High Temperature Fuel Cladding Chemical Interactions Between Unirradiated TRIGA Fuels and 304 Stainless Steel

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Introduction



- High temperature chemical interactions between TRIGA (Training, Research, Isotopes, General Atomics) fuels and their type-304 stainless steel (304SS) claddings during irradiation and different TRIGA reactor transient scenarios are of interest from the standpoint of safe reactor operation.
- During irradiation and high temperature exposure, the fuel may come into close contact with the 304SS cladding, and an opportunity exists for interdiffusion to take place between fuel and 304SS.
- Annealing experiments using diffusion couples have been used to mimic this scenario.
- Interactions within the 304SS are given a priority.

Experimental Details: Fuel

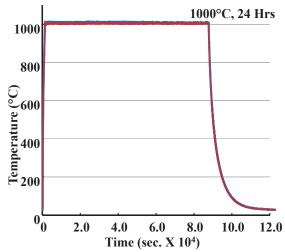




- Annealing experiments for this study were carried out in a furnace placed inside of a Ar-atmosphere glovebox.
 - Thermocouples monitored and logged temperatures.
 - A trickle flow (0.5 ft³/hr) of high purity (<10ppb) Ar gas maintained low O₂ levels inside of the furnace.
 - Ta-foil was used to provide an additional O₂ barrier.
- TRIGA fuel samples, ¼" X ¼" X 1/8" pieces, and ¼" diameter rods of 304SS were used these experiments.

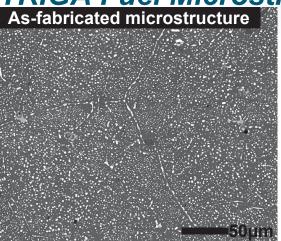
- Fuel samples were annealed at 950°C for 5, 16 and 24 hours to study the fuel evolution as a function of time.
- Diffusion couples of TRIGA fuel vs. 304SS were assembled and annealed at 1000°C for 5 and 24 hours.
- The microstructures of as-fabricated fuel, annealed fuels and the diffusion couples were examined by:
 - Scanning electron microscopy (SEM):
 - Secondary (SE) and backscatter (BE) electron imaging with compositional analysis by energy dispersive spectroscopy (EDS) and X-ray maps.
 - Transmission Electron Microscopy (TEM) and Scanning TEM (STEM):
 - Site-specific specimen preparation via Focused Ion Beam In-Situ Lift-Out (FIB-INLO)
 - Bright Field and High Angle Annular Dark Field (HAADF) imaging.
 - Selected Area Electron Diffraction (SAED).
 - X-Ray Diffraction (XRD)
 - Micro-XRD was used as a complementary characterization method.
- The presence of hydrogen was inferred from the crystallographic analysis.

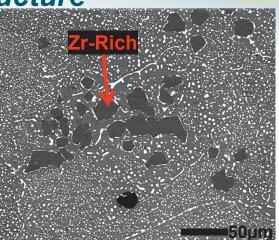


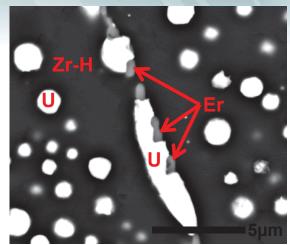


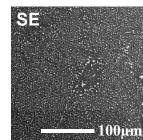
As-Fabricated TRIGA Fuel Microstructure

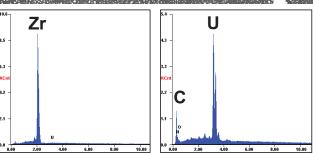




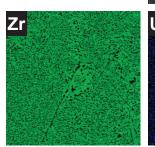


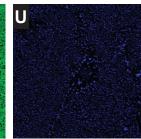


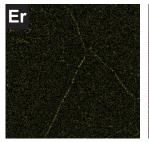




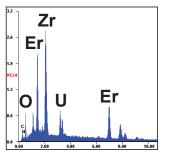
- TRIGA fuels are composed of U-particles dispersed in a Zr-H matrix.
- Analysis was carried out by EDS and X-ray mapping to validate the compositions of the fuel and the matrix.
 - The matrix was rich in Zr and did not contain U or other impurities.
 - Note: EDS cannot detect hydrogen.
 - The U-particles contained measurable concentrations of carbon.
 - Zr-rich grains were observed within the fuel matrix.
 - Er was observed within the grain boundaries.





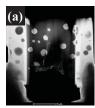


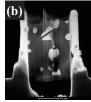






Fuel Characterization: TEM

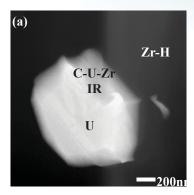


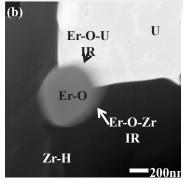




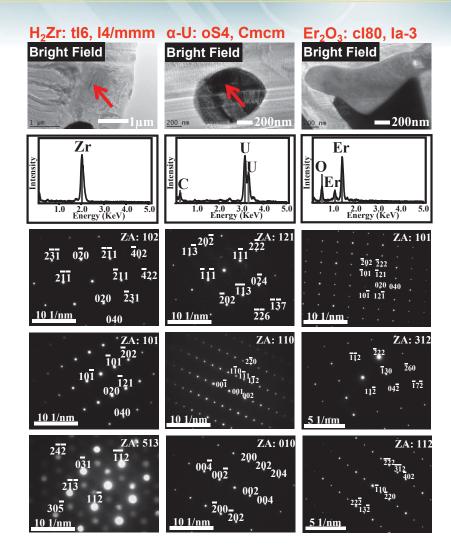






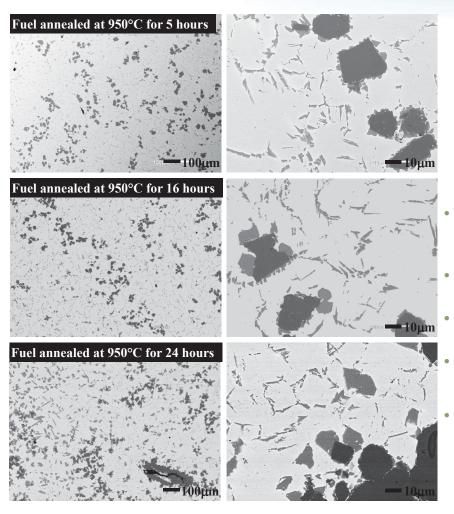


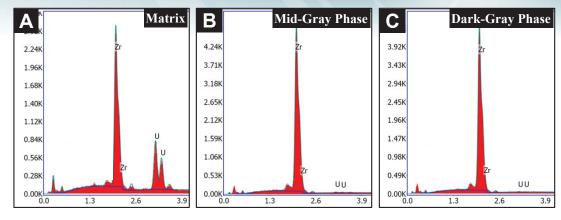
- Interaction regions (IR) were apparent between the Zr-H, U, and Er.
- The matrix was found to be composed of the tetragonal-H₂Zr phase.
 - The cubic-H_{1.6}Zr phase did not present the best fit.
- The U-particles were composed of the expected orthorhombic-α-U phase.
- The Erbium additions were found to be composed of the cubic-Er₂O₃ phase.



Annealed TRIGA Fuel Microstructure



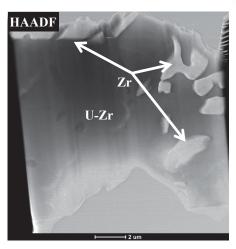


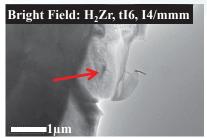


- After annealing, the U-particles fully interacted with the matrix.
 - U-particles were not observed in the annealed fuel.
- The Zr-rich phase regions remained within the matrix.
- Another Zr-rich phase appear to have developed.
- A phase with acicular microstructure developed throughout the matrix.
- EDS indicated that:
 - The matrix contained Zr and U.
 - Phase contrast in the BE micrographs suggest that the two Zr-rich phases contain differing compositions.
 - These phases may contain differing hydrogen concentrations.

Annealed Fuel Characterization: TEM





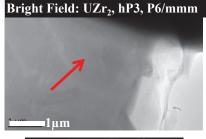


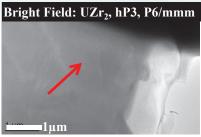
Zone 1

10 1/nm.

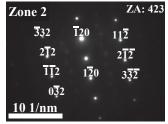
Zone 2

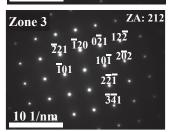
ZA: 001



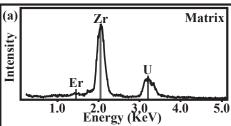


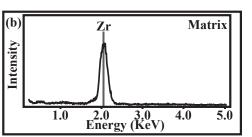


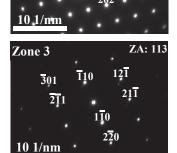




- One sample was prepared for characterization from the fuel annealed at 950°C for 24 hours.
- The sample captured the matrix and acicular phases.
- The dark-gray contrast phases were not captured.
- EDS verified that the matrix was composed of U-Zr, and contained localized small concentrations of Er.
- The acicular phase was Zr-rich and did not contain Er or U.
- The matrix consisted of the hexagonal-UZr₂ phase.
- The Zr-rich particles likely contained the tetragonal H₂Zr phase.
 - The cubic- $H_{1.6}$ Zr phase did not present the best fit.





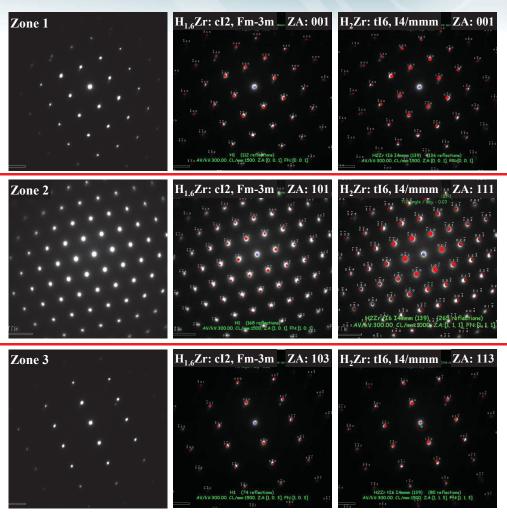


Annealed Fuel Characterization by TEM



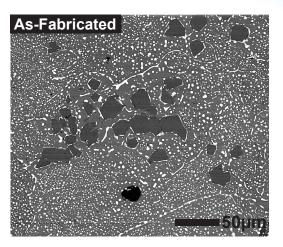
- Diffraction analysis was facilitated by the JEMS electron microscopy software package version 4.3905U2016.
- Characterization analysis of the Zrhydride phases was complicated by close similarities in the spot patterns of the cubic-H_{1.66}Zr and tetragonal-H₂Zr phases.
- The spot patterns did not reliably differentiate between the two phases.

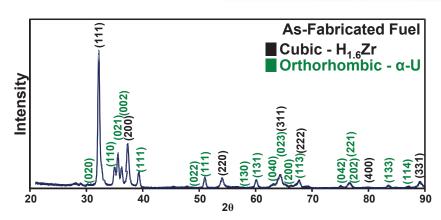




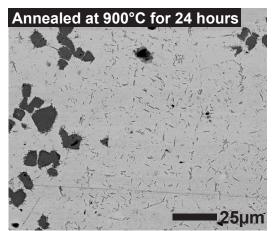
As-Received vs. Annealed Fuel Microstructures

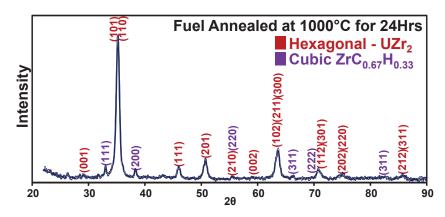






- Significant microstructural changes were observed between the as-fabricated and the annealed microstructures of the TRIGA fuel.
- XRD shows significant changes in the phases present within the microstructure of the matrix.





As-received:

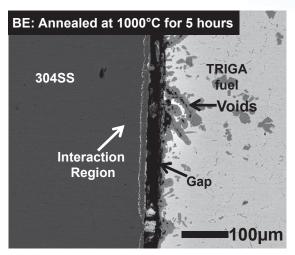
- Cubic H_{1.6}Zr
- Orthorhombic α-U
- Tetragonal H₂Zr (TEM)

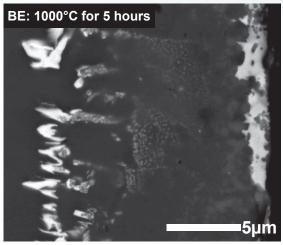
Annealed:

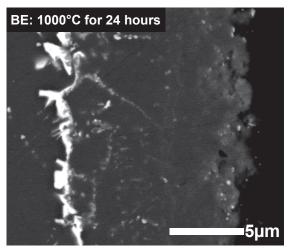
- Hexagonal UZr₂
- Cubic $ZrC_{0.67}H_{0.33}$

Diffusion Couples Annealed at 1000°C for 5 and 24 hours

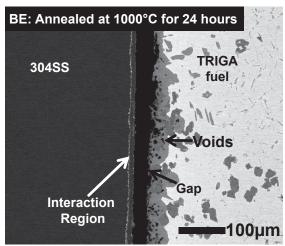








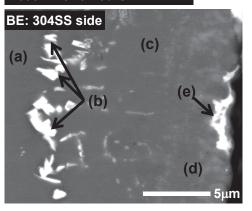
- Diffusion couples of TRIGA fuel vs. 304SS were assembled and annealed at 1000°C for 5 and 24 hours.
- Interaction regions developed within the 304SS side of the diffusion couple and contained multiple phases.
- Typically voids were observed in the TRIGA fuel side of the couple next to the interaction region.
 - Voids were not observed where interaction regions did not develop.
- During annealing, the couples generally experienced failure at or near the interdiffusion interface between the fuel and the 304SS after some time.
 - Volumetric change of the fuel is suspected to cause the interface failure.
- Both diffusion couples developed interaction regions of similar thickness.
 - This result suggests that the couple failure occurred at approximately the same annealing time.

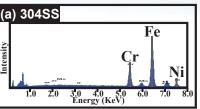


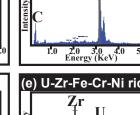
Interaction Regions



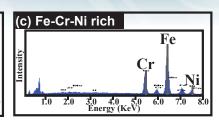
Diffusion Couple Annealed at 1000°C for 5 hours

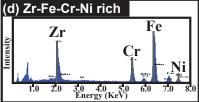


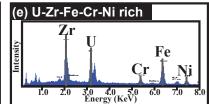




(b) U-Fe-Cr-Ni rich

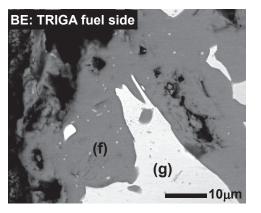


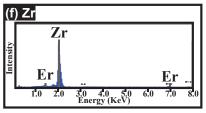


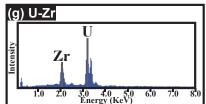


Fe

- An interaction region developed within the 304SS.
 - It contained four distinct phase regions with measurable compositional differences:
 - U-Fe-Cr-Ni
 - Fe-Cr-Ni
 - Zr-Fe-Cr-Ni
 - U-Zr-Fe-Cr-Ni



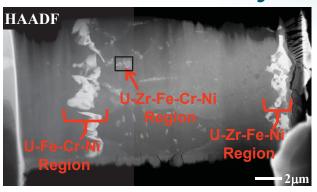


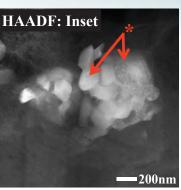


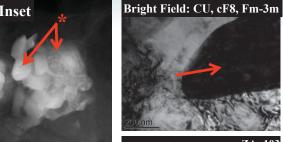
- The TRIGA fuel side of the diffusion couple did not show any elements from the 304SS.
- A Zr-rich layer was observed within the TRIGA fuel near the interdiffusion interface.
- Some of these regions may contain multiple finely distributed phases.
- Due to the interaction volume of the electron beam some of the measured regions contained more than one phase.

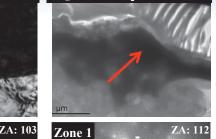










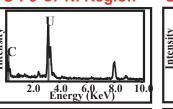


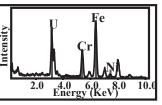
10 1/nm

Bright Field: Fe₂U cF24 Fd-3m

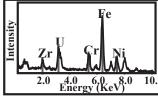


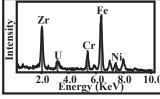


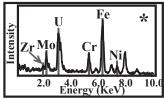




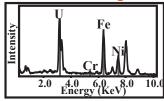


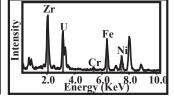




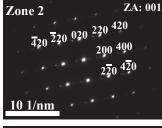




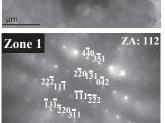


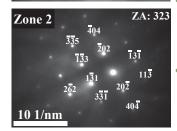


Zone 1 ZA: 103 $\frac{33}{331} \overline{311} \overline{311} \overline{331}$ 020 040 040 020 331 311 311 33110 1/nm



Zone 3 $\overline{13}1$ ZA: 114 $2\overline{2}0 \qquad \overline{3}\overline{1}1$ $5\overline{1}\overline{1} \qquad \overline{5}11$ $31\overline{1} \qquad \overline{2}20$ $13\overline{1}$ $10 \ 1/nm$





Zone 3 ZA: 111 C

\[\frac{7}{2}\text{20} & \text{02\tilde{7}} & \text{22\tilde{4}} \]

\[\frac{7}{2}\text{0} & \text{20\tilde{7}} & \text{40\tilde{4}} \]

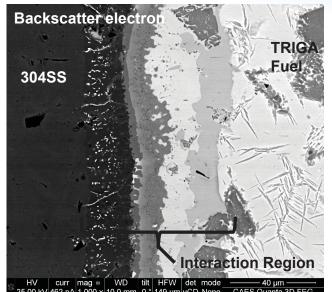
\[\frac{7}{2}\text{4} & \text{0\tilde{7}} & \text{2\tilde{7}} & \text{0} \]

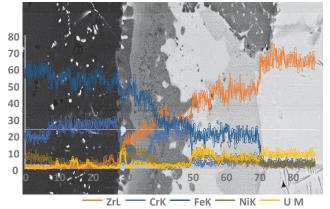
\[\text{040} & \text{10 1/nm} \]

- The interaction region developed grains with significant (U,Zr)-(Fe-Cr-Ni) compositional variations.
- *Mo was observed in some grains within the region containing U-Zr-Fe-Cr-Ni.
- To date, the cubic-Fe₂U, cF24, phase has been identified in the U-Zr-Fe-Ni rich region.
- The cubic-uraniumcarbide (CU), cF8, phase was found in the U-Fe-Cr-Ni region.
- EDS results suggest that several other phases may be present within the interaction region.

Current and Future Work

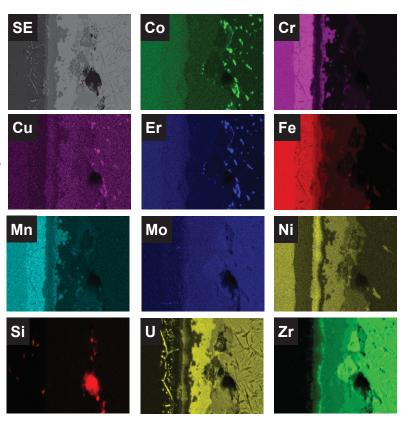






TRIGA Fuel vs. 304SS Diffusion Couple Annealed at 900°C for 24 hours

- X-ray maps have been collected from the interaction at locations of interest.
- SEM analysis showed that the interaction region developed a layered microstructure.
- The maps showed that different layers differed in composition. Layers enrichedin and depleted-in Cr and Ni and vice versa were observed. U showed similar behavior.
- Concentration profiles generated by EDS correlated well with the observations made from the X-ray maps.
- Because the diffusion couple was cooled in the furnace over a period of several hours, cooling effects on microstructural development cannot be ruled out at this time.
- Detailed characterization by TEM is planned.



Summary



- TRIGA fuels in the as-fabricated and annealed conditions were characterized by SEM and TEM to identify the constituents phases in the fuel.
 - The TRIGA fuel matrix is composed of α-U particles dispersed in a $H_{1.6}$ Zr matrix with Er_2O_3 additions concentrated within grain boundaries. Some H_2 Zr was observed by TEM.
 - After annealing at 950°C, the U-particles fully interacted with the matrix to develop a UZr₂ matrix with ZrC_{0.67}H_{0.33} and H₂Zr secondary phases.
- Diffusion couple experiments of 304SS vs. TRIGA were carried out at 1000°C for 5 and 24 hours.
 - The couples developed interaction regions within the 304SS.
 - U-Fe-Cr-Ni, U-Zr-Fe-Cr-Ni, and U-Zr-Fe-Ni regions developed within the interaction region.
 - The UC and Fe₂U phase were identified in the microstructure.
 - Other phases may be present.

