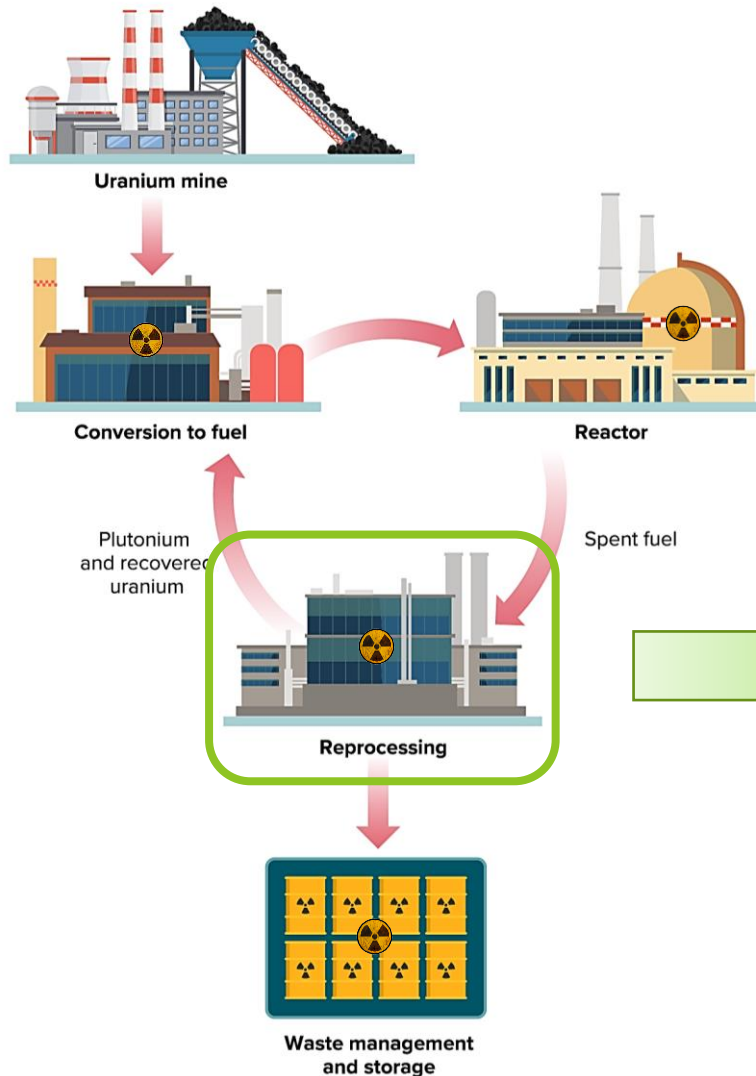
September 26-29, 2022 | Avignon, France

Actinide Radiation Chemistry

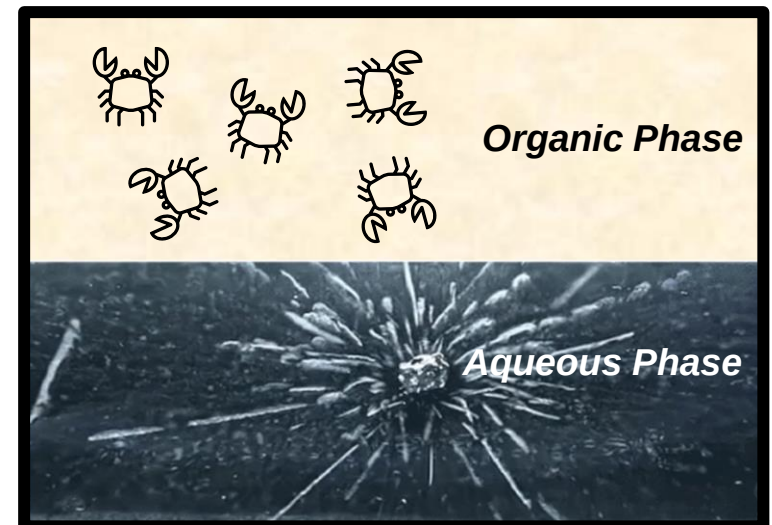


- Gordon, Sullivan, and Ross, *J. Phys. Chem. Ref. Data* **1986**, 15, 1357.
- Pikaev, Gogolev, and Shilov, Fedoseev, *Isotopenpraxis*, **1990**, 26, 465.
- Mincher and Mezyk, *Radiochim. Acta* **2009**, 97, 519.
- Buxton, Greenstock, Helman, and Ross, *J. Phys. Chem. Ref. Data* **1988**, 17, 513.
- Katsumura, *The Chemistry of Free Radicals: N-Centered Radicals*, John Wiley & Sons, Chichester, **1998**.
- Horne, Grimes, Mincher, and Mezyk, *J. Phys. Chem. B* **2016**, 120 (49), 12643.
- Horne, Grimes, Bauer, Dares, Pimblott, Mezyk, and Mincher, *Inorg. Chem.* **2019**, 58, 8551.

Reprocessing Used Nuclear Fuel



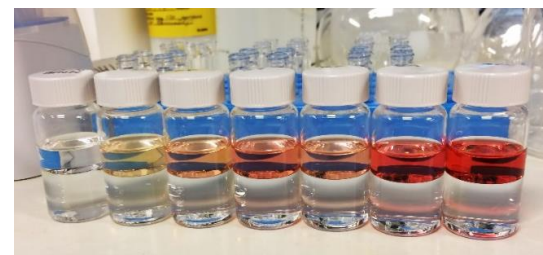
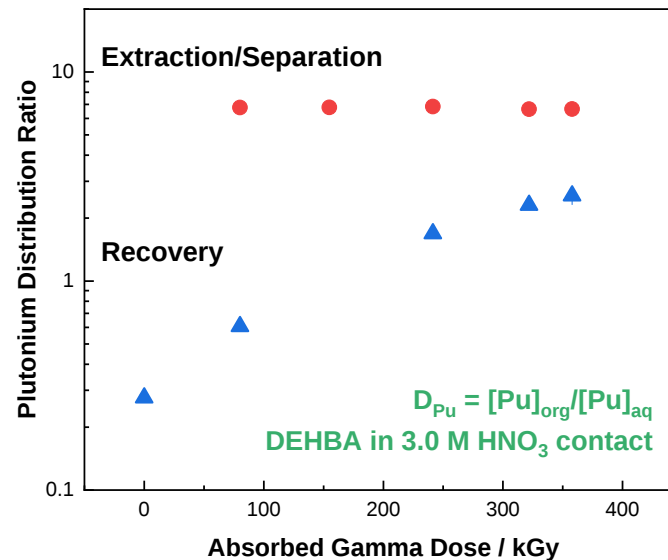
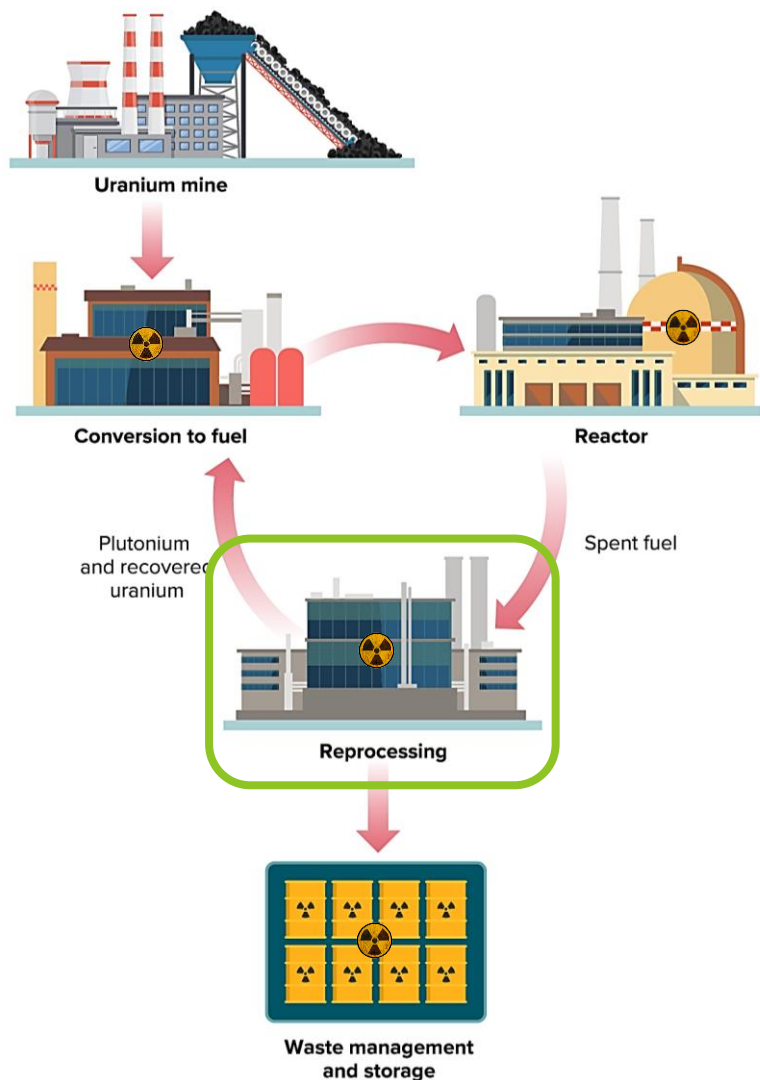
Solvent Extraction Reprocessing
Ligands/**organic diluent**: $\text{HNO}_3/\text{H}_2\text{O}$
($\pm$ additives)



• Strategies and Considerations for the Back End of the Fuel Cycle, Nuclear Technology Development and Economics, NEA No. 7469, 2021.

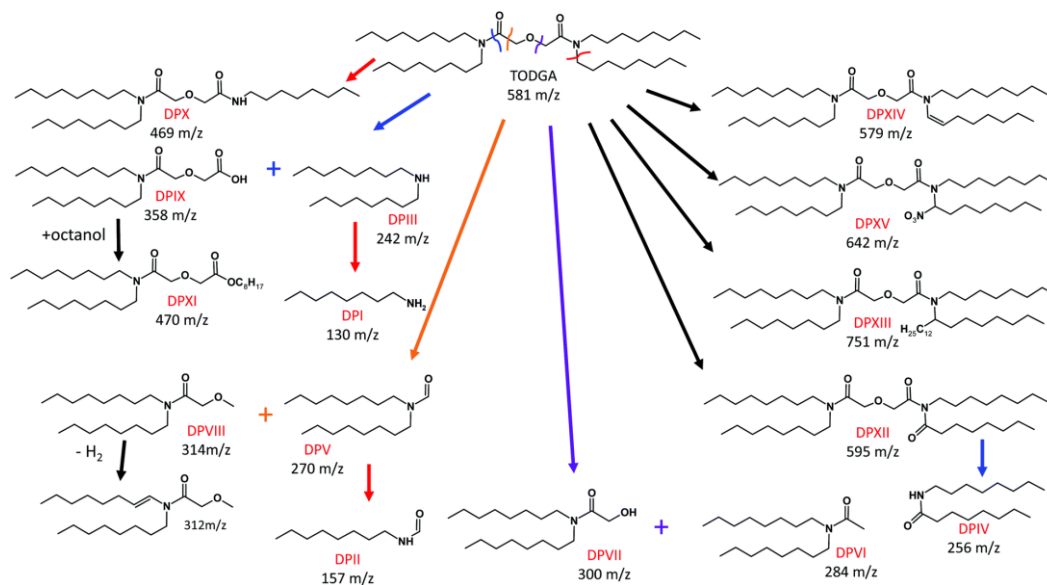
• Bruffey *et al.*, Innovative Separations R&D Needs for Advanced Fuel Cycles Workshop, August 30-September 1, 2021. Report for the US Department of Energy, Office of Nuclear Energy Workshop on Innovative Separations R&D Needs for Advanced Fuel Cycles, 2022.

Reprocessing Used Nuclear Fuel

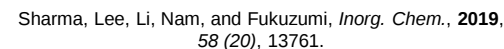


Increasing Gamma Dose $\rightsquigarrow \rightsquigarrow$

- Horne, Zarzana, Grimes, Rae, Ceder *et al.*, *Dalton Trans.*, **2019**, 48, 14450.
- Horne, Mezyk, Mincher, Zarzana, Rae, Tillotson, Schmitt, Ball, Ceder, Charbonnel, Guilbaud, Saint-Louis, and Berthon, *Rad. Phys. Chem.*, **2020**, 170, 108608.

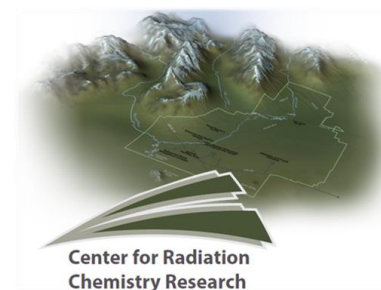


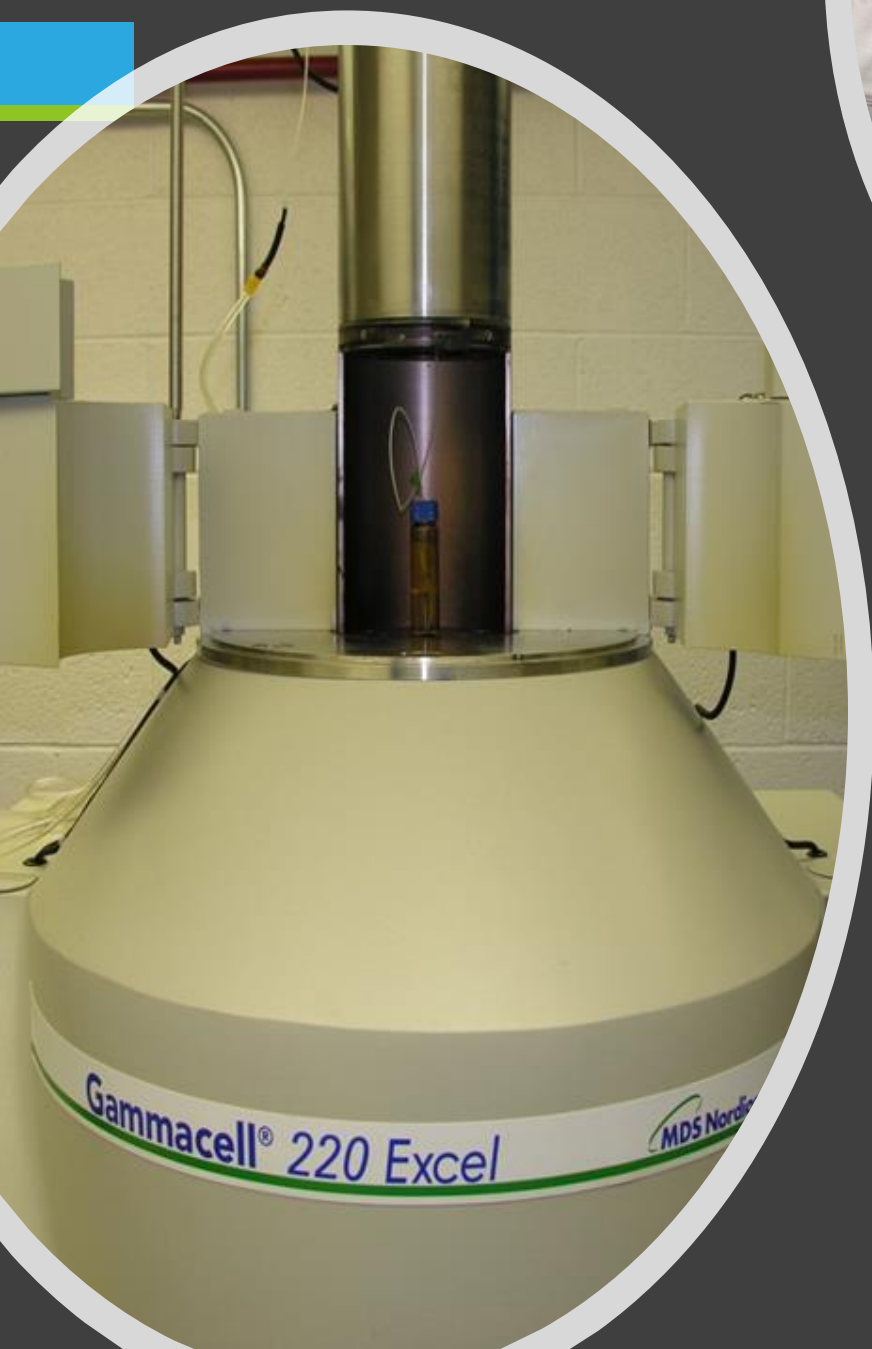
“...in the presence of macroconcentration of lanthanides and actinides, TODGA degradation by radiolysis is minimal and does not generate problematic degradation products.” Kimberlin *et al.*, *PCCP*, **2022, 24, 9213.**



Research Goals

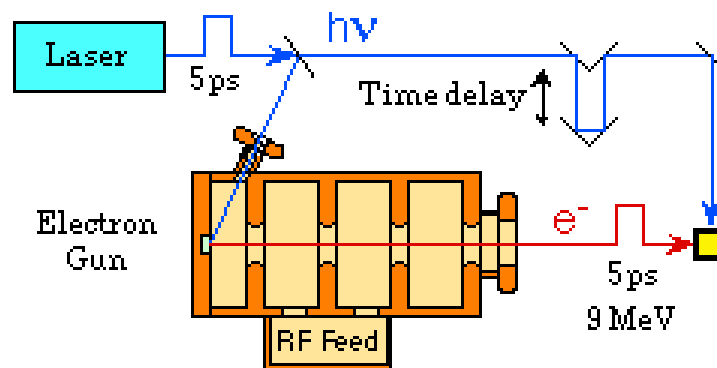
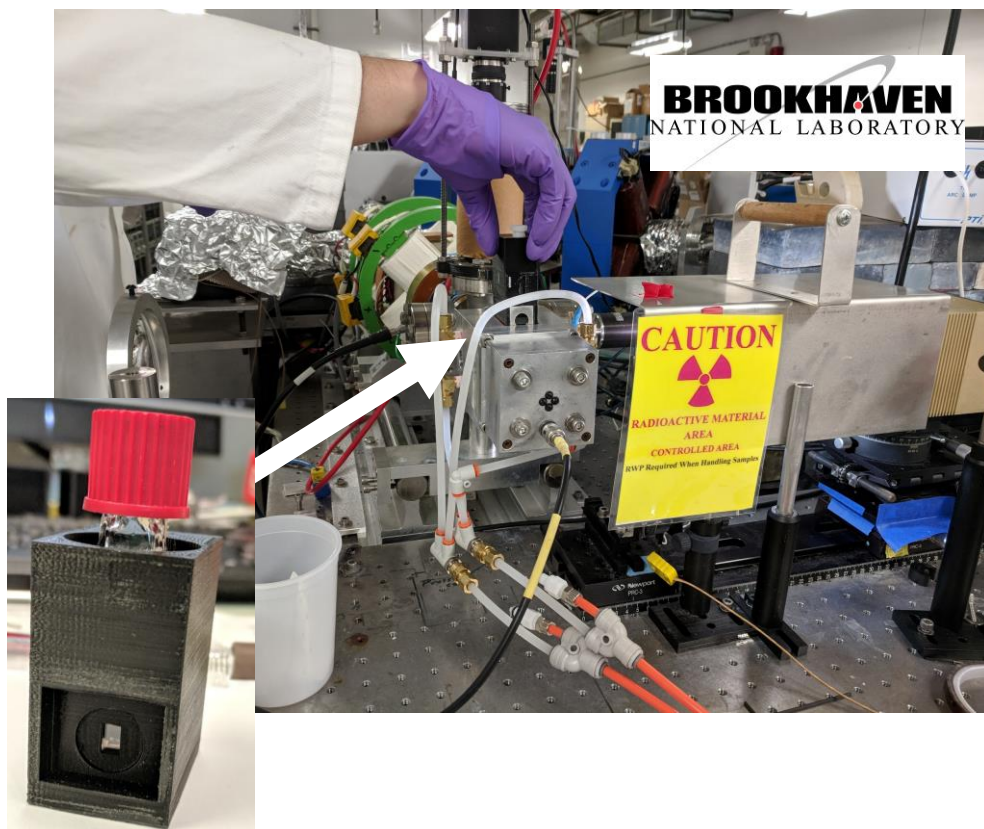
1. Understand the basic radiation chemistry of the actinides in formally non-complexing media.
2. Elucidate the mechanisms underpinning the impact of actinide complexation on ligand radiolysis.
3. Develop multiscale modeling codes for the prediction of radiation-induced chemistry, speciation, and transport of the actinide series in any media.



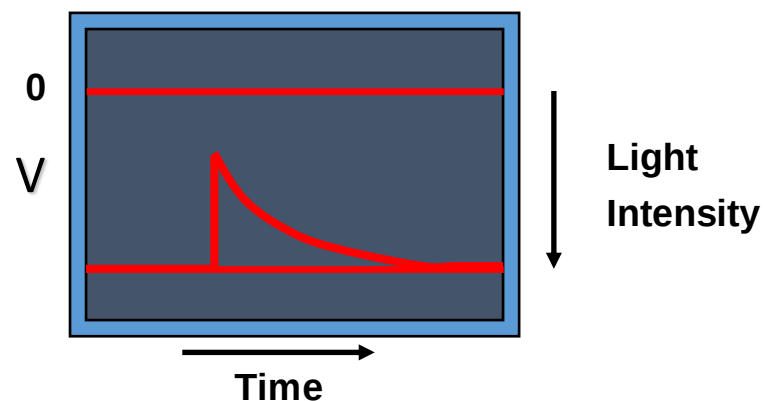


Radiation Chemistry Techniques

Electron Pulse Radiolysis

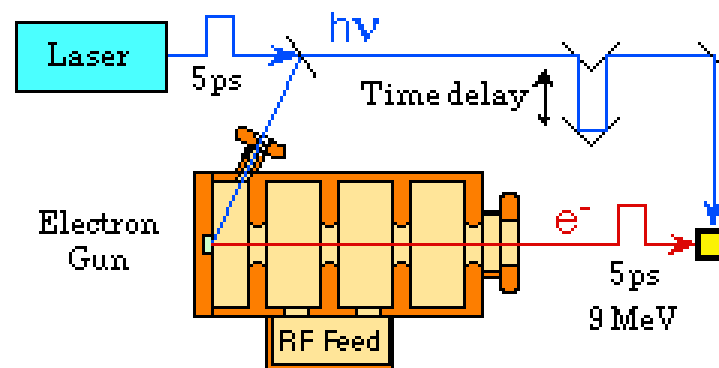


Transients are detected by optical absorption changes.

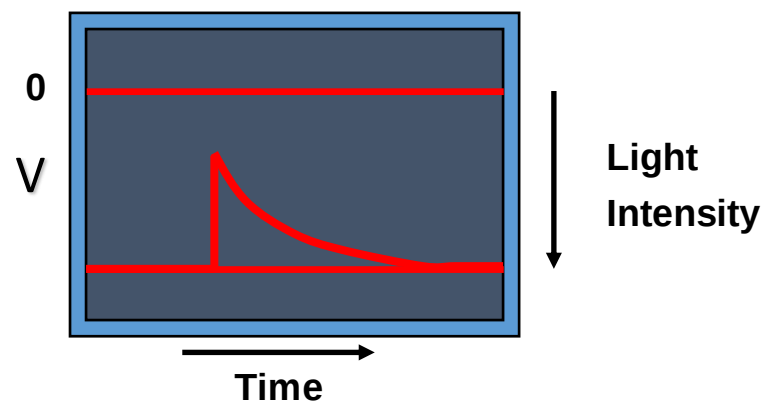


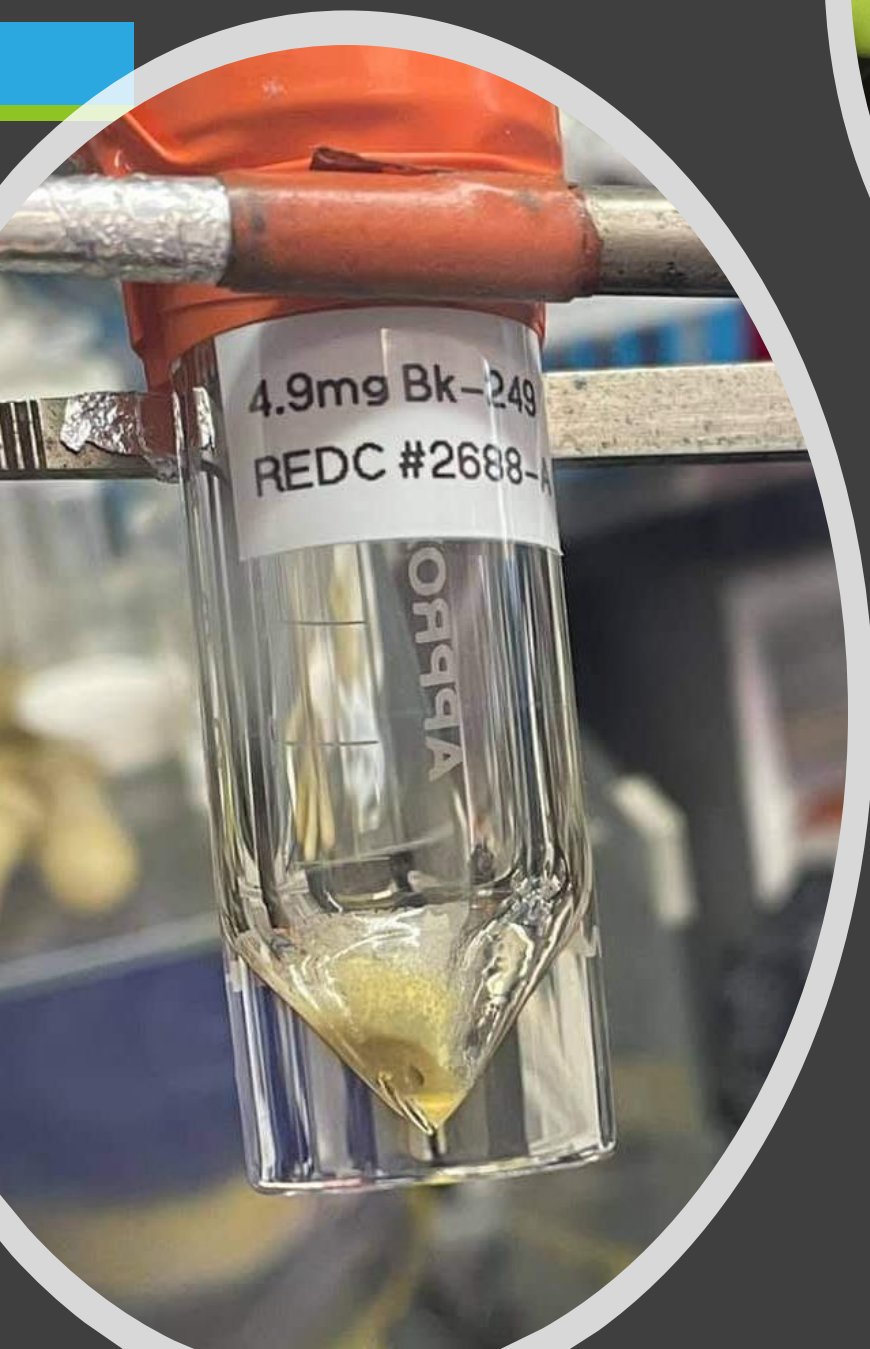
Radical Isolation

- **Hydrated Electron (e_{aq}^-):**
HClO₄/0.5 M tBuOH/N₂-saturated.
- **Hydrogen Atom ($H^\bullet$):** ~ 100 μM
PCB/HClO₄/50 mM tBuOH/N₂-saturated.
- **Hydroxyl Radical ($\bullet OH$):** ~ 100 μM
KSCN/10 mM HClO₄/N₂O-saturated.
- **Nitrate Radical ($\bullet NO_3$):** ~ 6.0 M
HNO₃/N₂O-saturated.
- **Dodecane Radical Cation ($RH^{\bullet+}$):**
0.5 M DCM/*n*-dodecane.



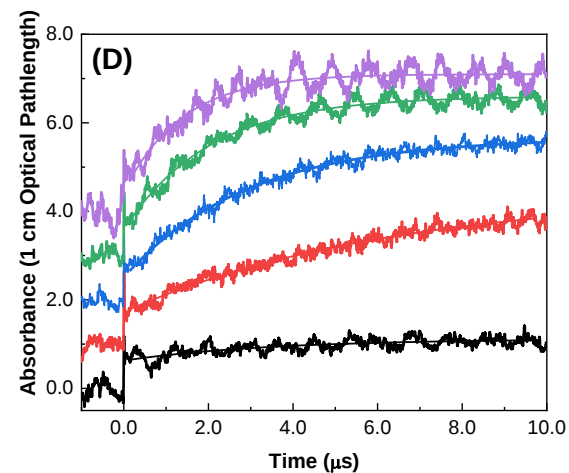
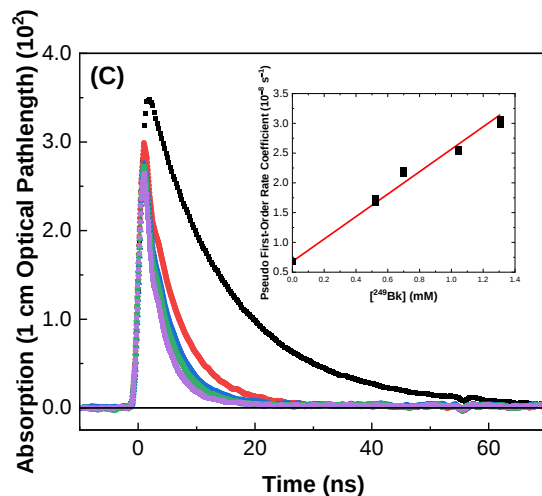
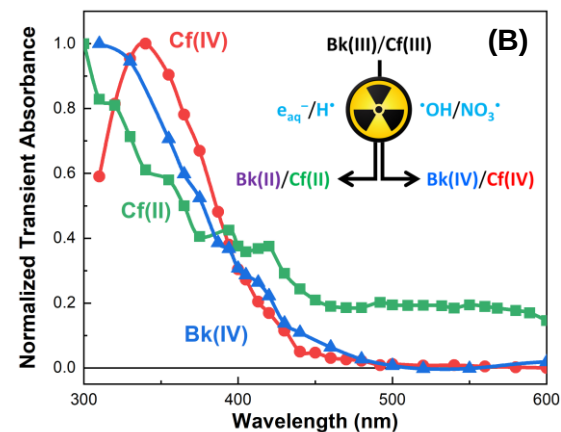
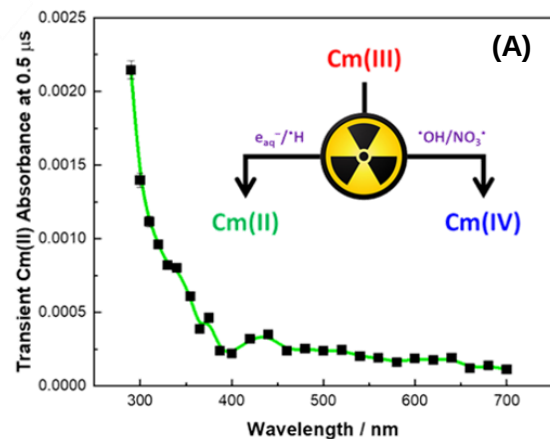
Transients are detected by optical absorption changes.



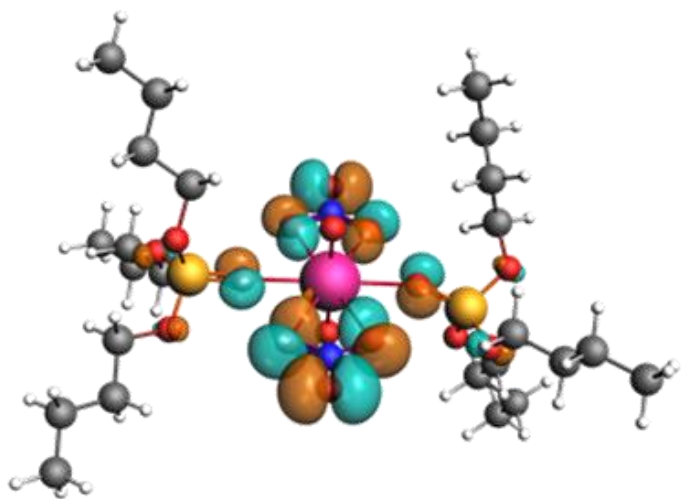
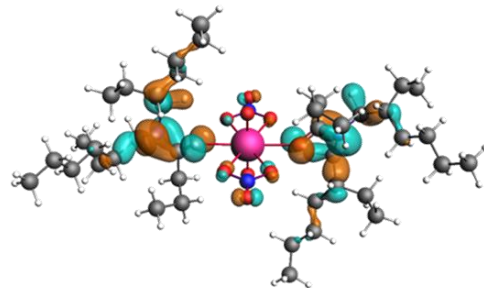


Radiation-Induced Actinide Redox Chemistry

Non-Traditional Oxidation States...

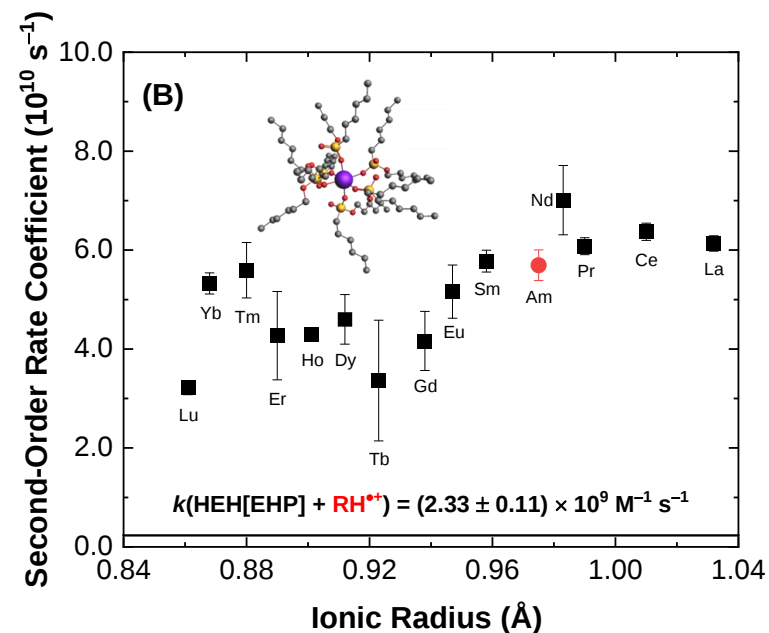
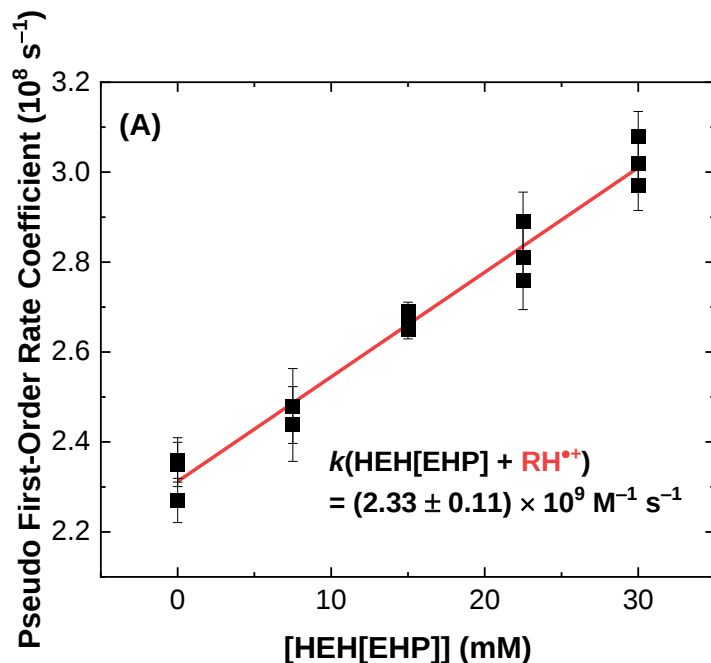


- Horne, Grimes, Zalupski, Meeker, Albrecht-Schönzart, Cook, and Mezyk, *Dalton Trans.* **2021**, 50, 10853.
- Horne, Rotermund, Grimes, Sperling, Meeker, Zalupski, Beck, Gomez Martinez, Beshay, Peterman, Layne, Johnson, Cook, Albrecht-Schönzart, and Mezyk, *Inorg. Chem.*, **2022**, 61 (28), 10822.



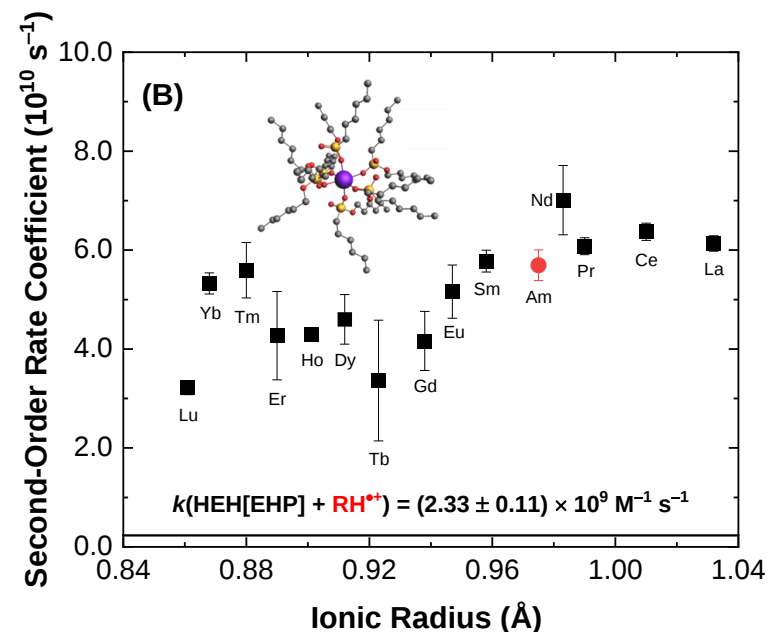
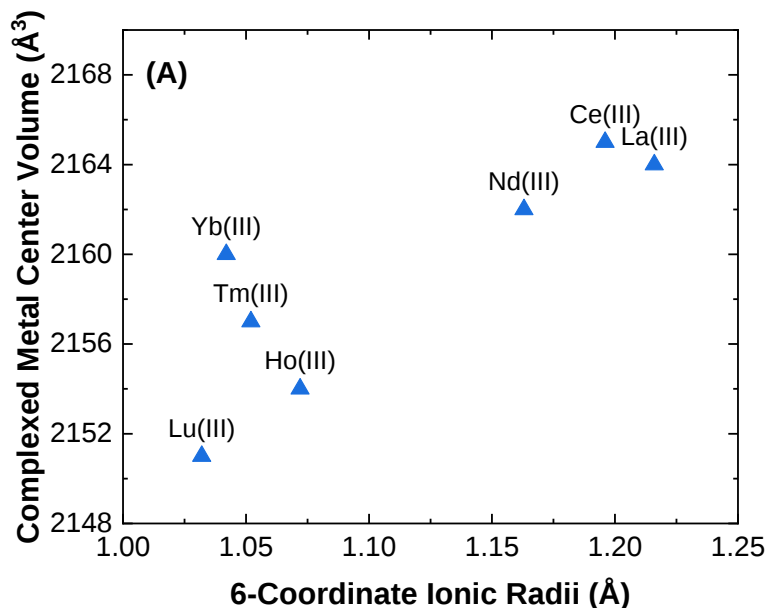
Impact of *f*-element Complexation on Radiolytic Behavior

f-element Complexes of HEH[EHP]



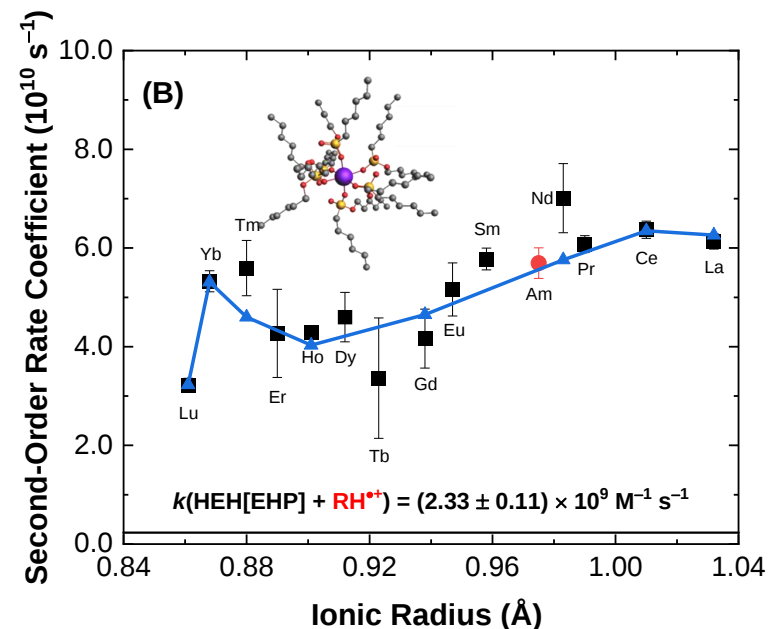
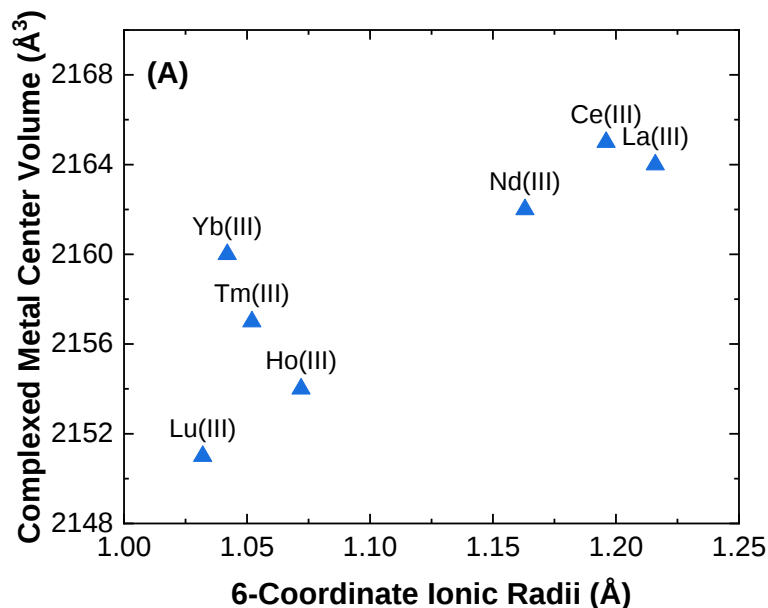
- Methodology:** $\Delta[\text{HEH}[\text{EHP}]]/[\text{M}\{(\text{HEH}[\text{EHP}])_2\}_3]$ in 0.5 M DCM/*n*-dodecane; RH^{2+} decay measured at 800 nm over 200 ns using the BNL *Laser Electron Accelerator Facility* (LEAF).

Size Matters...



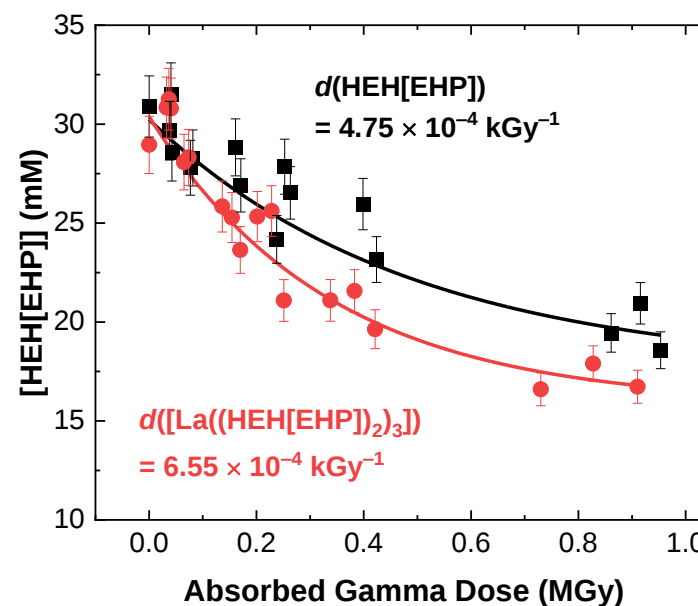
- Computations:** Geometry optimization using ADF2020 (GGA *OPBE* functional + *ZORA/STO-TZP* basis set for all atoms); Wave functions using ORCA (*Hybrid PBE0* functional + *DKH-def2-SVP/DKH-def2-TZVP/SARC-DKH-TZVP*); and QTAIM metrics obtained from DFT wave functions using the AIMALL software.

Size Matters...



- Computations:** Geometry optimization using ADF2020 (GGA *OPBE* functional + *ZORA/STO-TZP* basis set for all atoms); Wave functions using ORCA (*Hybrid PBE0* functional + *DKH-def2-SVP/DKH-def2-TZVP/SARC-DKH-TZVP*); and QTAIM metrics obtained from DFT wave functions using the AIMALL software.

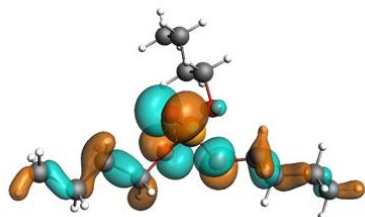
Steady-State Impact?



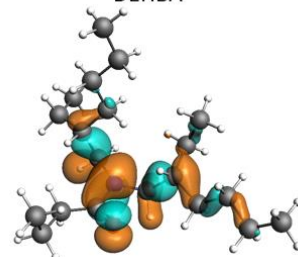
- **Steady-State Gamma Irradiations:** Cobalt-60 irradiation of 30 mM HEH[EHP] under organic only (■) and loaded with 2.5 mM La(III) from PIPPS/HNO₃ solution (●). Dose constants (d) were calculated from linear fits to $[\text{La}((\text{HEH}[\text{EHP}])_2)_3]$ vs. absorbed gamma dose.

Uranyl Complexes of TBP, DEHBA, and DEHiBA

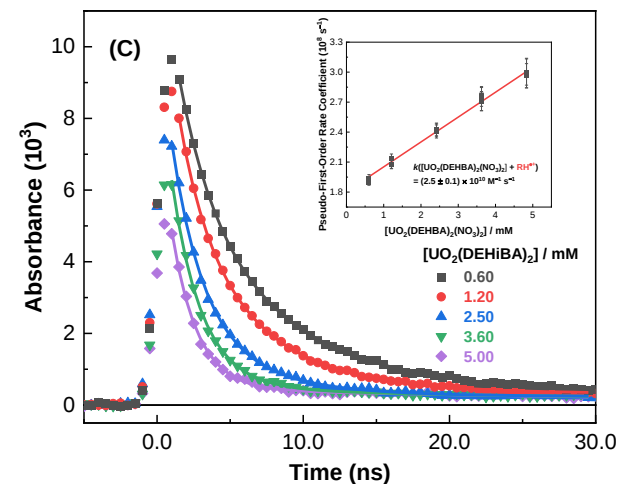
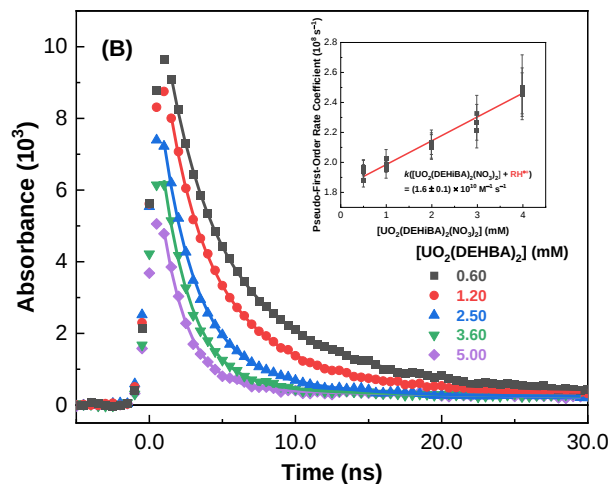
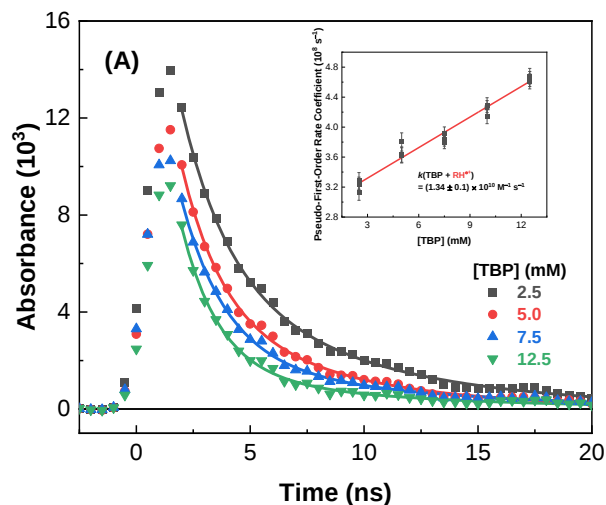
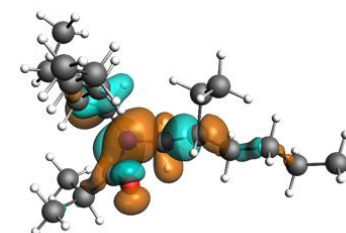
TBP^{•+}



DEHBA^{•+}



DEHiBA^{•+}



- UO_2^{2+} complexation had negligible effect on the reaction of TBP with $\text{RH}^{\bullet+}$, $k(\text{TBP} + \text{RH}^{\bullet+}) = (1.3 \pm 0.1) \times 10^{10} \text{ M}^{-1} \text{ s}^{-1}$.
- For DEHBA and DEHiBA, UO_2^{2+} complexation afforded a 2.6× and 1.4× increase in their respective rate coefficients, respectively.

Changes in Reaction Mechanism

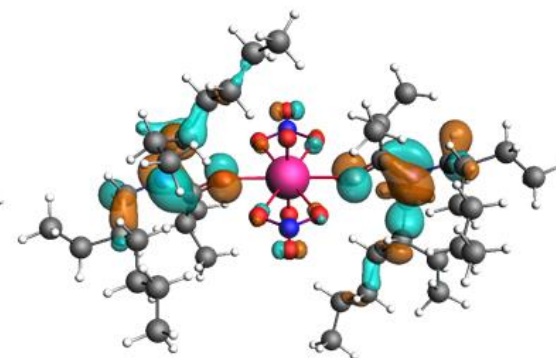
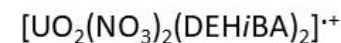
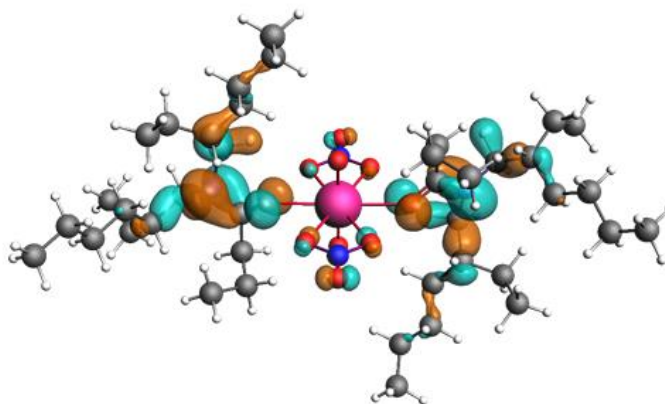
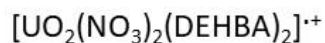
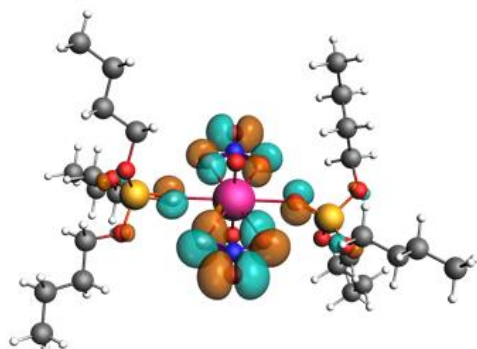
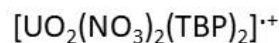
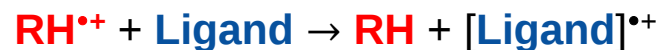


Table 1 Electronic structure calculation free energy (ΔG) values for the reaction of $\text{RH}^{\bullet+}$ with TBP, DEHBA, and DEHiBA for electron/hole transfer and proton transfer scenarios

Ligand	$\Delta G_{\text{electron/hole transfer (eV)}}$	$\Delta G_{\text{proton transfer (eV)}}$
TBP	0.16	-0.36
DEHBA	-0.88	-0.60
DEHiBA	-0.90	-0.57

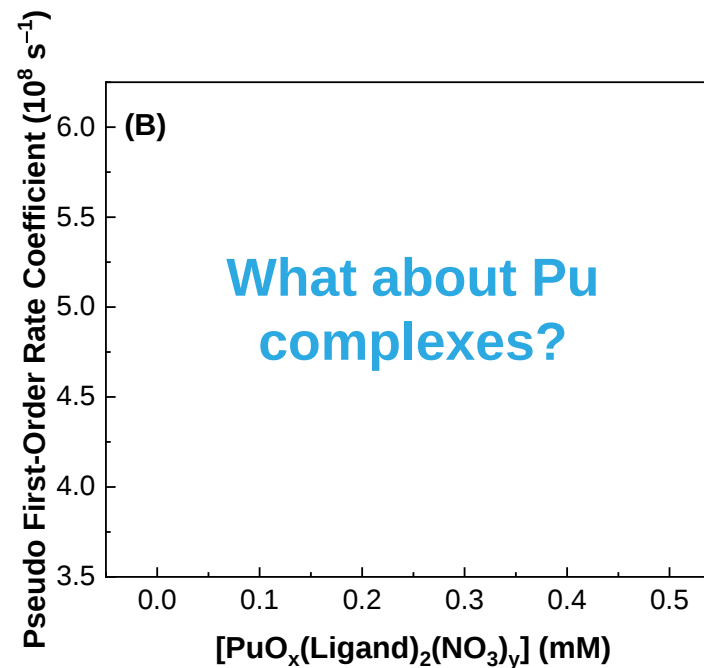
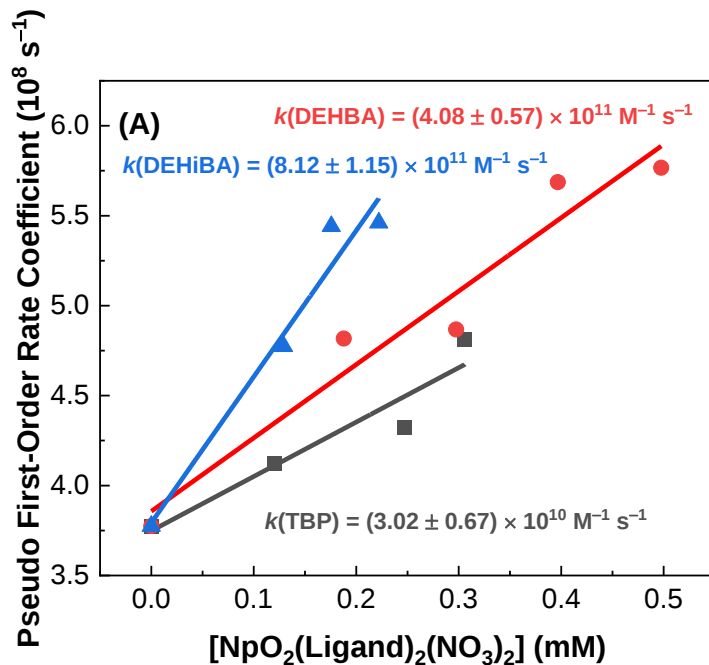
Electron/Hole Transfer



Proton Transfer



Transuranic Complexation...



- NpO_2^{2+} complexation by **TBP** afforded $\sim 3\times$ faster rate of reaction with **$\text{RH}^{\bullet+}$** , and over an order of magnitude increase for complimentary **DEHBA** and **DEHiBA** complexes.
- Evidence for electron transfer with the complexed metal center?

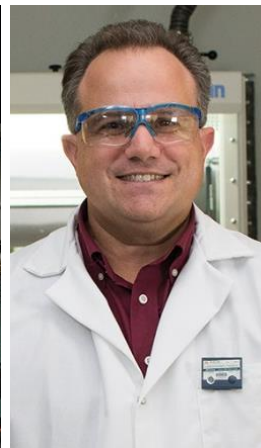
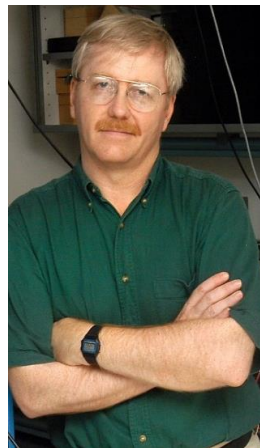
Conclusions

- Actinide radiation chemistry is as fundamental to the actinides as their $5f$ electrons.
- “Unbound” actinides attain exotic radiation-induced transient oxidation states with lifetimes sufficient to propagate chemistry.
- Actinide complexation has significant effects on the radiation robustness of their complexes.
- **Overall, the presented data provide important missing information that will augment current and future studies into the unique chemistry of these elements and their periodicity.**

Acknowledgements



U.S. DEPARTMENT OF
ENERGY



CALIFORNIA STATE UNIVERSITY
LONG BEACH



FLORIDA STATE
UNIVERSITY

BROOKHAVEN
NATIONAL LABORATORY



Material
Recovery &
Waste Form
Development



IDAHO NATIONAL LABORATORY



Idaho National Laboratory

New Postdoctoral/Staff Scientist Position(s)

- The INL *Aqueous Separations and Radiochemistry* group and the *Center for Radiation Chemistry Research* are seeking to add an enthusiastic, self-motivated Post-Doctoral Research Fellow and/or Staff Scientist to their team.
- An ideal candidate(s) would have previous experience with:
 - organic and/or inorganic synthesis as well as characterization techniques used for validation;
 - previous familiarity with vitrification, radiation, and/or actinide chemistry is favorable;
 - and seeking funds through grant proposal writing.

If interested, please contact either **Colt Heathman** (colt.heathman@inl.gov), **Corey Pilgrim** (corey.pilgrim@inl.gov), **Melissa Warner** (melissa.warner@inl.gov), and/or myself (gregory.horne@inl.gov) for more information.