

Effective parameterization of phase-field models of fission gas bubble growth

March 2024

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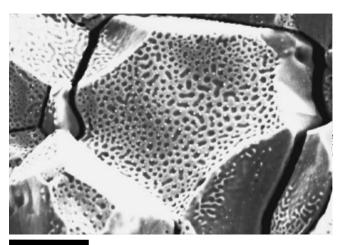
Prepared for the U.S. Department of Energy Under DOE Idaho Operations Office Contract DE-AC07-05ID14517

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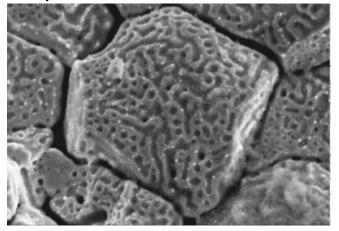
Larry Aagesen, Sourabh Kadambi Idaho National Laboratory



Fission Gas Evolution and Release: Background



10 μm

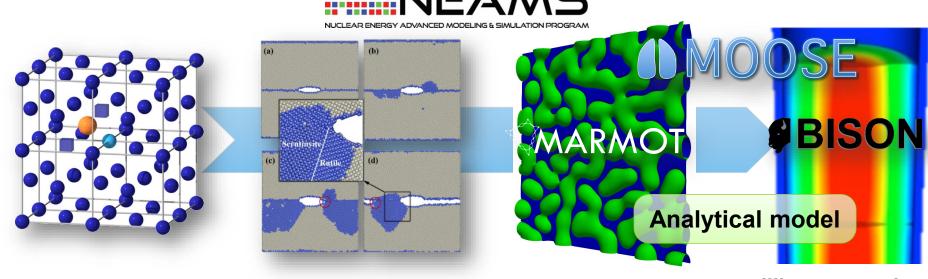


Intergranular Bubbles in UO₂ R.J. White, J. Nuc. Mater., 325, 61-77 (2004)

- Fission products produced in fuel matrix
- Bubbles of low-solubility fission products (Xe, Kr) nucleate
- Intragranular bubbles: 1-5 nm
 - Size limited by re-solution
- Intergranular bubbles: >50 nm
 - Bubbles grow and percolate on grain faces
 - When percolated faces connect to free surface, fission gas is released
 - Microstructure has a strong influence on fission gas release
- Consequences of fission gas release:
 - Reduced thermal conductivity of fuel-cladding gap
 - Increase in plenum pressure: mechanical properties of cladding

Phase-field modeling of fission gas bubble microstructural evolution

- Fundamental scientific understanding
- Inform engineering-scale nuclear fuel performance codes



nanometers First Principles

- Identify critical bulk mechanisms
- Determine bulk properties

100's of nanometers Molecular Dynamics

- Identify interfacial mechanisms
- Determine interfacial properties

microns Mesoscale

- Predict microstructure evolution
- Determine impact on properties

millimeters and up Engineering Scale

- Use analytical theory
- Predict fuel performance

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Phase-field modeling evolution equations: defect species

- Most complete picture:
 - Vacancies

•
$$\frac{\partial c_v}{\partial t} = \nabla \cdot M_v \nabla \mu_v + S_v - K_{iv} c_i c_v$$
Diffusional transport Source Recombination Sink

Interstitials

•
$$\frac{\partial c_i}{\partial t} = \nabla \cdot M_i \nabla \mu_i + S_i - K_{iv} c_i c_v - K_{is} c_i$$

Gases

•
$$\frac{\partial c_g}{\partial t} = \nabla \cdot M_g \nabla \mu_g + S_g$$

 Interstitials can significantly increase computational time due to much more rapid diffusion

Phase-field modeling evolution equations: effective vacancy production

- Due to preferential absorption of interstitials at dislocations and faster diffusion, normally interstitial concentration is much lower than vacancy concentration and there is a net excess of vacancies
- Several past models have included vacancies with a net vacancy production rate
- Source-Only (SO)

•
$$\frac{\partial c_v}{\partial t} = \nabla \cdot M_v \nabla \mu_v + S_v$$

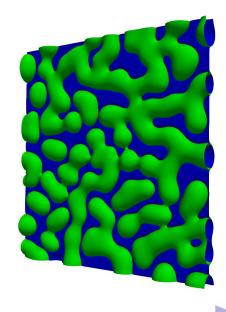
- S_{v} : an effective vacancy source, but what is the right value?
- Source + Sink (SS)

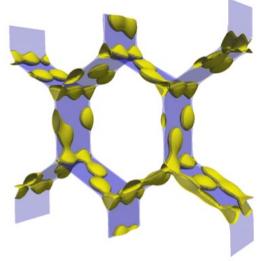
•
$$\frac{\partial c_v}{\partial t} = \nabla \cdot M_v \nabla \mu_v + S_v - K_v c_v$$

- Can use a physical value of $S_{m v}$ and $K_{m v}$ is an effective vacancy sink
- Maintain steady-state vacancy concentration $c_v^{\rm SS} = S_v/K_v$ in bulk far from bubbles

Overview

- Goal: compare SO and SS phasefield models to analytical solutions
 - Simplified geometry
 - Determine advantages/disadvantages
- Determine parameters for phasefield modeling by comparing to analytical model that includes vacancies and interstitials in a simplified geometry
 - "Chemical stress" model





Analytical solution for Source-only (SO)

approach

For simplicity consider vacancies only in spherical domain (void)

Spherical coordinates:

•
$$\frac{\partial c_v}{\partial t} = \frac{D_v}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial c_v}{\partial r} \right) + S_v$$

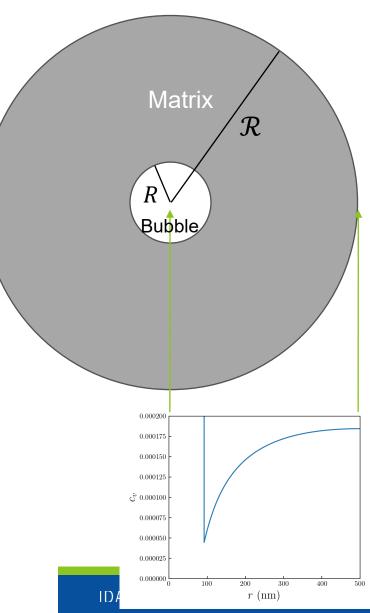
Known analytical solution for quasi-steady state case:

•
$$\frac{\partial c_v}{\partial t} \approx 0 \approx \frac{D_v}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial c_v}{\partial r} \right) + S_v$$

- Boundary conditions: Gibbs-Thomson condition at bubble-matrix interface, 0 gradient at domain boundary
- Analytical solution

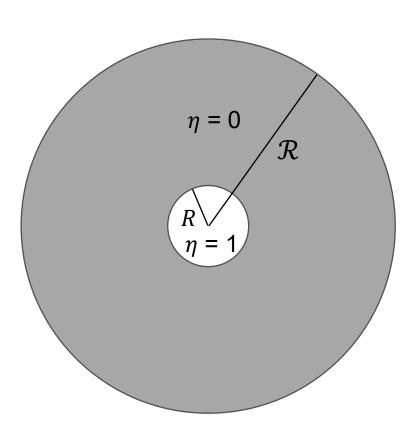
•
$$c_v(r) \approx c_{vR} + \frac{S_v \mathcal{R}^3}{3D_v R} \left(1 - \frac{R}{r}\right)$$

Take derivative to find flux, growth rate:



Phase-field model: Essential features

- Single order parameter η to represent void (bubble) and fuel matrix phase
- Track vacancies
 - SO and SS approaches
- Chemical energy contribution
- Solid-bubble interfacial energy
 - Kim-Kim-Suzuki (KKS) approach to remove bulk energy contribution to interfacial energy



Phase-field model: Free energy functional

•
$$F = \int_{V} \left[f_{chem} + Wg(\eta) + \frac{\kappa}{2} |\nabla \eta|^{2} \right] dV$$

• f_{chem} = bulk chemical free energy density.

•
$$f_{chem} = [1 - h(\eta)] f_{chem}^m(c_v^m) + h(\eta) f_{chem}^b(c_v^b)$$

- $h(\eta)$ is a smooth interpolation function.
- Chemical free energy of the matrix, bubble phases: Parabolic approximations

•
$$f_{chem}^{m} = \frac{k_v}{2} (c_v^m - c_v^{m,min})^2$$

•
$$f_{chem}^b = \frac{k_v}{2} (c_v^b - c_v^{b,min})^2$$

Evolution equations

Allen-Cahn for order parameter:

•
$$\frac{\partial \eta}{\partial t} = L \left[\frac{dh}{d\eta} \left[(f_{chem}^m - f_{chem}^b) - \mu_v (c_v^m - c_v^b) \right] - W \frac{dg}{d\eta} + \kappa \nabla^2 \eta \right]$$

Cahn-Hilliard for vacancies, SO approach:

•
$$\frac{\partial c_v}{\partial t} = \nabla \cdot M_v \nabla \mu_v + S_v$$

Cahn-Hilliard for vacancies, SS approach:

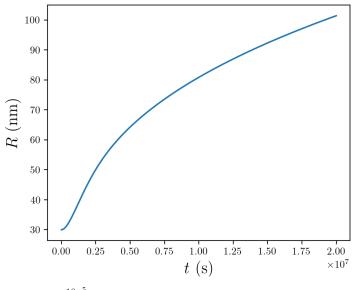
•
$$\frac{\partial c_v}{\partial t} = \nabla \cdot M_v \nabla \mu_v + S_v - K_v c_v$$

Mobilities are a function of defect diffusivities:

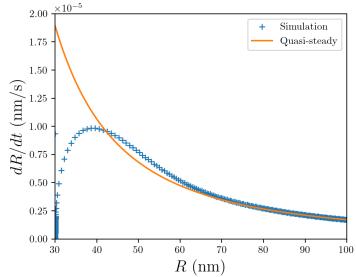
$$M_{v} = \frac{D_{g}(\phi)}{f_{c_{v}c_{v}}} = \frac{hD_{v}^{b} + (1-h)D_{v}^{m}}{f_{c_{v}c_{v}}}$$

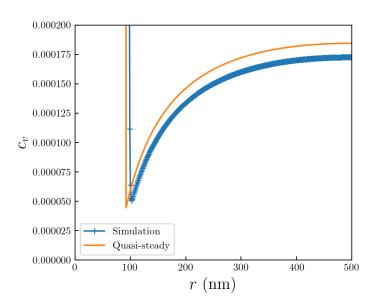
+ KKS system constraints

Source-only model simulations and comparison to quasi-steady state analytical solution



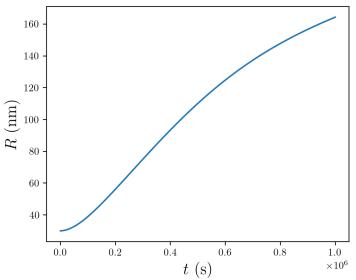
- Vacancy diffusivity based on UO₂ at 700 K
- $S_v = 4.09 \times 10^{-10}$ s⁻¹ (5x gas atom production rate)
- Good match between analytical solution and phase-field simulation after c_v reaches quasi-steady state



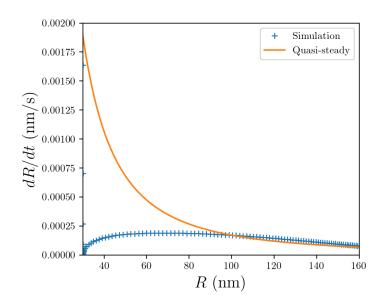


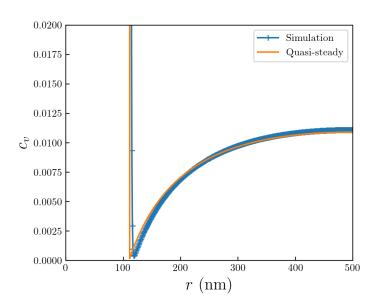
Source-only model simulations: effect of increasing

Sn



- Increase $S_v = 4.09 \times 10^{-8} \text{ s}^{-1}$ (500x gas atom production rate)
- Growth rate increases significantly
- Agreement with analytical solution is only reasonable after bubble grows significantly (dR/dt becomes much smaller)





Analytical solution for Source+Sink (SS)

approach

Spherical coordinates, quasi-steady state:

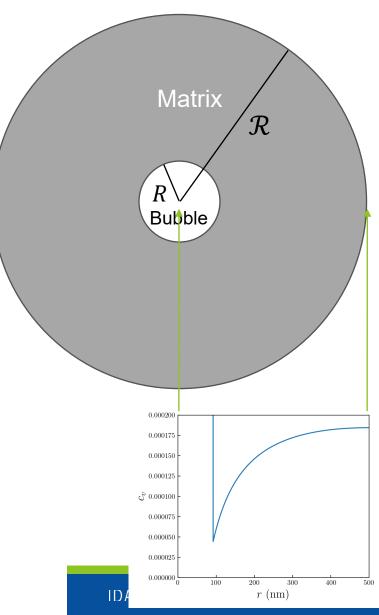
•
$$\frac{\partial c_v}{\partial t} \approx 0 \approx \frac{D_v}{r^2} \frac{\partial}{\partial r} \left(r^2 \frac{\partial c_v}{\partial r} \right) + S_v - K_v c_v$$

- Analytical solution not previously studied. Solve by splitting into homogeneous and particular solutions.
 - Homogeneous part: Modified spherical Bessel equation equation with n=0
 - Solutions: modified spherical Bessel functions of the first and second kinds
 - Boundary conditions: Gibbs-Thomson condition at bubble-matrix interface, 0 gradient at domain boundary
- Analytical solution

•
$$c_v = \left(c_{vR} - \frac{S_v}{K_v}\right) \frac{\lambda R}{e^{-\lambda R}} \frac{e^{-\lambda r}}{\lambda r} + \frac{S_v}{K_v}$$
, where $\lambda = \sqrt{S_v/K_v}$

Take derivative to find flux, growth rate:

•
$$\frac{dR}{dt} = D_v(c_v^{SS} - c_{vR}) \left(\sqrt{\frac{K_v}{D_v}} + \frac{1}{R} \right)$$



"Chemical Stress" model accounts for bubble growth due to both vacancies and interstitials

- "Produces bubble growth like a mechanical force"
- Growth due to vacancies and interstitials:

•
$$\frac{dR}{dt} = \frac{1}{R} [D_v (c_v^{SS} - c_{vR}) + D_i (c_i^{SS} - c_{iR})]$$

Far from bubbles,

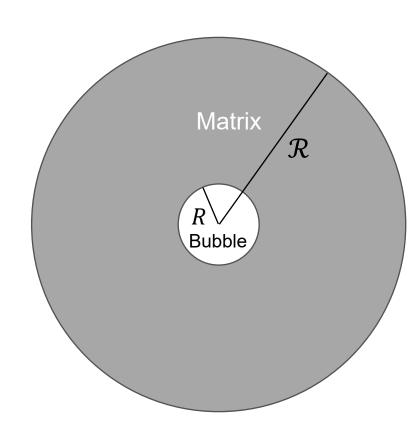
•
$$Z_v D_v (c_v^{SS} - c_{vR}) = Z_i D_i (c_i^{SS} - c_{iR})$$

· Combining results in

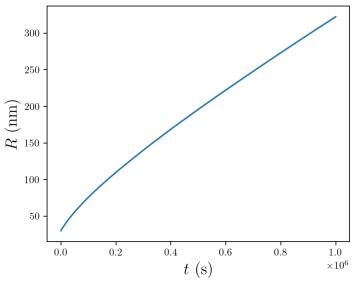
•
$$\frac{dR}{dt} = \frac{D_v}{R} (c_v^{SS} - c_{vR}) \left(1 - \frac{Z_v}{Z_i} \right)$$

Compare to SS analytical solution

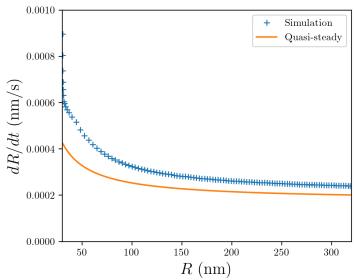
•
$$\frac{dR}{dt} = D_v(c_v^{SS} - c_{vR}) \left(\sqrt{\frac{K_v}{D_v}} + \frac{1}{R} \right)$$

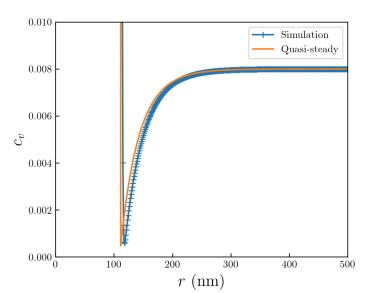


Source+sink model simulations and comparison to quasi-steady state analytical solution



- $S_v = 4.09 \times 10^{-6} \text{ s}^{-1}$ (10⁴ x fission rate, physically reasonable)
- Expect $c_v^{SS} \approx 8 \times 10^{-3}$ based on cluster dynamics calculations
 - Set $K_v = S_v/c_v^{SS} = 1.02 \times 10^{-3}$
- Reasonably agreement between phasefield model and analytical solution





Growth rate in source+sink model has undesirable variation with parameter choices

From intuition and chemical stress model

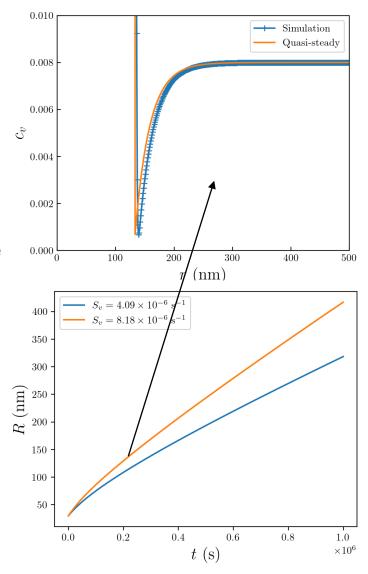
•
$$\frac{dR}{dt} = \frac{D_v}{R} (c_v^{SS} - c_{vR}) \left(1 - \frac{Z_v}{Z_i} \right)$$

- Expect that in the SS model, as long as c_v^{SS} remains constant, the growth rate should remain the same.
- Test by doubling S_v and K_v
 - · Growth rate increases significantly!
- Insight provided by analytical solution to SS growth rate

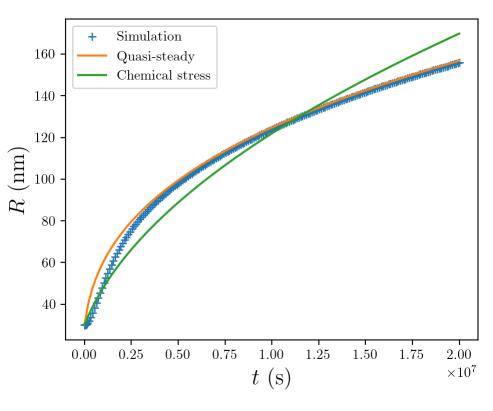
•
$$\frac{dR}{dt} = D_v(c_v^{SS} - c_{vR}) \left(\sqrt{\frac{K_v}{D_v}} + \frac{1}{R} \right)$$

• More difficult to parameterize SS model to match vacancy+inters tiak model compared to SO, where

$$\frac{dR}{dt} = \frac{S_v \mathcal{R}^3}{3R^2}$$



Parameterizing Source-Only model to match vacancy+interstitial model



Source-only analytical solution:

•
$$R = [S_v \mathcal{R}^3 t + R_0^3]^{1/3}$$

• Chemical stress model:

•
$$R = \left[2D_v \left(1 - \frac{z_v}{z_i}\right)c_v^{SS}t + R_0^2\right]^{1/2}$$

- Fit to determine S_v that gives best match between them
- Phase-field simulation parameterized with this value matches vacancy+interstitial model reasonably well
 - $S_v = 1.53 \times 10^{-9} \text{ s}^{-1}$, approximately 16x greater than Xe production rate

Conclusions

- Compared source-only and source+sink phase-field models to analytical solution
 - Developed new analytical solution for the source+sink case
- Analytical models predict phase-field model behavior reasonably well when bubble growth rate is small
 - Provide qualitative and quantitative insight into model behavior
- Source+sink model has significant disadvantages in parameterization
- Developed strategy to determine effective vacancy source term for source-only model
 - For conditions considered, vacancy source term approx. 16x larger than gas source term. Past work has used values in the range 1-20x gas source term