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Changing the World's Energy Future

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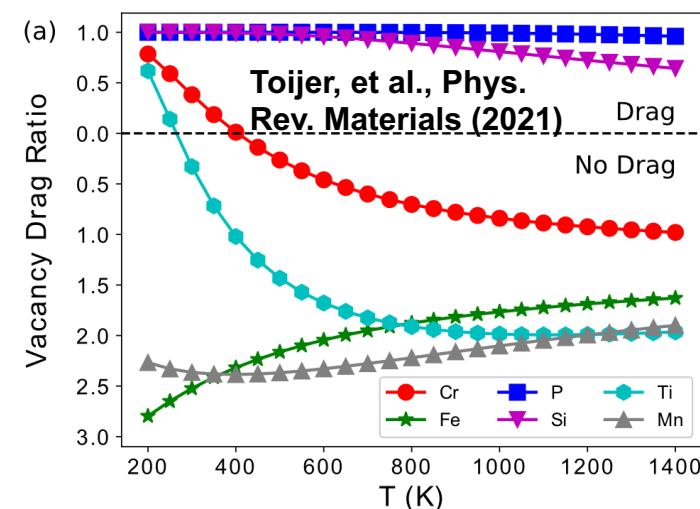
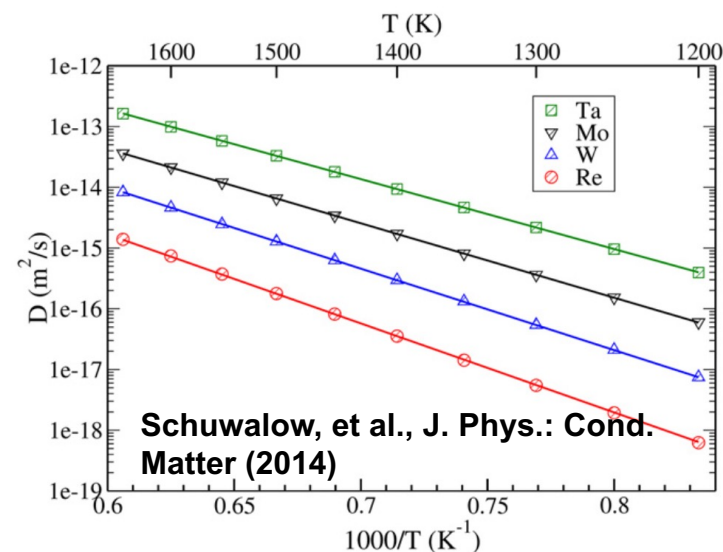


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Literature

Several DFT studies on solute diffusion in FCC Ni

- DFT + five-frequency model:
 - interactions limited to the 1st nn.
 - Tucker, et al., J. Nucl. Mater. (2010): Cr and Fe solutes
 - Hargather et al., Acta Materialia (2018): diffusivities of 26 solutes in FCC Ni
- DFT + KMC :
 - Schuwalow et al. calculated diffusivities of Mo, Re, W, and Ta solutes in FCC Ni
- DFT + SCMF :
 - taking into account kinetic correlations to calculate Onsager transport coefficients.
 - Toijer et al. studied the solute-point defects interaction in FCC Ni using DFT (solutes: Cr, Fe, Si, Al, P, and Mn): vacancy drag and segregation tendencies



Methodology

- In this work:
 - we adopt the methodology of DFT+SCMF to investigate the segregation tendencies of Co, Cr, Mo, Re, W, and Ta solutes in Ni-based superalloys
 - we extend this model to include the segregation tendencies under strain using the linear elasticity theory as implemented in the KineCluE code
- Using density functional theory (DFT):
 - ➔ vacancy-solute interactions: formation, binding and migration energies
 - ➔ phonon calculations to obtain attempt frequencies (diffusion prefactors) and vacancy formation entropy
- Using self-consistent mean field theory (SCMF):
 - ➔ cluster transport coefficients and partition functions

DFT Computational Details

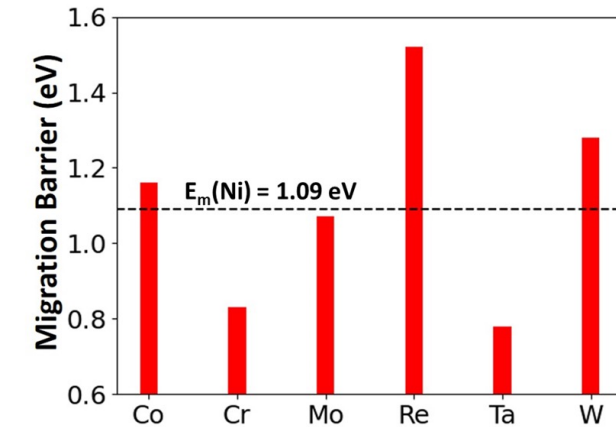
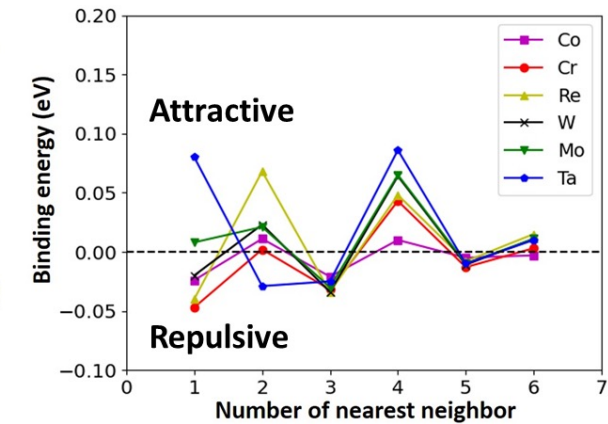
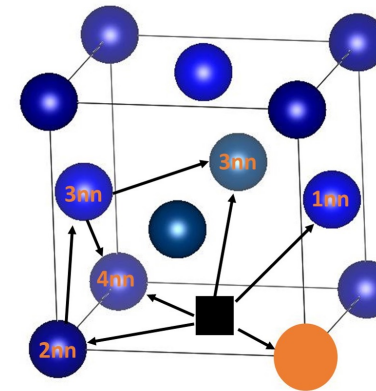
- PAW pseudopotentials as implemented in VASP
 - GGA-PBE exchange-correlation functionals
 - 108-atom and 192-atom supercells
- Climbing-image NEB to get the saddle point energy for atomic jumps
 - One intermediate image
- Frozen phonons method to calculate the attempt frequencies
 - 0.02 Å atomic displacements to get force constants

Vacancy-Solute Interactions

- Vacancy-solute interactions up to the 4th nn are considered
- For each solute, the migration barrier for 8 jumps are calculated:
 - vacancy-solute exchange
 - 7 vacancy jumps (dissociation/association and reorientation)
(1nn→1nn, 1nn→2nn, 1nn→3nn, 1nn→4nn, 2nn→3nn, 3nn→3nn, 3nn→4nn)

Findings:

- Ta, Cr, and Mo are relatively fast diffusers
- Co, W, and Re are relatively slow diffusers.



SCMF Computational Details

$$\begin{pmatrix} J_V \\ J_B \end{pmatrix} = \frac{-1}{k_B T} \overbrace{\begin{pmatrix} L_{VV} & L_{VB} \\ L_{VB} & L_{BB} \end{pmatrix}}^{\text{Onsager matrix}} \begin{pmatrix} \nabla \mu_V \\ \nabla \mu_B \end{pmatrix}$$

- Cluster expansion:

$$\begin{pmatrix} L_{VV} & L_{VB} \\ L_{VB} & L_{BB} \end{pmatrix} = C \underbrace{\begin{pmatrix} L_{VV}^{(V)} & 0 \\ 0 & 0 \end{pmatrix}}_{\text{Contribution from clusters with vacancies}} + f_{VB} \underbrace{\begin{pmatrix} L_{VV}^{(VB)} & L_{VB}^{(VB)} \\ L_{VB}^{(VB)} & L_{BB}^{(VB)} \end{pmatrix}}_{\text{Contribution from clusters with vacancy-solute pairs}}$$

- f_V and f_{VB} are the fractions of vacancies and vacancy-solute pairs, respectively, and are obtained from the SCMF partition functions
- Kinetic range (r_{kin}) = 4 lattice constants (after performing convergence tests)
- Dilute limit: maximum solute concentration [B] using this cluster size is 0.1 at. %

Cluster fractions

$$f_V = \frac{C_V - C_{\text{corr}}}{C}, f_{VB} = \frac{C_{VB}}{C}$$

$$C_V = Z_V[V] = Z_V e^{\frac{-E_{\text{form}}^{\text{vac}}}{k_B T}} e^{\frac{S_{\text{form}}^{\text{vac}}}{k_B T}}$$

$$C_{VB} = Z_{VB}[V][B]$$

Z_{VB} : pair partition function

$$Z_{VB} = \sum_i e^{\frac{E_{\text{bind}}^i}{k_B T}}$$

$$C_{\text{corr}} = Z_{VB}^0[V][B]$$

(sites that single vacancies cannot occupy because of the geometrical definition of the vacancy-solute pair)

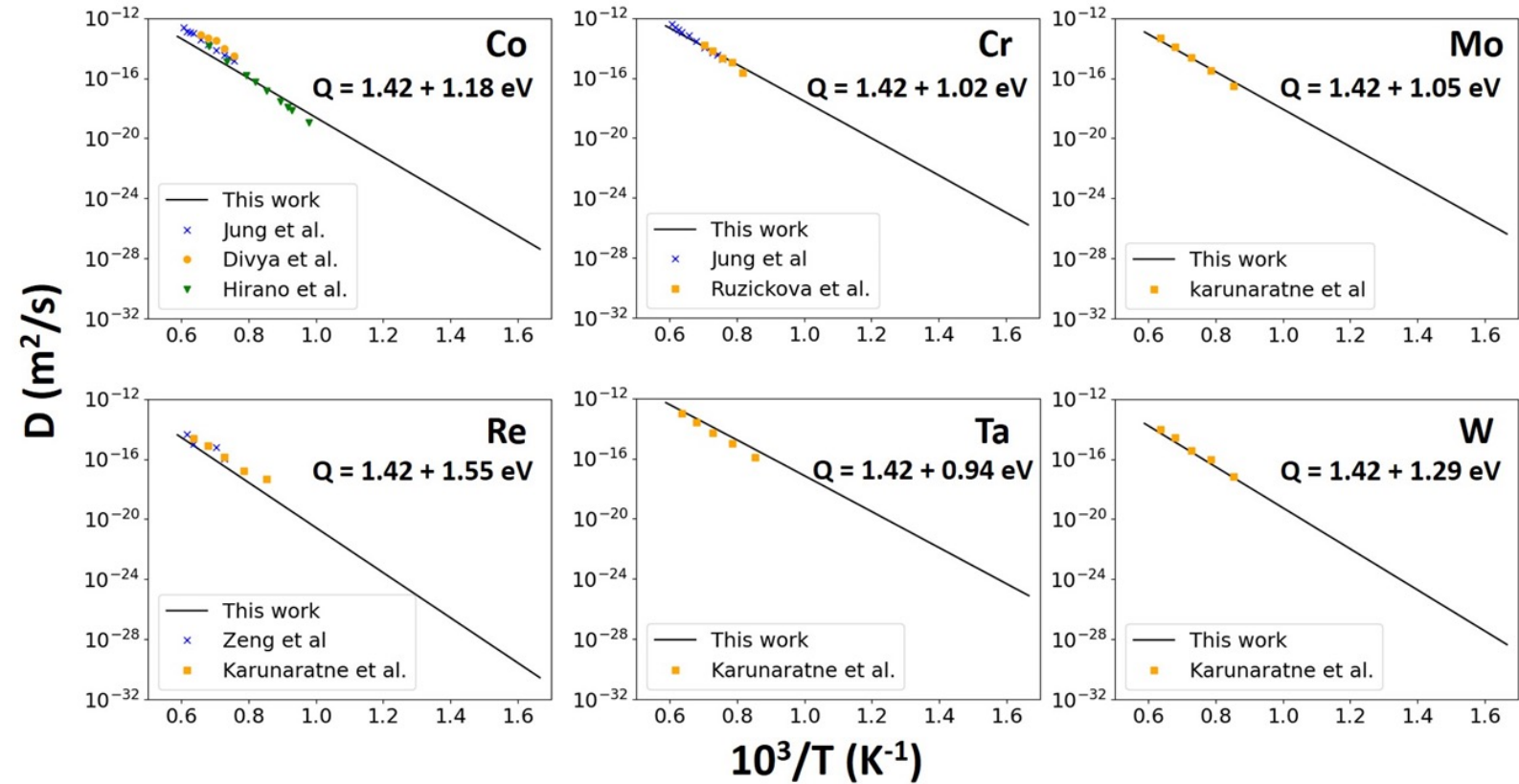
Z_{VB}^0 : non-interacting cluster partition function (# of possible pair configurations within r_{kin}).

Solute Diffusivities

- In the dilute limit:

$$D_B = \frac{L_{BB}}{[B]}$$

- Good agreement of calculations with experimental data



Solute vacancy drag and partial diffusion coefficient ratio

We analyze flux coupling from transport coefficients

→ Vacancy drag ratio (G) indicates if a solute

- follows the vacancy in its migration ($G > 0$) or
- diffuses in the opposite direction ($G < 0$)

$$G = \frac{L_{VB}^{(VB)}}{L_{BB}^{(VB)}}$$

→ PDC ratio describes the diffusion of solute atoms relative to matrix atoms

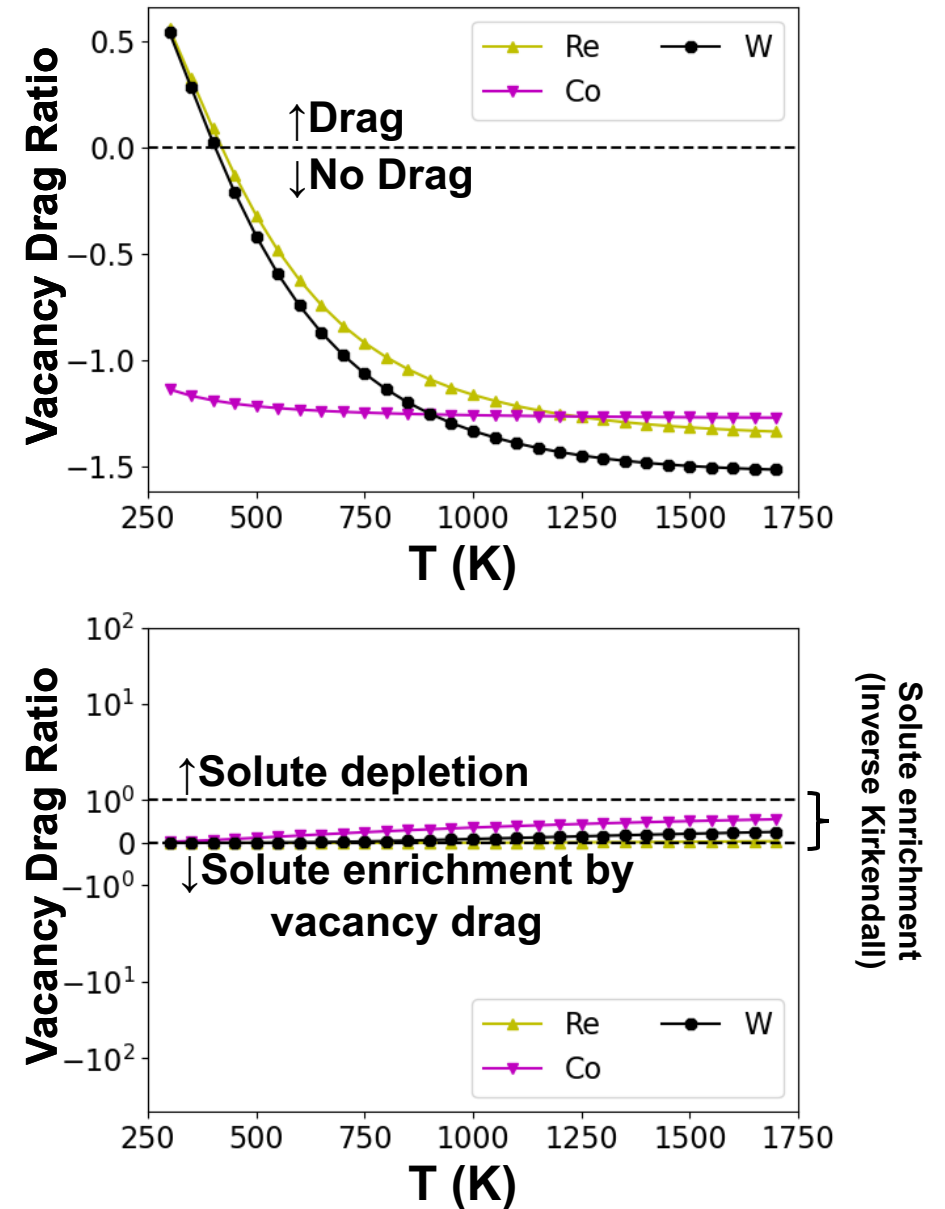
- **-ve PDC ratio: vacancy drag** → solute enrichment at sinks
- **+ve PDC ratio: inverse Kirkendall**
 - PDC ratio > 1 : solute depletion at sinks
(favorable solute-vacancy exchange)
 - 0 < PDC ratio < 1 : solute enrichment
(slower than the host atoms, so effectively enriched)

$$PDC = \frac{1 - C_B}{C_B} \frac{L_{VB}}{L_{AV}}$$

Segregation Tendencies

(A) Slow diffusers

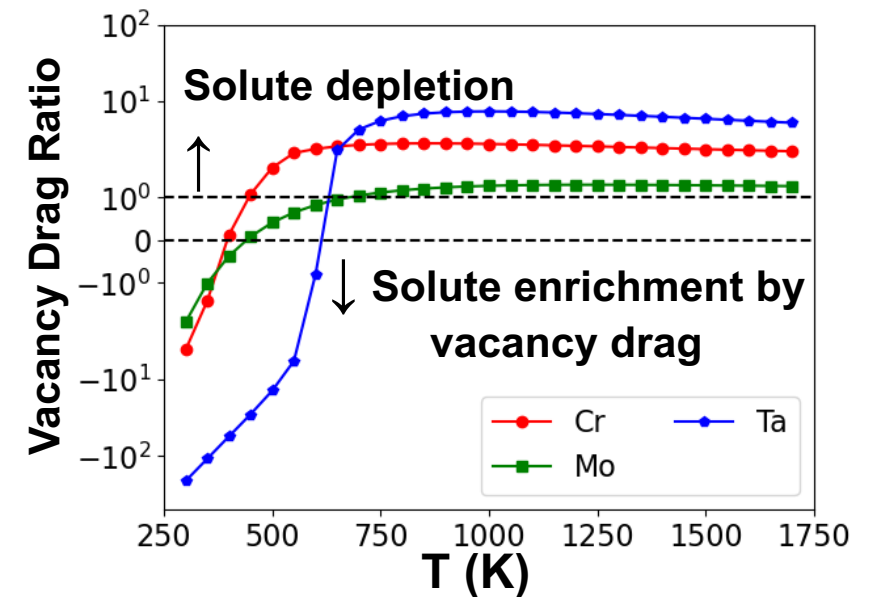
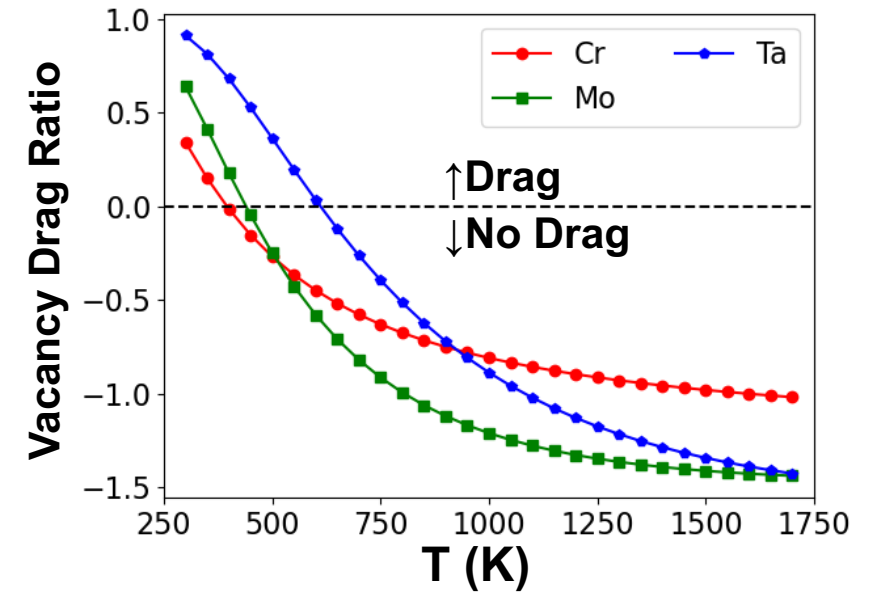
- Re and W :
 - dragged by vacancies at low temperature below ~400 K
 - At higher temperatures, they move opposite to the vacancy flux but are enriched (inverse Kirkendall mechanism)
- Co:
 - not dragged by vacancies at any temperature due to weak binding
 - Enrichment by inverse Kirkendall mechanism



Segregation Tendencies

(B) Fast diffusers

- Cr and Mo:
 - dragged by vacancies at low temperature below ~400 K (enriched)
 - Cr depletion at $T > 500$ K
 - Mo depletion at $T > 600$ K
- Ta:
 - Dragged by vacancies up to 600 K due to strong binding
 - Ta depletion above 650 K



Strain Effects

Linear elasticity theory approach

- Instead of repeating DFT calculations on strained supercells:
- The binding and migration energies under strain tensor ε_{ij} :

$$E_b^x(\varepsilon = \varepsilon_{ij}) = E_b^x(\varepsilon = 0) + P_{ij}^x \varepsilon_{ij}$$
$$E_m^{x1 \Rightarrow x2}(\varepsilon = \varepsilon_{ij}) = E_m^{x1 \Rightarrow x2}(\varepsilon = 0) - P_{ij}^{ts(x1 \Rightarrow x2)} \varepsilon_{ij}$$

- P_{ij}^x is the DFT elastic dipole tensor of a defect configuration (x)
- $P_{ij}^{ts(x1 \Rightarrow x2)}$ is the DFT elastic dipole tensor of a transition state between two configurations x1 and x2

- In this work, we are only considering a tensile hydrostatic strain $\begin{pmatrix} +e & 0 & 0 \\ 0 & +e & 0 \\ 0 & 0 & +e \end{pmatrix}$

Segregation Tendencies Under Strain

- Two factors control the sensitivity of strain effect on the segregation tendency
 1. **Relative mobility of the solute**
 - The segregation tendency of slow diffusers (Co, W, Re) is not affected by strain (even if the vacancy drag ratio changes)
 - The segregation tendency of fast diffusers (Mo, Cr, Ta) is significantly affected by strain
 2. **The stress introduced by the solute in the lattice (relaxation volume)**
 - The vacancy drag of solutes with higher residual stresses have higher sensitivity to strain

Segregation Tendencies Under Strain:

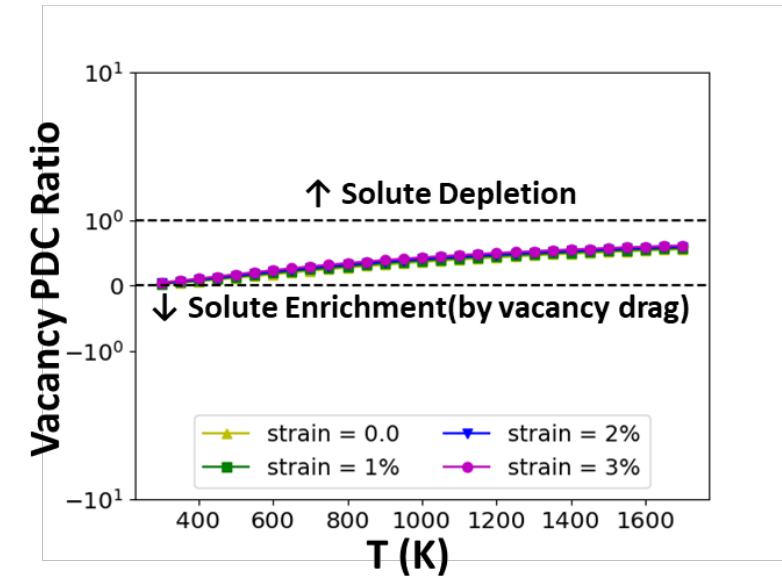
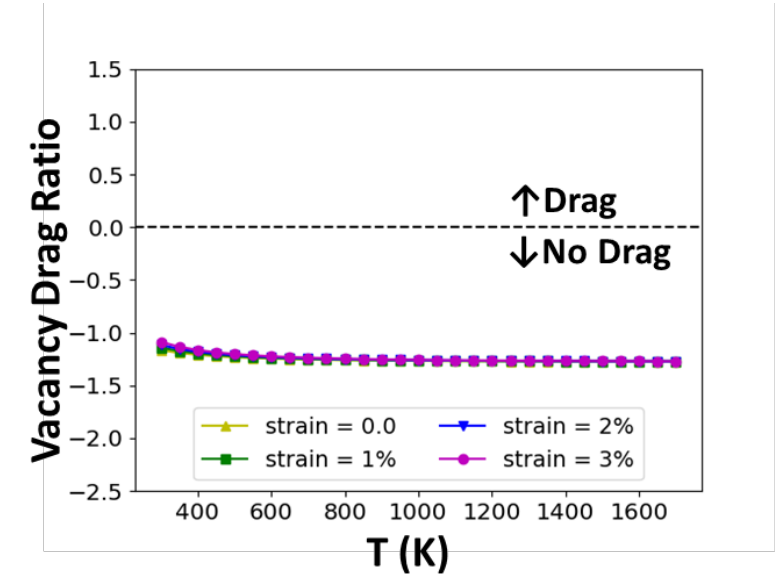
(A) Slow diffusers

(1) Co

(Slow diffuser, zero residual stress)

- No change in vacancy drag
(because of the zero elastic dipole tensor)
- No change in segregation tendency

	E_m (eV)	Stress (eV)	Solute-vacancy pair stress (eV)
Vacancy	1.09	-4.6 /	-
Co	1.16	0.0 / <i>(No residual stress)</i>	$\begin{pmatrix} -4.7 & 0 & 0 \\ 0 & -4.6 & 0.11 \\ 0 & 0.11 & -4.6 \end{pmatrix}$



Segregation Tendencies Under Strain:

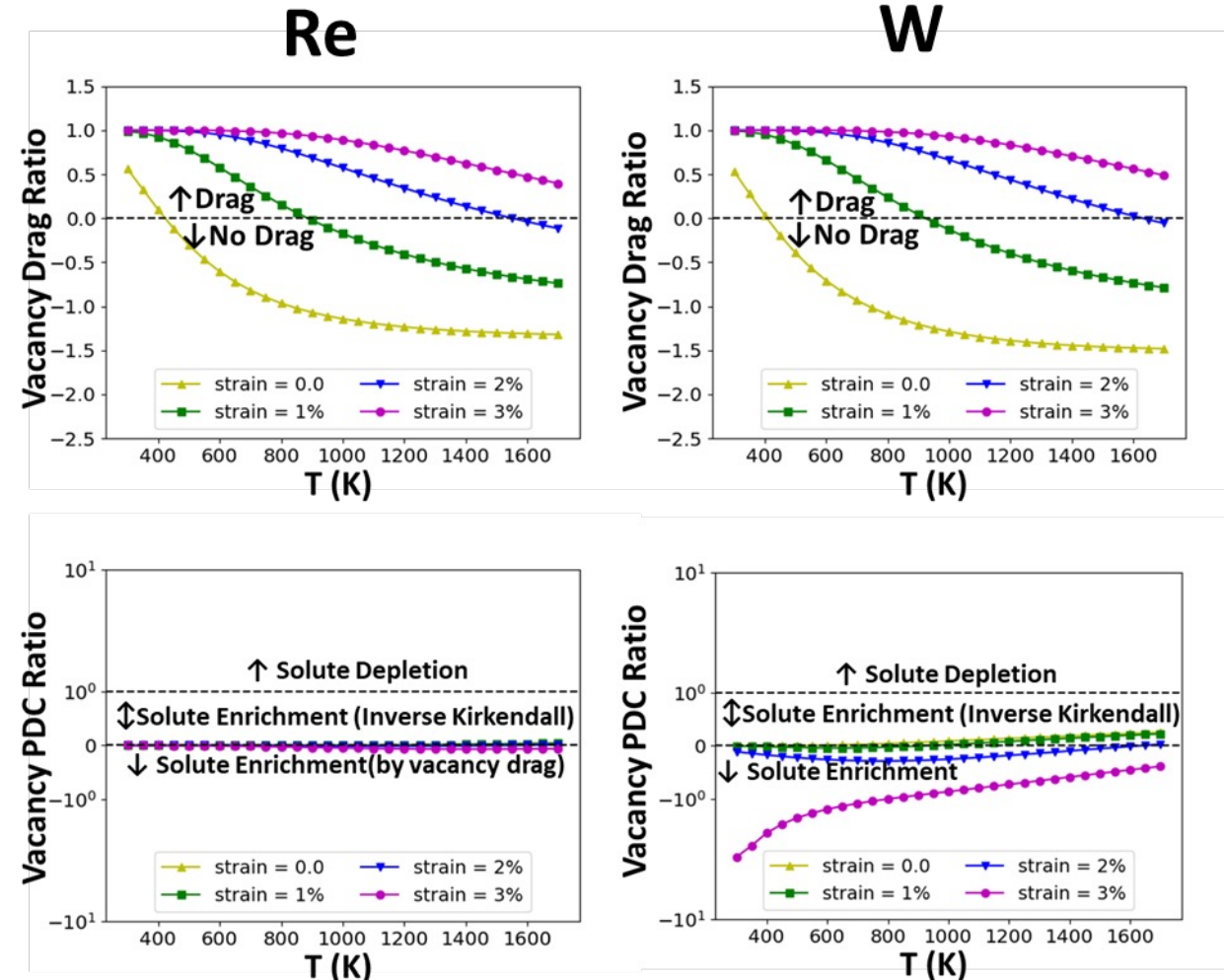
(A) Slow diffusers

(2) Re and W

(Slow diffusers,
non-zero residual stress)

- Although vacancy drag increases with tensile strain, Re enrichment is minimally affected
(too slow to be strongly dragged)
- W enrichment increases by vacancy drag mechanism
(faster than Re)

	E_m (eV)	Stress (eV)
Re	1.52 (slowest)	5.0 /
W	1.28	5.5 /

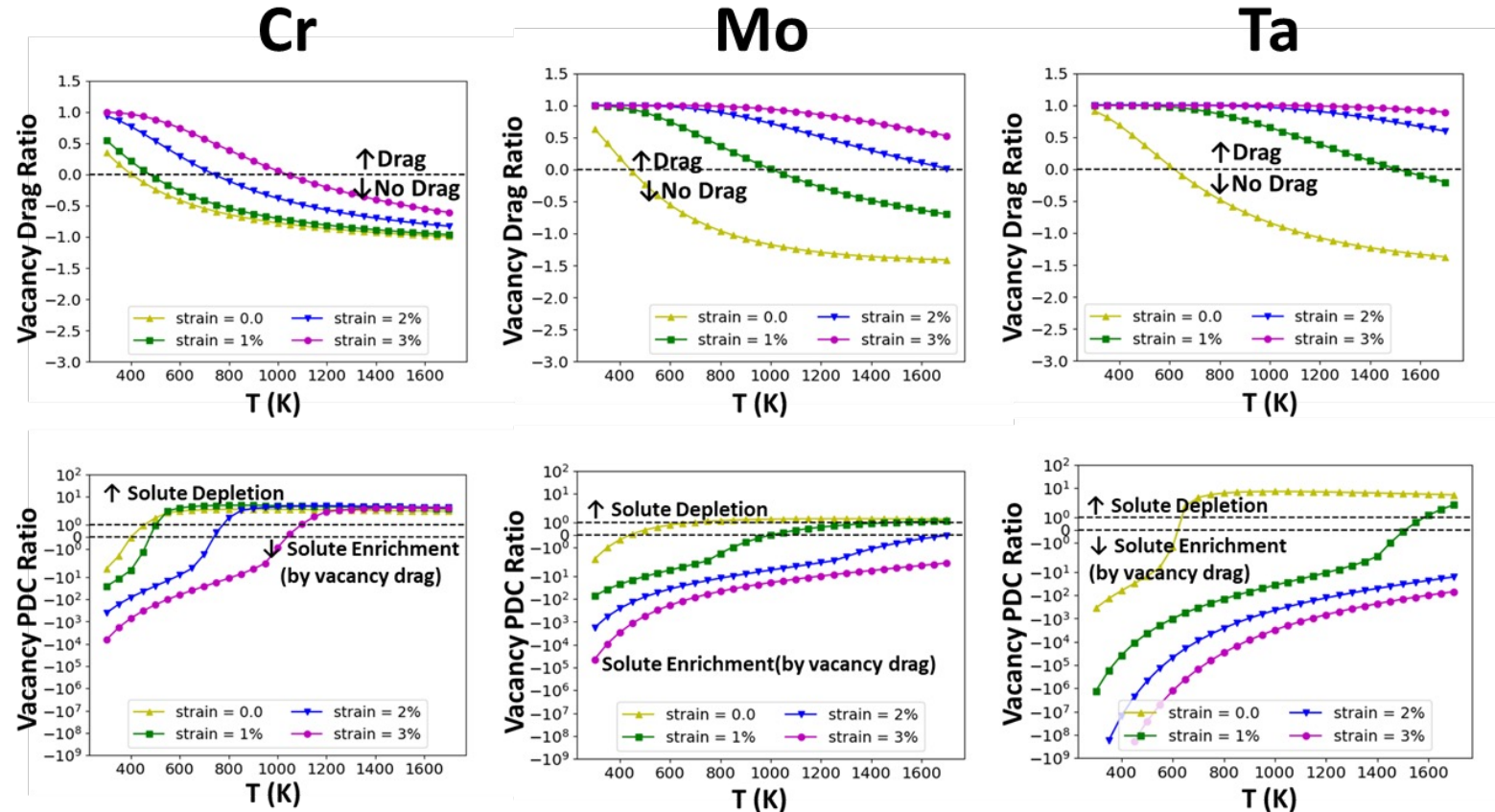


Segregation Tendencies Under Strain: (B) Fast diffusers

(3) Cr, Mo, and Ta (fast diffusers)

- Multiple orders of magnitudes changes in PDC ratio
- The increase of enrichment tendency is the highest for Ta (fastest & highest residual stress)

	E_m (eV)	stress (eV)
Mo	1.07	5.4 /
Cr	0.84	2.7 /
Ta	0.78	7.9 /



Summary and Conclusions

- Solute-vacancy defect energetics in FCC Ni were calculated using first-principles calculations (DFT)
- The Onsager transport coefficients were evaluated using the SCMF approach
- Fast diffusers (Ta, Cr, and Mo)
 - depleted at sinks due to favorable vacancy-solute exchange (inverse Kirkendall) at high temperatures
 - enriched at sinks by vacancy drag at low temperatures
- Slow diffusers (Co, Re, and W)
 - Enriched at sinks by the inverse Kirkendall mechanism
- The segregation tendency of fast-diffusing solutes is more sensitive to strain



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